

FINAL JEE–MAIN EXAMINATION – APRIL, 2023

(Held On Tuesday 11 April, 2023)

TIME : 9 : 00 AM to 12 : 00 NOON

MATHEMATICS

SECTION-A

1. The value of the integral

$$\int_{-\log_e 2}^{\log_e 2} e^x (\log_e (e^x + 1 + e^{2x})) dx$$

- (1) $\log_e \frac{1+5}{2}$
 (2) $\log_e \frac{2(3-5)}{1+5}$
 (3) $\log_e \frac{(2+5)^2}{1+5}$
 (4) $\log_e \frac{2(2+5)}{1+5}$

Official Ans. by NTA (4)

Sol. $I = \int_{-\ln 2}^{\ln 2} e^x (\ln(e^x + 1 + e^{2x})) dx$

Put $e^x = t$ $e^x dx = dt$

$$I = \int_{1/2}^2 \ln(t + t^2) dt$$

Applying integration by parts.

$$= t \ln(t + t^2) - \int_{1/2}^2 (1 + 2t) dt$$

$$= 2 \ln(2 + 5) - \left[\frac{1}{2} t^2 + t^2 \right]_{1/2}^2$$

$$= 2 \ln(2 + 5) - \left[\frac{1}{2} (4) + 4 - \left(\frac{1}{2} \left(\frac{1}{4} \right) + \frac{1}{4} \right) \right]$$

$$= \ln \frac{(2+5)^2}{5+1}$$

TEST PAPER WITH SOLUTION

2. If equation of the plane that contains the point $(-2, 3, 0)$ and is perpendicular to each of the planes

$$2x + 4y + 5z = 8 \text{ and } 3x - 2y + 3z = 5$$

$$ax + by + cz + 97 = 0 \text{ then } a + b + c =$$

- (1) 17
 (2) 16
 (3) 15
 (4) 10

Official Ans. by NTA (3)

Sol. The equation of plane through $(-2, 3, 0)$ is

$$a(x + 2) + b(y - 3) + c(z - 0) = 0$$

it is perpendicular to $2x + 4y + 5z = 8$ & $3x - 2y + 3z = 5$

$$\begin{cases} 2a + 4b + 5c = 0 \\ 3a - 2b + 3c = 0 \end{cases}$$

$$3a - 2b + 3c = 0$$

$$\frac{a}{-2} = \frac{-b}{3} = \frac{c}{-2}$$

$$\frac{a}{22} = \frac{b}{9} = \frac{c}{-16}$$

Equation of Plane is

$$22(x + 2) + 9(y - 3) - 16z = 0$$

$$22x + 9y - 16z + 97 = 0$$

$$\text{Comparing with } ax + by + cz + 97 = 0$$

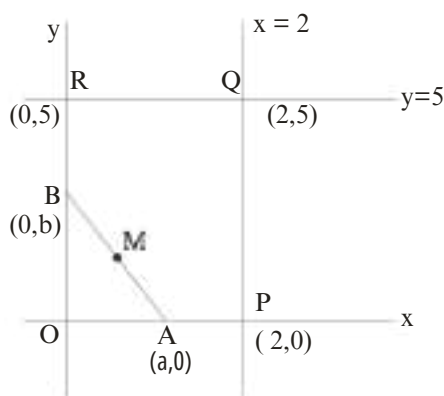
$$\text{We get } a + b + c = 22 + 9 - 16 = 15$$



3. Let R be a rectangle given by the lines $x = 0$, $x = 2$, $y = 0$ and $y = 5$. Let $A(a, 0)$ and $B(0, b)$ be such that the line segment AB divides the area of the rectangle R in the ratio $1:1$. Then, the mid-point of AB lies on a (1) parabola (2) hyperbola (3) straight line (4) circle
Official Ans. by NTA (2)

Sol. $\frac{\text{ar OPQR}}{\text{ar OAB}} = 4$

Let M be the mid-point of AB.



$$M\left(\frac{a}{2}, \frac{b}{2}\right)$$

10 $\frac{1}{2}ab = 4$

5 $ab = 10$ $\frac{1}{2}ab = 4$

2h)2K)=4

Locus of M is $xy = 1$

Which is a hyperbola.

Let sets A and B have n elements each. Let the mean of the elements in sets A and B be \bar{a} and \bar{b} respectively and the variance of the elements in sets A and B be s_1^2 and s_2^2 respectively. A new set C of n elements is formed by subtracting \bar{a} from each element of A and adding \bar{b} to each element of B. Then the sum of the mean and variance of the elements of C is _____.

- (1) 32
(2) 38
(3) 40
(4) 36

Official Ans. by NTA (2)

Sol. $wA = \{a_1, a_2, a_3, a_4, a_5\}$

$B = \{b_1, b_2, b_3, b_4, b_5\}$

Given, $\sum_{i=1}^n a_i = 20$, $\sum_{i=1}^n b_i = 40$

$$\frac{\sum_{i=1}^n a_i}{n} = \frac{20}{5} = 4, \quad \frac{\sum_{i=1}^n b_i}{n} = \frac{40}{5} = 8$$

$\sum_{i=1}^n a_i^2 = 185$, $\sum_{i=1}^n b_i^2 = 420$

Now, $C = \{C_1, C_2, \dots, C_{10}\}$

s.f. $C = a_i - 3$ or $b_i + 2$
First five elements

Mean of C, $\bar{C} = \frac{\sum_{i=1}^{10} (a_i - 3) + \sum_{i=1}^{10} (b_i + 2)}{10}$

$$\bar{C} = \frac{10 + 50}{10} = 6$$

$$s_2^2 = \frac{\sum_{i=1}^{10} (a_i - 3)^2 + \sum_{i=1}^{10} (b_i + 2)^2}{10} - (\bar{C})^2$$

$$= \frac{\sum_{i=1}^{10} a_i^2 - 6 \sum_{i=1}^{10} a_i + 4 \sum_{i=1}^{10} b_i^2 + 65}{10} - 36$$

$$= \frac{185 + 420 - 150 + 65}{10} - 36$$

= 32

Mean + Variance = $\bar{C} + s_2^2 = 6 + 32 = 38$

- Let $f(x) = x^2 - x + \lfloor -x + [x] \rfloor$, where $[x]$ denotes the greatest integer less than or equal to x . Then, f is
- continuous at $x = 0$, but not continuous at $x = 1$
 - continuous at $x = 0$ and $x = 1$
 - not continuous at $x = 0$ and $x = 1$
 - continuous at $x = 1$, but not continuous at $x = 0$
- Official Ans. by NTA (ε)

Sol. Here $f(x) = x^2 - x + \lfloor -x + [x] \rfloor$

$f(0) = -1 + 0 = -1$	$f(1) = 0 + 0 = 0$
$f(0) = 0$	$f(1) = 0$
	$f(1) = -1 + 1 = 0$

- $f(x)$ is continuous at $x = 1$, discontinuous at $x = 0$.
The number of triplets (x, y, z) , where x, y, z are distinct non negative integers satisfying $x + y + z = 10$, is
- 8
 - 11
 - 9
 - 13
- Official Ans. by NTA (γ)

Sol. $x + y + z = 10$
Total no. of solution = $10 + 3 - 1 = 12$... (1)
Let $x = y = z$
 $2x + z = 15$
 $z = 15 - 2x$
 $x \in \{0, 1, 2, \dots, 7\}$
∴ 8 solutions
there are 8 solutions in which exactly
Two of x, y, z are equal ... (2)
There is one solution in which $x = y = z$... (3)
Required answer = $12 - 2 - 1 = 9$

- For any vector $a = a_1\hat{i} + a_2\hat{j} + a_3\hat{k}$ with $|a| < 1, i = 1, 2, 3$, consider the following statements:
- (A) : $\max\{|a_1|, |a_2|, |a_3|\} \leq |a|$
(B) : $|a| \leq 3 \max\{|a_1|, |a_2|, |a_3|\}$
- Only (B) is true
 - Only (A) is true
 - Neither (A) nor (B) is true
 - Both (A) and (B) are true
- Official Ans. by NTA (ε)

Sol. Without loss of generality
Let $a_1 \leq a_2 \leq a_3$

$$|a|^2 = a_1^2 + a_2^2 + a_3^2 \leq 3a_3^2$$

$$|a| \leq \sqrt{3} a_3 = \sqrt{3} \max\{a_1, a_2, a_3\}$$

(B) is true

- Let w_1 be the point obtained by the rotation of $z_1 = 5 + 3i$ about the origin through a right angle in the anticlockwise direction, and w_2 be the point obtained by the rotation of $z_2 = 3 + 5i$ about the origin through a right angle in the clockwise direction. Then the principal argument of $w_1 - w_2$ is equal to
- $-\pi + \tan^{-1} \frac{13}{5}$
 - $-\pi - \tan^{-1} \frac{13}{5}$
 - $-\pi + \tan^{-1} \frac{18}{9}$
 - $\pi - \tan^{-1} \frac{18}{9}$
- Official Ans. by NTA (ε)

Sol. $W_1 = zii = 5(-4ii) = -4 + 5i \quad \dots (i)$

$W_2 = z(2-i) = (3+5i)(-i) = 5-3i \quad \dots (2)$

$W_1 - W_2 = -9 + 8i$

Principal argument $= \tan^{-1} \frac{8}{-9} = \pi - \tan^{-1} \frac{8}{9}$

9. An organization awarded 48 medals in event 'A', 25 in event 'B' and 18 in event 'C'. If these medals went to total 70 men and only five men got medals in all the three events, then, how many received medals in exactly two of three events?

(1) 10

(2) 9

(3) 21

(4) 10

Official Ans. by NTA (3)

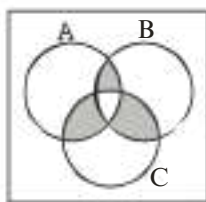
Sol. $A = 48$

$B = 25$

$C = 18$

$A \cup B \cup C = 60$ Total

$A \cap B \cap C = 5$



$A \cup B \cup C = A + B + C - A \cap B - A \cap C - B \cap C + A \cap B \cap C$

$\Rightarrow A \cap B \cap C = 48 + 25 + 18 + 5 - 60$

$= 36$

No. of men who received exactly 2 medals

$= A \cap B + A \cap C + B \cap C - 3A \cap B \cap C$

$= 36 - 10$

$= 26$

10. Let $S = \{M = [a_{ij}], a_{ij} \in \{0, 1, 2\}, 1 \leq i, j \leq 2\}$ be a sample space and $A = \{M \in S : M \text{ is invertible}\}$ be an event. Then $P(A)$ is equal to

(1) $\frac{50}{81}$

(2) $\frac{47}{81}$

(3) $\frac{49}{81}$

(4) $\frac{16}{27}$

Official Ans. by NTA (1)

Sol. $M = \begin{pmatrix} a & b \\ c & d \end{pmatrix}$, where $a, b, c, d \in \{0, 1, 2\}$
 $n(S) = 3^4 = 81$

we first find $P(A)$

$|M| = 0 \Rightarrow ad = bc$

$ad = bc = 0 \Rightarrow$ no. of $(a, b, c, d) = (3^2 - 2^2) = 5$

$\therefore ad = bc = 1 \Rightarrow$ no. of $(a, b, c, d) = 12 = 4$

$ad = bc = 2 \Rightarrow$ no. of $(a, b, c, d) = 12 = 4$

$ad = bc = 4 \Rightarrow$ no. of $(a, b, c, d) = 12 = 4$

$\therefore P(A) = \frac{31}{81} \Rightarrow p(A) = \frac{50}{81}$

11. Consider ellipses $E_k: kx^2 + ky^2 = 1, k = 1, 2, \dots, 20$

Let C_k be the circle which touches the four chords joining the end points (one on minor axis and another on major axis) of the ellipse E_k . If r_k is

the radius of the circle C_k , then the value of $\sum_{k=1}^{20} \frac{1}{r_k}$ is

(1) 30.80

(2) 32.10

(3) 33.20

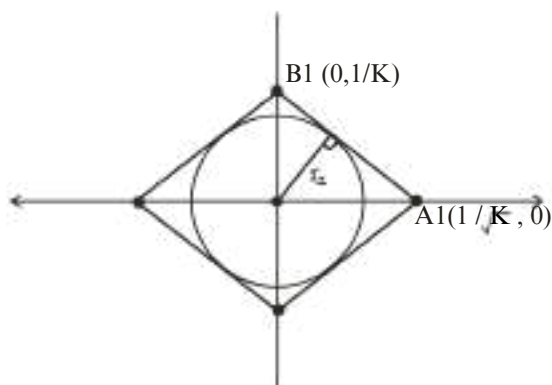
(4) 28.70

Official Ans. by NTA (1)

Sol. $Kx^2 + Ky^2 = 1$

$$\frac{x^2}{1/K} + \frac{y^2}{1/K} = 1$$

Now



Equation of

$$A_1B_1; \frac{x}{1/\sqrt{K}} + \frac{y}{1/K} = 1 \Rightarrow \sqrt{K}x + Ky = 1$$

$r_k =$ distance of (x, y) from line A_1B_1

$$r_k = \frac{|(0+0-1)|}{\sqrt{K+K^2}} = \frac{1}{\sqrt{K+K^2}}$$

$$\frac{1}{r^2 K} = K + K^2 \Rightarrow \sum_{k=1}^{20} \frac{1}{r_k^2} = \sum_{k=1}^{20} (K + K^2)$$

$$= \sum_{k=1}^{20} K + \sum_{k=1}^{20} K^2$$

$$= \frac{20 \cdot 21}{2} + \frac{20 \cdot 21 \cdot 41}{6}$$

$$= 210 + 10 \cdot 7 \cdot 41$$

$$= 210 + 2870$$

$$= 3080$$

12. The number of integral solutions x of

$$\log_{\frac{x+7}{2}} \frac{x-7}{2x-3} \geq 0$$

(1) 6

(2) 8

(3) 5

(4) 7

Official Ans. by NTA (1)

$$\log_{\frac{x+7}{2}} \frac{x-7}{2x-3} \geq 0$$

$$\text{Feasible region } x + \frac{7}{2} > 0 \Rightarrow x > -\frac{7}{2}$$

$$\text{And } x + \frac{7}{2} < 1 - x \Rightarrow x < -\frac{5}{2}$$

$$\text{And } \frac{x-7}{2x-3} \geq 0 \text{ and } 2x-3 \neq 0$$

$$\Rightarrow x < -\frac{5}{2} \text{ or } x > \frac{7}{2}$$

$$\text{Taking intersection: } x < -\frac{5}{2} \text{ and } x > \frac{7}{2}$$

$$\text{Now } \log_b a \geq 0 \text{ if } a < 1 \text{ and } b < 1$$

Or

$$a < 1 \text{ and } b > 1$$

$$C-I; x + \frac{7}{2} > 1 \text{ and } \frac{x-7}{2x-3} \geq 1$$

$$x > -\frac{5}{2} \Rightarrow (2x-3)^2 - (x-7)^2 \leq 0$$

$$(2x-3+x-7)(2x-3-x+7) \leq 0$$

$$(3x-10)(x+4) \leq 0$$

$$\text{Intersection: } x < -\frac{5}{2} \text{ and } x > \frac{7}{2}$$

$$C-II; x > \frac{7}{2} \text{ and } \frac{x-7}{2x-3} \geq 1$$

$$0 < x < \frac{7}{2} < 1 \Rightarrow \frac{x-7}{2x-3} \geq 1$$

$$-\frac{7}{2} < x < \frac{7}{2} \Rightarrow (x-7)^2 < (2x-3)^2$$

$$x < -\frac{5}{2} \text{ or } x > \frac{7}{2}$$

No common values of x .

Hence intersection with feasible region

$$\text{We get } x < -\frac{5}{2} \text{ and } x > \frac{7}{2}$$

Integral value of x are $\{-2, -1, 0, 1, 2, 3\}$

No. of integral values = 6

١٣. Area of the region $\{(x,y): x^2 + (y-2)^2 \leq 4, x^2 \geq 2y\}$ is

(١) $2p - \frac{16}{3}$

(٢) $p - \frac{8}{3}$

(٣) $p + \frac{8}{3}$

(٤) $2p + \frac{16}{3}$

Official Ans. by NTA (١)

Sol. $x^2 + (y-2)^2 \leq 4$ and $x^2 \geq 2y$

Solving circle and parabola simultaneously :

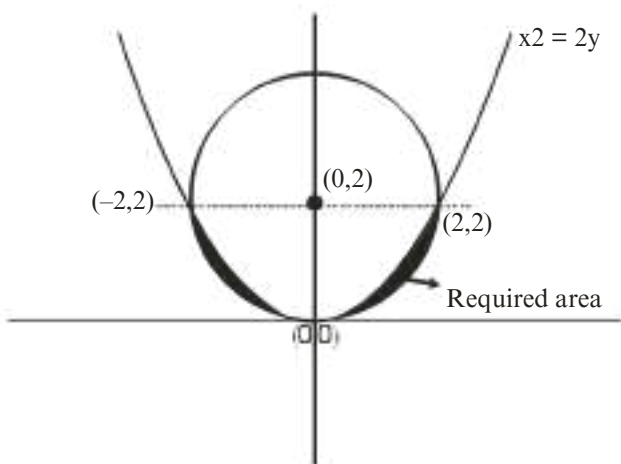
$$2y + y^2 - 4y + 4 = 4$$

$$y^2 - 2y = 0$$

$$y = 0, 2$$

Put $y = 2$ in $x^2 = 2y$ $\Rightarrow x = \pm 2$

P($2, 2$) and ($-2, 2$)



$$= 2 \int_0^2 2 - \frac{1}{4} x^2 dx = 4 - p$$

Required area $= 2 \int_0^2 x^2 dx - (4 - p)$

$$= 2 \left[\frac{x^3}{3} \right]_0^2 - 4 + p$$

$$= 2 \left[\frac{8}{3} + p - 4 \right]$$

$$= 2p - \frac{8}{3}$$

$$= 2p - \frac{16}{6}$$

١٤. Let $f: [2, 4] \rightarrow \mathbb{R}$ be a differentiable function such that $\log_e x f'(x) = f(x) - 1$,

$$x \in [2, 4] \text{ with } f(2) = \frac{1}{2} \text{ and } f(4) = \frac{1}{4}.$$

Consider the following two statements :

(A) : $f(x) \leq 1$, for all $x \in [2, 4]$

(B) : $f(x) \geq \frac{1}{8}$, for all $x \in [2, 4]$

Then,

(١) Only statement (B) is true

(٢) Neither statement (A) nor statement (B) is true

(٣) Both the statement (A) and (B) are true

(٤) Only statement (A) is true

Official Ans. by NTA (٣)

Sol. $x \ln x f'(x) + \ln x f(x) + f(x)^3 = 1, x \in [2, 4]$

$$\text{And } f(2) = \frac{1}{2}, f(4) = \frac{1}{4}$$

$$\text{Now } x \ln x dy + (1 + \ln x) y^3 dx$$

$$\frac{d}{dx} (y \times x \ln x)^3 = 1$$

$$\frac{d}{dx}$$

$$dx$$

$$= \frac{d}{dx} (x \ln x f(x) - x)^3 = 0, x \in [2, 4]$$

The function $g(x) = x \ln x f(x) - x$ is increasing in

$[2, 4]$

$$\text{And } g(2) = 2 \ln 2 \left(\frac{1}{2} \right) - 2 = \ln 2 - 2$$

$$g(4) = 4 \ln 4 \left(\frac{1}{4} \right) - 4 = \ln 4 - 4$$

$$= 2(\ln 2 - 2)$$

$$\text{Now } g(2) \leq g(x) \leq g(4)$$

$$\ln 2 - 2 \leq x \ln x f(x) - x \leq 2(\ln 2 - 2)$$

$$\frac{\ln 2 - 2}{x \ln x} + \frac{1}{\ln x} \leq f(x) \leq \frac{2 \ln 2 - 2}{x \ln x} + \frac{1}{\ln x}$$

Now for $x \in [2, 4]$

$$\frac{2 \ln 2 - 2}{x \ln x} + \frac{1}{\ln x} < \frac{2(\ln 2 - 2)}{2 \ln 2} + \frac{1}{\ln 2} = 1 - \frac{1}{\ln 2} < 1$$

$\Rightarrow f(x) \leq 1$ for $x \in [2, 4]$

Also for $x \in [2, 4]$:

$$\frac{\ln 2 - 2}{x \ln x} + \frac{1}{\ln x} \geq \frac{\ln 2 - 2}{4 \ln 4} + \frac{1}{\ln 4} = \frac{1}{8} + \frac{1}{2 \ln 2} > \frac{1}{8}$$

$\Rightarrow f(x) \geq \frac{1}{8}$ for $x \in [2, 4]$

Hence both A and B are true.

LMVT on $(y, \ln x)$ not satisfied.

Hence no such function exists.

Therefore it should be bonus.

15. Let $y = y(x)$ be a solution curve of the differential equation, $(1 - x^2 y^2) dx = y dx + x dy$.

If the line $x = 1$ intersects the curve $y = y(x)$ at $y = 2$ and the line $x = 2$ intersects the curve $y = y(x)$ at $y = a$, then a value of a is

(1) $\frac{3e^2}{2(3e^2 - 1)}$

(2) $\frac{3e^2}{2(3e^2 + 1)}$

(3) $\frac{2(3e^2 + 1)}{1 + 3e^2}$

(4) $\frac{2(3e^2 - 1)}{2(3e^2 - 1)}$

Official Ans. by NTA (4)

Sol. $(1 - x^2 y^2) dx = y dx + x dy$

$y(2) = \mu = ?$

$$dx = \frac{d(xy)}{1 - (xy)^2}$$

$$\int \frac{d(xy)}{1 - (xy)^2}$$

$$x = \frac{1}{2} \ln \left| \frac{1 + xy}{1 - xy} \right| + C$$

Put $x = 1$ and $y = 2$:

$$1 = \frac{1}{2} \ln \left| \frac{1 + 2}{1 - 2} \right| + C$$

$$C = 1 - \frac{1}{2} \ln 3$$

Now put $x = 2$:

$$2 = \frac{1}{2} \ln \left| \frac{1 + 2a}{1 - 2a} \right| + 1 - \frac{1}{2} \ln 3$$

$$1 + \frac{1}{2} \ln 3 = \frac{1}{2} \ln \left| \frac{1 + 2a}{1 - 2a} \right|$$

$$2 \ln 3 \ln \left| \frac{1 + 2a}{1 - 2a} \right| = 0$$

$$\left| \frac{1 + 2a}{1 - 2a} \right| = 3e^2$$

$$\frac{1 + 2a}{1 - 2a} = 3e^2, -3e^2$$

$$\frac{1 + 2a}{1 - 2a} = 3e^2 \Rightarrow a = \frac{3e^2 - 1}{2(3e^2 - 1)}$$

$$\text{And } \frac{1 + 2a}{1 - 2a} = -3e^2 \Rightarrow a = \frac{3e^2 + 1}{2(3e^2 - 1)}$$

16. Let A be a 2×2 matrix with real entries such that

$A' = aA + I$, where $a \in \{-1, 1\}$. If \det

$(A - A) = \lambda$, then the sum of all possible values of a is equal to

(1) $\frac{3}{2}$

(2) $\frac{5}{2}$

Official Ans. by NTA (2)

Sol. $AT = aA + I$

$$A = aAT + I$$

$$A = a(aA + I) + I$$

$$A(a^2A + a + 1)I$$

$$1 - a^2 = a + 1$$

$$A = \frac{I}{1-a} \quad \dots (1)$$

$$|A| = \frac{1}{(1-a)^2} \quad \dots (2)$$

$$|A^2 - A| = |A(A - I)| \quad \dots (3)$$

$$A - I = \frac{I}{1-a} - \frac{a}{1-a}I$$

$$|A - I| = \frac{1}{(1-a)^2} \quad \dots (4)$$

$$\text{Now } |A^2 - A| = 4$$

$$|A(A - I)| = 4$$

$$|A| |A - I| = 4$$

$$|A| = \pm 2$$

$$2(1-a)^2 = \pm a$$

$$(C_1) \quad 2(1-a)^2 = a$$

$$2a^2 - 5a + 2 = 0$$

$$a_1 + a_2 = \frac{5}{2}$$

$$\text{Sum of value of } a = \frac{5}{2}$$

17. Let (a, b, g) be the image of the point $P(x, y, z)$ in

the plane $2x + y - 3z = 6$. Then $a + b + g$ is equal to

$$(1) 10$$

$$(2) 0$$

$$(3) 12$$

$$(4) 9$$

Official Ans. by NTA (1)

$$\text{Sol. } \frac{a-2}{2} = \frac{b-3}{1} = \frac{g-5}{-3} = \frac{2a^2+3-3^2-5-6}{22+12+1-32} = 2$$

$$\frac{a-2}{2} = 2$$

$$a = 6$$

$$\frac{b-3}{1} = 2$$

$$b = 5$$

$$\frac{g-5}{-3} = 2$$

$$g = -1$$

(2, 3, 5)



(a, b, g)

$$a + b + g = 10$$

18. Let \vec{a} be a non-zero vector parallel to the line of intersection of the two planes described by $\vec{r} \cdot \vec{i} + \vec{r} \cdot \vec{j} + \vec{r} \cdot \vec{k} = 6$ and $\vec{r} \cdot \vec{i} - \vec{r} \cdot \vec{j} + \vec{r} \cdot \vec{k} = 4$. If θ is the angle

between the vector \vec{a} and the

$\vec{b} = 2\vec{i} - 2\vec{j} + \vec{k}$ and $a \cdot b = 6$ then the ordered

(θ, \vec{a}) is equal to

$$(1) \frac{\pi}{6}, 3\sqrt{6}$$

$$(2) \frac{\pi}{6}, 3\sqrt{6}$$

$$(3) \frac{\pi}{6}, 6\sqrt{6}$$

$$(4) \frac{\pi}{4}, 6\sqrt{6}$$

$$(5) \frac{\pi}{4}, 6\sqrt{6}$$

Official Ans. by NTA (3)

Sol. \vec{n}_1 and \vec{n}_2 are normal vector to the plane

$\vec{r} \cdot \vec{i} + \vec{r} \cdot \vec{j} + \vec{r} \cdot \vec{k} = 6$ and $\vec{r} \cdot \vec{i} - \vec{r} \cdot \vec{j} + \vec{r} \cdot \vec{k} = 4$ respectively

$$\vec{n}_1 = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & 1 & 1 \\ 1 & 0 & 1 \end{vmatrix} = \hat{i} - \hat{j} - \hat{k}$$

$$\vec{n}_2 = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & -1 & 1 \\ 1 & 0 & 1 \end{vmatrix} = \hat{i} + \hat{j} + \hat{k}$$

$$\vec{a} = \frac{1}{|\vec{n}_1|} \vec{n}_1$$

$$= l \begin{vmatrix} i & j & k \\ 1 & -1 & -1 \\ 1 & 1 & 1 \\ 0 & 4 & 2 \end{vmatrix} = l(-2j^{\wedge} + 2k^{\wedge})$$

$$\vec{a} \times \vec{b} = l \begin{vmatrix} i & j & k \\ 1 & -1 & -1 \\ 1 & 1 & 1 \\ 0 & 4 & 2 \end{vmatrix} = 6$$

$$\vec{r} = -2j^{\wedge} + 2k^{\wedge}$$

$$\cos q = \frac{\vec{a} \cdot \vec{r}}{|\vec{a}| |\vec{r}|}$$

$$\cos q = \frac{6}{2\sqrt{2} \cdot 3} = \frac{1}{\sqrt{2}}$$

$$q = \frac{\pi}{4}$$

$$\text{Now } |\vec{a}|^2 + |\vec{r}|^2 = |\vec{a} + \vec{r}|^2$$

$$36 + 4 = 8 + 9 = 72$$

$$|\vec{a} + \vec{r}|^2 = 36$$

$$|\vec{a} + \vec{r}| = 6$$

19. The number of elements in the set $S = \{x \in \mathbb{R} : \cos x = \cos q - \sin q + 2\}$ is

(1) 1

(2) 2

(3) 3

(4) 4

Official Ans. by NTA (3)

$$\text{Sol. } 3\cos 4q - 5\cos 2q - 2\sin 6q + 2 = 0$$

$$\Rightarrow 3\cos 4q - 3\cos 2q - 2\cos 2q - 2\sin 6q + 2 = 0$$

$$\Rightarrow 3\cos 4q - 3\cos 2q + 2\sin 2q - 2\sin 6q = 0$$

$$\Rightarrow 3\cos 2q(\cos 2q - 1) + 2\sin^2 q(\sin^2 q - 1) = 0$$

$$\Rightarrow -3\cos 2q\sin 2q + 2\sin^2 q(1 + \sin 2q \cos 2q - 1)$$

$$\Rightarrow \sin 2q \cos 2q(1 + 2\sin 2q - 3) = 0$$

$$\Rightarrow \sin 2q \cos 2q(2\sin 2q - 2) = 0$$

(C1) $\sin 2q = 0 \Rightarrow$ solution $q = 0, \pi, 2\pi$

(C2) $\cos 2q = 0 \Rightarrow$ solution $q = \frac{\pi}{4}, \frac{3\pi}{4}, \frac{5\pi}{4}, \frac{7\pi}{4}$

(C3) $\sin 2q = 1 \Rightarrow$ solution $q = \frac{\pi}{2}, \frac{3\pi}{2}, \frac{5\pi}{2}, \frac{7\pi}{2}$

No. of solution = 9

20. Let x_1, x_2, \dots, x_{100} be in an arithmetic progression, with $x_1 = 2$ and their mean equal to 200. If

$y_i = i(x_i - i), \forall i \in \{1, 2, \dots, 100\}$, then the mean of y_1, y_2, \dots, y_{100}

is

(1) 10049.5

(2) 10050.5

(3) 10049.5

(4) 10049.5

Official Ans. by NTA (3)

Sol. Mean =

$$\frac{1}{100} \sum_{i=1}^{100} (2 + (i-1)d) = 200$$

$$\Rightarrow 4 + 99d = 400$$

$$\Rightarrow d = \frac{396}{99} = 4$$

$$y_i = i(x_i - i)$$

$$= i(2 + (i-1)4 - i) = 3i^2 - 2i$$

$$\text{Mean} = \frac{1}{100} \sum_{i=1}^{100} y_i$$

$$= \frac{1}{100} \sum_{i=1}^{100} (3i^2 - 2i)$$

$$= \frac{1}{100} \left(\frac{3 \cdot 100 \cdot 101 \cdot 201}{6} - \frac{2 \cdot 100 \cdot 101}{2} \right)$$

$$= \frac{101 \cdot 201}{2} - 101 = 101 \cdot 99.5$$

$$= 10049.5$$

Official Ans. by NTA (२७३६)

Of $x^2 = 9C_2 2^7$

Of $x^7 = 9C^7 \times 2$

$$\begin{aligned}\text{Mean} &= \frac{{}^9C_1 \times 2^8 + {}^9C_2 \times 2^7 + \dots + {}^9C_7 \times 2}{7} \\ &= \frac{(1+2)^9 - {}^9C_0 \cdot 2^0 - {}^9C_8 \times 2^0 - {}^9C_9}{7} \\ &= \frac{3^9 - 29 - 18 - 1}{7} \\ &= \frac{19152}{7} = 2736\end{aligned}$$

Then the value of $(16S - 25)^{-0.5}$ is equal to

Official Ans. by NTA (۲۱۷۵)

$$\begin{aligned} \text{Sol. } S &= 109 + \frac{108}{5} + \frac{107}{5} + \dots + \frac{1}{5108} \\ S &= \frac{109}{5} + \frac{108}{5} + \dots + \frac{1}{5109} \\ 4S &= \frac{109-1}{5} + \frac{108-1}{5} + \dots + \frac{1-1}{5109} \\ &= 109 - \frac{1}{5109} \\ &= 109 - \frac{1}{4} + \frac{1}{4} - \frac{1}{5109} \\ &= 109 - \frac{1}{4} + \frac{1}{4} - \frac{1}{5109} \\ s &= 109 - \frac{1}{4} + \frac{1}{4} - \frac{1}{5109} \\ 16S &= 20 \cdot 109 - 5 + \frac{1}{5108} \\ 16S - (25) &= 2180 - 5 = 2175 \end{aligned}$$

$\neg a(\neg \phi, \neg \psi) \wedge a(\neg \phi, \phi) = p(\neg \psi) \neg \phi$, then p is equal to _____.

Official Ans. by NTA (۳۲)

Sol. $a(m, n) = \frac{1}{2} \left(\frac{m}{n} + \frac{n}{m} \right)$

If $11a(10, 6) \rightarrow 18a(11, 5) \rightarrow p(14)$ then P

$$\begin{aligned}
&= 1 \overset{2}{\underset{0}{\partial}} \overset{10}{\underset{\Pi}{\partial}} \frac{(1+3\overset{6}{t})}{\overset{1}{\underset{\overset{11}{\underset{11}{t}}}{I}}} + 1 \overset{2}{\underset{0}{\partial}} \overset{11}{\underset{t}{\partial}} (1+3\overset{6}{t}) \overset{2}{\underset{t}{\partial}} t \\
&= 1 \overset{6}{\underset{e}{\partial}} (1+3\overset{6}{t}) - \overset{6}{\underset{0}{\partial}} (1+3\overset{6}{t}) \times 3 \overset{11}{\underset{u}{\partial}} \overset{11}{\underset{u}{\partial}} + 1 \overset{2}{\underset{0}{\partial}} \overset{11}{\underset{t}{\partial}} (1+3\overset{6}{t}) \overset{2}{\underset{t}{\partial}} t \\
&= \left(\overset{11}{\underset{t}{\partial}} (1+3\overset{6}{t}) \right)^2_{\underset{0}{\partial}} \\
&= 21(7)^6 \\
&= 25(14)^6 \\
&= 32(14)^6
\end{aligned}$$

48. In an examination, 60 students have been allotted their seats as per their roll numbers. The number of ways, in which none of the ~~allotted seats is the~~ _____.

Official Ans. by NTA (११)

Sol. Derangement of 6 students

$$\begin{aligned}
 D5 &= \frac{5!}{e} - \frac{1}{1!} + \frac{1}{2!} - \frac{1}{3!} + \frac{1}{4!} - \frac{1}{5!} \cdot \frac{0}{0} \\
 &= 120 \frac{1}{e} - \frac{1}{6} + \frac{1}{24} - \frac{1}{120} \cdot \frac{0}{0} \\
 &= 60 - 20 + 5 - 1 \\
 &= 40 + 4 \\
 &= 44
 \end{aligned}$$

20. Let a line l pass through the origin and be perpendicular to the lines

$$l_1: \vec{r} = (1+j-k) + s(2j+k) \quad l_2: \vec{r} = -s\hat{i} + k\hat{j} + m(2\hat{i} + k\hat{j})$$

$$\text{and } l_2: \vec{r} = -s\hat{i} + k\hat{j} + m(2\hat{i} + k\hat{j})$$

If P is the point of intersection of l and l_1 and $Q(a, b, g)$ is the foot of perpendicular from P on l_2

then $a+b+g$ is equal to _____.

Official Ans. by NTA (0)

Sol. Let $l = (0\hat{i} + 0\hat{j} + 0\hat{k}) + g(a\hat{i} + b\hat{j} + c\hat{k})$

$$= g(a\hat{i} + b\hat{j} + c\hat{k})$$

$$a\hat{i} + b\hat{j} + c\hat{k} = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & 2 & 3 \\ 2 & 2 & 1 \end{vmatrix}$$

$$= \hat{i}(2-6) - \hat{j}(1-6) + \hat{k}(2-4)$$

$$= -4\hat{i} + 5\hat{j} - 2\hat{k}$$

$$l = g(-4\hat{i} + 5\hat{j} - 2\hat{k})$$

P is intersection of l and l_1

$$-4g = 1+2s, 5g = -1+2s, -2g = -1+3s$$

By solving these equations $g = -1, P(-1, 0, 1)$

Let $Q(-1+2m, 2m, 1+m)$

$$\vec{PQ} \times (2\hat{i} + 2\hat{j} + \hat{k}) = 0$$

$$-2 + 4m + 4m + 1 + m = 0$$

$$9m = 1$$

$$m = \frac{1}{9}$$

$$Q = \frac{7}{9}\hat{i} + \frac{2}{9}\hat{j} + \frac{10}{9}\hat{k}$$

$$9(a+b+g) = 9\left(\frac{7}{9} + \frac{2}{9} + \frac{10}{9}\right)$$

$$= 0$$

21. The number of integral terms in the expansion of

$$\left(x^2 + \frac{1}{x}\right)^{680}$$
 is equal to

Official Ans. by NTA (111)

Sol. The number of integral term in the expression of

$$\left(x^2 + \frac{1}{x}\right)^{680}$$
 is equal to

$$\text{General term} = {}^{680}C_r x^{2(680-r)} x^{-r} = {}^{680}C_r x^{1360-3r}$$

$$= {}^{680}C_{\frac{1360-r}{3}}$$

Value's of r , where $\frac{1360-r}{3}$ goes to integer

$$r = 0, 3, 6, 9, 12, \dots, 678$$

All value of r are accepted for $\frac{1360-r}{3}$

No. of integral terms = 111. The number of

ordered triplets of the truth values

of p, q and r such that the truth value of the

statement $(p \cup q) \cup (p \cup r) \cup (q \cup r)$ is True, is

equal to _____.

Official Ans. by NTA (v)

Sol.

p	q	r	$P \vee q$	$P \vee r$	$(p \vee q) \cup (p \vee r)$	$q \vee r$	$(p \vee q) \cup (p \vee r) \cup (q \vee r)$
T	T	T	T	T	T	T	T
T	T	F	T	T	T	T	T
T	F	T	T	T	T	T	T
T	F	F	T	T	T	F	F
F	T	T	T	T	T	T	T
F	T	F	T	F	F	T	T
F	F	T	F	T	F	T	T
F	F	F	F	F	F	F	T

Hence total no of ordered triplets are v

PHYSICS

SECTION-A

31. The electric field in an electromagnetic wave is

$$\text{given as } E = E_0 \sin \left[\frac{2\pi}{\lambda} x - \frac{2\pi}{c} t \right]$$

Where ω and c are angular frequency and velocity of electromagnetic wave respectively. The energy contained in a volume of 0.1 m^3 will be

(Given $\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{Nm}^2$)

(1) $28.5 \times 10^{-13} \text{ J}$ (2) $17.7 \times 10^{-13} \text{ J}$

(3) $8.85 \times 10^{-13} \text{ J}$ (4) $88.5 \times 10^{-13} \text{ J}$

Official Ans. by NTA (3)

Sol. $E = E_0 \sin \left[\frac{2\pi}{\lambda} x - \frac{2\pi}{c} t \right]$

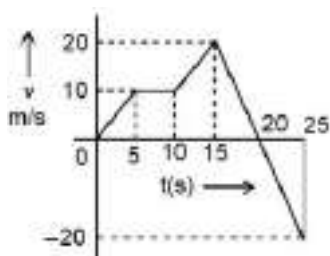
Average energy density of an em wave $E = \frac{1}{2} \epsilon_0 E^2$

Energy stored = $\frac{1}{2} \epsilon_0 E^2 \times \text{volume}$

$\frac{1}{2} \times 8.85 \times 10^{-12} \times (20)^2 \times 0.1$

$8.85 \times 10^{-13} \text{ J}$

32. From the $v-t$ graph shown, the ratio of distance to displacement in 20 s of motion



(1) $\frac{3}{5}$

(2) $\frac{1}{3}$

(3) $\frac{5}{3}$

(4) 1

Official Ans. by NTA (3)

TEST PAPER WITH SOLUTION

Sol. Area under the graph from $t=0$ to $t=20 \text{ sec} = 200 \text{ m}$

Area under the graph from $t=20$ to $t=25 \text{ sec} = 50 \text{ m}$

So distance covered $= (200 + 50) \text{ m} = 250 \text{ m}$

Displacement $= (200 - 50) \text{ m} = 150 \text{ m}$

$\frac{250}{150} = \frac{5}{3}$

33. The radii of two planets 'A' and 'B' are 'R' and 'R' and their densities are ρ and $\rho/2$ respectively. The ratio of acceleration due to gravity at their surfaces ($g_A : g_B$) will be :

(1) 1 : 16

(2) 3 : 16

(3) 3 : 4

(4) 4 : 3

Official Ans. by NTA (3)

Sol. $g = \frac{GM}{R^2} = \frac{G \rho \frac{4}{3} \pi R^3}{R^2} = \frac{4}{3} \pi G \rho R$

$\frac{g_A}{g_B} = \frac{R \rho}{\frac{1}{2} R \frac{\rho}{2}} = \frac{3}{4}$

34. A coin placed on a rotating table just slips when it is placed at a distance of 1 cm from the center. If the angular velocity of the table is halved, it will just slip when placed at a distance of _____ from the centre :

(1) 2 cm

(2) 1 cm

(3) 8 cm

(4) 4 cm

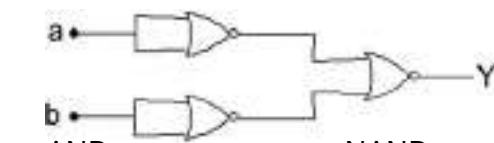
Official Ans. by NTA (4)

Sol. $f_s \max \mu mg = m \omega^2 R \Rightarrow R = \frac{\mu g}{\omega^2}$

So if ω becomes $\frac{\omega}{2}$, R will become 4R.

So distance from the center will be 4 cm.

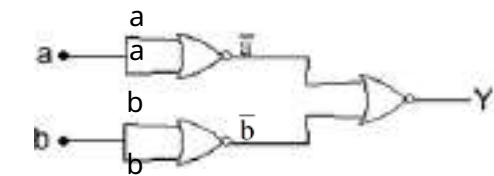
30. The logic performed by the circuit shown in figure is equivalent to :



- (1) AND (2) NAND
(3) OR (4) NOR

Official Ans. by NTA (1)

Sol.



$$Y = \bar{a} \cdot \bar{b}$$

The truth table for the given circuit will be

a	b	output
0	0	0
0	1	0
1	0	0
1	1	1

Hence it will be equivalent to AND gate.

36. A parallel plate capacitor of capacitance γ F is charged to a potential V . The energy stored in the

capacitor is E_1 . The capacitor is now connected to another uncharged identical capacitor in parallel

combination. The energy stored in the combination is E_2 . The ratio E_2/E_1 is :

Official Ans. by NTA (2)

- (1) 1 : 2

Sol. Initially

$$Q_1 = CV = (\gamma) V$$

$$E_1 = \frac{1}{2} CV^2 = \frac{1}{2} (\gamma) V^2$$

Finally

$$\text{Charge on each capacitor, } Q = \frac{Q_1}{2} = \frac{\gamma V}{2}$$

$$E_2 = 2 \left(\frac{1}{2} \frac{Q^2}{C} \right) = \frac{E_1}{2}$$

Two identical heater filaments are connected first in parallel and then in series. At the same applied voltage, the ratio of heat produced in same time for parallel to series will be :

- (1) 2 : 1 (2) 1 : 2 (3) 1 : 4 (4) 4 : 1

Official Ans. by NTA (1)

Sol. Parallel combination

$$H_p = \frac{V^2}{R} t$$

Series combination

$$H_s = \frac{V^2}{4R} t$$

38. A transmitting antenna is kept on the surface of the earth. The minimum height of receiving antenna required to receive the signal in line of sight at x km distance from it is $x \times 1.7$ m. The value of x is (Let, radius of earth $R = 6400$ km)

- (1) 120 (2) 12.0 (3) 1.20 (4) 1200

Official Ans. by NTA (1)

$$\text{Sol. } d = \sqrt{2Rh}$$

$$1.7 = \sqrt{2 \times 6400 \times h}$$

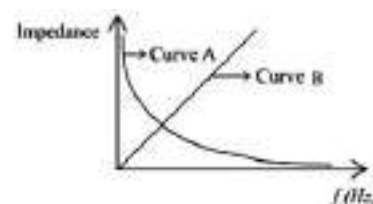
39. As per the given graph choose the correct representation for curve A and curve B.

Where X_C = reactance of pure capacitive circuit connected with A.C. source

X_L = reactance of pure inductive circuit connected with A.C. source

R = impedance of pure resistive circuit connected with A.C. source

Z = Impedance of the LCR series circuit



- (1) $A = X_C$, $B = R$ (2) $A = X_L$, $B = Z$
(3) $A = X_C$, $B = X_L$ (4) $A = X_L$, $B = R$

Official Ans. by NTA (2)

Sol. $X_c = \frac{1}{2fC}$

$X_c = \frac{1}{fC}$

Curve A

$X_L = 2fL$

$X_L = fL$

Curve B

Ex. 1 kg of water at 100°C is converted into steam at 100°C by boiling at atmospheric pressure. The volume of water changes from $1.0 \times 10^{-3} \text{ m}^3$ as a liquid to 1.671 m^3 as steam. The change in internal energy of the system during the process will be (Given latent heat of vaporisation = 2257 kJ/kg .)

Atmospheric pressure = $1 \times 10^5 \text{ Pa}$

(1) $+2090 \text{ kJ}$ (2) -2090 kJ

(3) -2426 kJ (4) $+2426 \text{ kJ}$

Official Ans. by NTA (1)

Sol. $Q = U + W$

$Q = U + W$

$mL_v + P\Delta V$

$1 \text{ kg} \times 2257 + 10^5 \times (1.671 - 10^{-3})$

$1 \times 10^5 \text{ Pa} \times 1.67 \text{ m}^3 = 1.67 \times 10^5 \text{ J}$

$2257 + 1.67 \times 10^5$

2090 kJ

Ex. The critical angle for a denser-rarer interface is 45° . The speed of light in rarer medium is $3 \times 10^8 \text{ ms}$. The speed of light in the denser medium is:

(1) $0.5 \times 10^8 \text{ m/s}$ (2) $2.12 \times 10^8 \text{ m/s}$

(3) $3.12 \times 10^8 \text{ m/s}$ (4) $1.5 \times 10^8 \text{ m/s}$

Official Ans. by NTA (2)

Sol. i_c Critical angle

$\frac{1}{\mu} \sin i_c = \sin 45^\circ$

$\frac{C}{\mu} = \frac{1}{\sqrt{2}} \Rightarrow \mu = \frac{C}{\frac{1}{\sqrt{2}}} = \frac{3 \times 10^8}{\frac{1}{\sqrt{2}}} = 2.12 \times 10^8 \text{ m/s}$

Ex. A metallic surface is illuminated with radiation of wavelength λ . the stopping potential is V_0 . If the same surface is illuminated with radiation of wavelength 2λ . the stopping potential becomes $\frac{V_0}{4}$. The threshold wavelength for this metallic surface will be -

(1) $\frac{\lambda}{2}$ (2) 4λ

(3) $\frac{3\lambda}{4}$ (4) 3λ

Official Ans. by NTA (4)

Sol. From the equation of photoelectric effect

$eV_0 = \frac{hc}{\lambda} - \frac{hc}{\lambda_0}$

$\& \frac{eV_0}{4} = \frac{hc}{2\lambda} - \frac{hc}{\lambda_0}$

$\frac{1}{4} \frac{hc}{\lambda} = \frac{hc}{2\lambda} - \frac{hc}{\lambda_0}$

$\frac{1}{4} \frac{1}{\lambda} = \frac{1}{2} - \frac{1}{\lambda_0}$

$\frac{1}{4} = \frac{1}{2} - \frac{1}{\lambda_0}$

$\lambda_0 = 3\lambda$

Ex. The free space inside a current carrying toroid is filled with a material of susceptibility 2×10^{-3} . The percentage increase in the value of magnetic field inside the toroid will be

(1) 2% (2) 0.2%

(3) 0.1% (4) 1%

Official Ans. by NTA (1)

Sol. As $X_m = \frac{\mu_r - \mu_0}{\mu_0}$

$\mu_r = 1 + X_m$

$B = \mu_r B_0$

So percentage increase in magnetic field

$\frac{B - B_0}{B_0} \times 100 = 2\%$

- | | |
|----------|-----------|
| (1) +25% | (2) - 50% |
| (3) Zero | (4) - 25% |

Official Ans. by NTA (1)

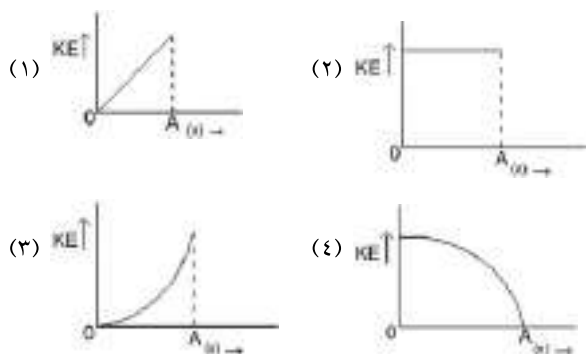
Sol. $I_s \propto \frac{NBA}{C}$ & $V_s \propto \frac{NBA}{CG}$

□V□_s If $\frac{I_G}{G}$ (galvanometer resistance) is

constant, then $V \propto \frac{1}{r}$

so percentage change in VS is also 20%.

Ex. 8. The variation of kinetic energy (KE) of a particle executing simple harmonic motion with the displacement (x) starting from mean position to extreme position (A) is given by



Official Ans. by NTA (ε)

Sol. For a particle executing SHM

$$KE = \frac{1}{2} m v^2 = A v^2 x^2$$

When $x = 0$, KE is maximum & when $x = A$, KE is zero and KE vs x graph is parabola.

86. On a temperature scale 'X'. The boiling point of water is 100°X and the freezing point is -10°X . Assume that the X scale is linear. The equivalent temperature corresponding to -10°X on the Fahrenheit scale would be:

- (1) -73°F (2) -112°F
(3) -48°F (4) -148°F

Official Ans. by NTA (ε)

Sol.

X□X freeze t□
X□X boil freeze □ ٢١٢٣٢٢

□95□□□ t□٢٢
٦٥□ □□١٥□ □ ١٨.

□٨. t□٢٢
٨. □ ١٨.

t□□١٨.□٢٢

t□□١٤٨□f

Given below are two statements :

Statements I : Astronomical unit (Au), Parsec (Pc) and Light year (ly) are units for measuring astronomical distances.

Statements II: $\text{Au} > \text{Parsec (Pc)} > \text{ly}$

In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Both Statements I and Statements II are correct.
- (2) Statements I is correct but Statements II is incorrect.
- (3) Both Statements I and Statements II are incorrect.
- (4) Statements I is incorrect but statements II is correct.

Official Ans. by NTA (۲)

Sol. $1 \text{ AU} = 1.496 \times 10^8 \text{ m}$
 $1 \text{ parsec} = 3.08 \times 10^{16} \text{ m}$
 $1 \text{ lightyear} = 9.46 \times 10^{15} \text{ m}$
 So, $\text{Au} \ll \text{Persec}$

48. Three vessels of equal volume contain gases at the same temperature and pressure. The first vessel contains neon (monoatomic), the second contains chlorine (diatomic) and third contains uranium hexafluoride (polyatomic). Arrange these on the basis of their root mean square speed (v_{rms}) and

choose the correct relation from the options given below:

- (1) $v_{rms}(\text{mono}) > v_{rms}(\text{diatomic}) > v_{rms}(\text{poly})$
 (2) $v_{rms}(\text{mono}) < v_{rms}(\text{diatomic}) < v_{rms}(\text{poly})$
 (3) $v_{rms}(\text{diatomic}) > v_{rms}(\text{mono}) > v_{rms}(\text{poly})$
 (4) $v_{rms}(\text{mono}) < v_{rms}(\text{diatomic}) < v_{rms}(\text{poly})$

Official Ans. by NTA (2)

Sol. $v_{rms}(\text{mono}) = \sqrt{\frac{3RT}{M_1}}$
 $v_{rms}(\text{diatomic}) = \sqrt{\frac{5RT}{M_2}}$
 $v_{rms}(\text{poly}) = \sqrt{\frac{6RT}{M_3}}$

So correct relation is

$$v_{rms}(\text{mono}) > v_{rms}(\text{diatomic}) > v_{rms}(\text{poly})$$

49. An average force of 120 N is applied on a machine gun firing bullets each of mass 10 g at the speed of 200 m/s to keep it in position. The number of bullets fired per second by the machine gun is

- (1) 5 (2) 50
 (3) 100 (4) 20

Official Ans. by NTA (2)

Sol. $F = nmv$

where n = number of bullets fired per second

$$n = \frac{F}{mv} = \frac{120}{0.01 \times 200} = 60$$

Two radioactive elements A and B initially have same number of atoms. The half life of A is same

as the average life of B. If λ_A and λ_B are decay constants of A and B respectively, then

choose the

correct relation from the given options.

- (1) $\lambda_A > \lambda_B$ (2) $\lambda_A < \lambda_B$
 (3) $\lambda_A = \lambda_B$ (4) $\lambda_A \neq \lambda_B$

Official Ans. by NTA (3)

Sol. $T_{1/2} = \frac{\ln 2}{\lambda}$
 $\lambda = \frac{\ln 2}{T_{1/2}}$
 $\lambda_A = \lambda_B$

SECTION-B

50. A monochromatic light is incident on a hydrogen sample in ground state. Hydrogen atoms absorb a fraction of light and subsequently emit radiation of six different wavelengths. The frequency of incident light is $x \times 10^{14}$ Hz. The value of x is

_____ (Given $h = 6.6 \times 10^{-34}$ Js)

Official Ans. by NTA (3)

Sol. $E_n = -\frac{13.6}{n^2}$
 $E_1 = -13.6$
 $E_2 = -3.4$
 $E_3 = -1.51$
 $E_4 = -0.85$
 $E_5 = -0.54$
 $E_6 = -0.38$
 $E_7 = -0.26$
 $E_8 = -0.18$
 $E_9 = -0.13$
 $E_{10} = -0.09$
 $E_{11} = -0.07$
 $E_{12} = -0.05$
 $E_{13} = -0.04$
 $E_{14} = -0.03$
 $E_{15} = -0.02$
 $E_{16} = -0.01$
 $E_{17} = -0.01$
 $E_{18} = -0.01$
 $E_{19} = -0.01$
 $E_{20} = -0.01$
 $E_{21} = -0.01$
 $E_{22} = -0.01$
 $E_{23} = -0.01$
 $E_{24} = -0.01$
 $E_{25} = -0.01$
 $E_{26} = -0.01$
 $E_{27} = -0.01$
 $E_{28} = -0.01$
 $E_{29} = -0.01$
 $E_{30} = -0.01$
 $E_{31} = -0.01$
 $E_{32} = -0.01$
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 $E_{39} = -0.01$
 $E_{40} = -0.01$
 $E_{41} = -0.01$
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 $E_{47} = -0.01$
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 $E_{60} = -0.01$
 $E_{61} = -0.01$
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 $E_{66} = -0.01$
 $E_{67} = -0.01$
 $E_{68} = -0.01$
 $E_{69} = -0.01$
 $E_{70} = -0.01$
 $E_{71} = -0.01$
 $E_{72} = -0.01$
 $E_{73} = -0.01$
 $E_{74} = -0.01$
 $E_{75} = -0.01$
 $E_{76} = -0.01$
 $E_{77} = -0.01$
 $E_{78} = -0.01$
 $E_{79} = -0.01$
 $E_{80} = -0.01$
 $E_{81} = -0.01$
 $E_{82} = -0.01$
 $E_{83} = -0.01$
 $E_{84} = -0.01$
 $E_{85} = -0.01$
 $E_{86} = -0.01$
 $E_{87} = -0.01$
 $E_{88} = -0.01$
 $E_{89} = -0.01$
 $E_{90} = -0.01$
 $E_{91} = -0.01$
 $E_{92} = -0.01$
 $E_{93} = -0.01$
 $E_{94} = -0.01$
 $E_{95} = -0.01$
 $E_{96} = -0.01$
 $E_{97} = -0.01$
 $E_{98} = -0.01$
 $E_{99} = -0.01$
 $E_{100} = -0.01$

The radius of curvature of each surface of a convex lens having refractive index 1.5 is 20 cm. The lens is now immersed in a liquid of refractive index 1.0. The ratio of power of lens in air to its power in the liquid will be $x : 1$. The value of x is _____.

Official Ans. by NTA (4)

Sol. $P = \frac{1}{f} = \frac{1}{R_1} - \frac{1}{R_2}$ by lens maker's formula

$$P' = \frac{1}{f'} = \frac{1}{R_1} - \frac{1}{R_2}$$

Dividing $\frac{P}{P'} = \frac{1.5}{1.0} = 1.5$

Q3. The equation of wave is given by

$$Y = 10 \sin \left(\frac{2\pi}{\lambda} x - \frac{2\pi}{T} t \right)$$

Where x and Y are in m and t in s. The speed of the wave is _____ km/h

Official Ans. by NTA (1152)

Sol. $V = \frac{\omega}{k} = \frac{2\pi \cdot 10}{2\pi \cdot 0.05} = \frac{10}{0.05} \text{ m/s}$
 $= \frac{10}{0.05} \cdot \frac{18}{5} \text{ km/h}$
 $= 1152 \text{ km/h}$

Q4. A force $F = 2x$ acts on a particle in the x direction where F is in newton and x is in meter.

The work done by this force during a displacement from $x = 0$ to $x = 2$ m, is _____ J.

Official Ans. by NTA (32)

Sol. $W = \int_0^2 2x \, dx$

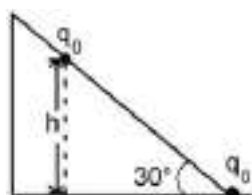
$$= \left[x^2 \right]_0^2 = \frac{2x^2}{2} \Big|_0^2$$

$$= 2 \cdot 2 = 4$$

$$= 4 \text{ J}$$

Q5. As shown in the figure, a configuration of two equal point charges ($q = +2 \mu\text{C}$) is placed on an inclined plane. Mass of each point charge is 1 g . Assume that there is no friction between charge and plane. For the system of two point charges to be in equilibrium (at rest) the height $h = x \times 10^{-2} \text{ m}$. The value of x is _____.

(Take $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ Nm}^2/\text{C}^2$, $g = 10 \text{ ms}^{-2}$)



Official Ans. by NTA (300)

Sol. For equilibrium along the plane

$$mg \sin \theta = \frac{q^2}{4\pi\epsilon_0 x^2} \cdot \frac{1}{\cos \theta}$$

$$h = \frac{1}{\sin \theta} \cdot \frac{q^2}{4\pi\epsilon_0 mg \cos \theta}$$

$$= \frac{1}{\sin 30^\circ} \cdot \frac{2 \times 10^{-6}}{9 \times 10^9 \cdot 10^{-3} \cdot \cos 30^\circ}$$

$$h = \frac{1}{0.5} \cdot \frac{2 \times 10^{-6}}{9 \times 10^9 \cdot 10^{-3} \cdot 0.866}$$

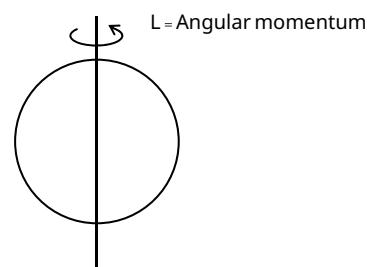
$$= 0.2 \text{ m}$$

$$= 200 \text{ mm}$$

Q6. A solid sphere of mass 0.1 kg and radius 0.1 m is rotated about one of its diameter with angular speed of 10 rad/s . If the moment of inertia of the sphere about its tangent is $x \times 10^{-4}$ times its angular momentum about the diameter. Then the value of x will be _____.

Official Ans. by NTA (30)

Sol.

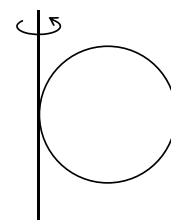


$$m = 0.1 \text{ kg}$$

$$R = 0.1 \text{ m}$$

$$\omega = 10 \text{ rad/sec}$$

moment of inertia about tangent = IT



$$I_t = I_c + MR^2$$

$$= \frac{2}{5} mR^2 + x \times 10^{-4} \cdot \frac{2}{5} mR^2$$

$$= \frac{2}{5} mR^2 (1 + x \times 10^{-4})$$

Q7. The length of wire becomes l_1 and l_2 when 100 N and 120 N tensions are applied respectively. If

$1.1l_2 = 1.1l_1$, the natural length of wire will be $\frac{l_1}{x}$.

Here the value of x is _____.

Official Ans. by NTA (2)

Sol. Let the original length be l .

When $T_1 = 100\text{ N}$, Extension $= \frac{l_1 - l}{l} \times 100$

When $T_2 = 120\text{ N}$, Extension $= \frac{l_2 - l}{l} \times 120$

Then $100 \times \frac{l_1 - l}{l} = 120 \times \frac{l_2 - l}{l}$... (1)

And $120 \times \frac{l_1 - l}{l} = 0$... (2)

$$\frac{1}{2} \times \frac{5}{6} = \frac{1}{6} \times \frac{1}{5}$$

$$5 \times 6 = 6 \times 5$$

$$6 \times 5 = 5 \times 6$$

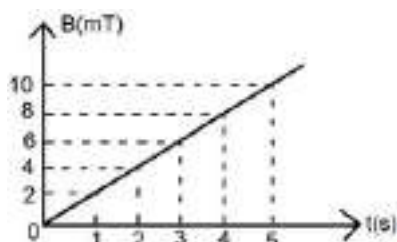
$$6 \times 5 = 5 \times 6$$

$$6 \times 5 = \frac{11l_1}{5}$$

$$l = \frac{l_1}{5}$$

$$x = 5$$

Q8. The magnetic field B crossing normally a square metallic plate of area 4 m^2 is changing with time as shown in figure. The magnitude of induced emf in the plate during $t = 2\text{ s}$ to $t = 4\text{ s}$, is _____ mV

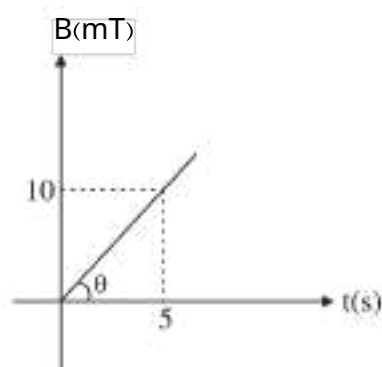


Official Ans. by NTA (4)

Sol. $m \tan \theta = \frac{10}{5} = 2$

$$B = 2t$$

$$B = 2t$$



$$\frac{dB}{dt} = \frac{10}{5} = 2 \text{ T/s}$$

$$\epsilon = \frac{d\Phi}{dt} = \frac{d(BA)}{dt} = A \frac{dB}{dt}$$

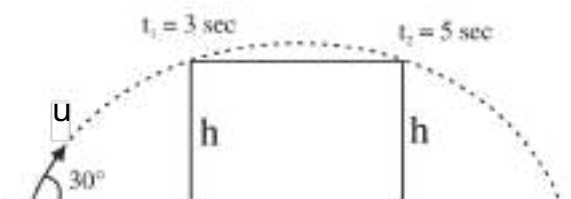
Q9. A projectile fired at 30° to the ground is observed to be at same height at time 2 s and 8 s after

projection. during its flight. The speed of projection of the projectile is _____ ms

(Given $g = 10\text{ m/s}^2$)

Official Ans. by NTA (40)

Sol. Time of flight $t = t_1 + t_2 = 10\text{ sec}$



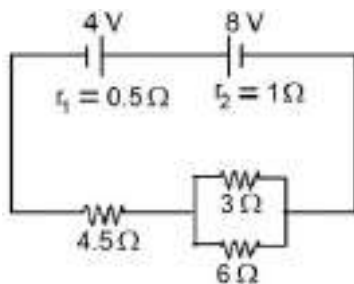
$$T = \frac{u \sin 2\theta}{g}$$

$$10 = \frac{u \sin 60^\circ}{10}$$

$$u = 40\text{ m/s}$$

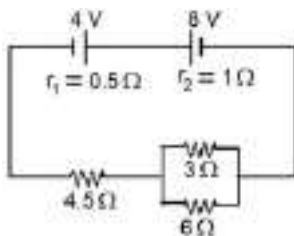
Q. In the circuit diagram shown in figure given below,
the current flowing through resistance $\frac{x}{y}$ A.

The value of x is _____.

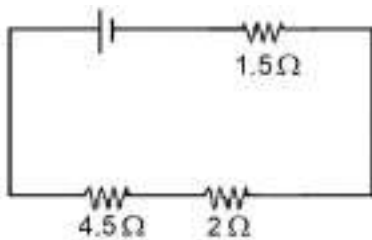


Official Ans. by NTA ()

Sol.

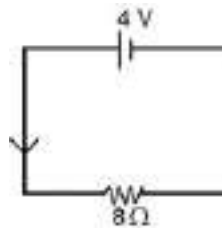


$$E_{\text{eq}} = E_1 + E_2 = 4 + 8 = 12 \text{ V}$$

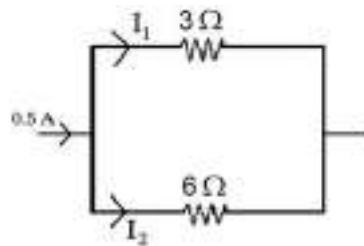


$$\frac{1}{R} = \frac{1}{r_1} + \frac{1}{r_2} + \frac{1}{R}$$

$$R = \frac{1}{\frac{1}{r_1} + \frac{1}{r_2} + \frac{1}{R}}$$



$$I = \frac{E}{R} = \frac{4}{8} = 0.5 \text{ A}$$



$$I_1 = \frac{E}{R} = \frac{4}{3} = 1.33 \text{ A}$$

$$I_2 = \frac{E}{R} = \frac{4}{6} = 0.67 \text{ A}$$

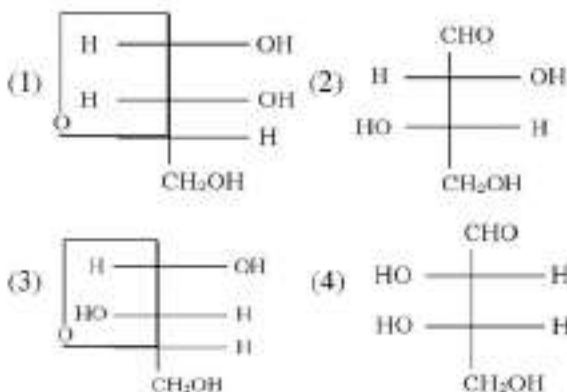
$$I_1 = \frac{E}{R} = \frac{4}{3} = 1.33 \text{ A}$$

SECTION-A

61. L-isomer of tetrose X ($C_4H_8O_4$) gives positive Schiff's test and has two chiral carbons. On acetylation, 'X' yields triacetate. 'X' also undergoes following reactions



'X' is

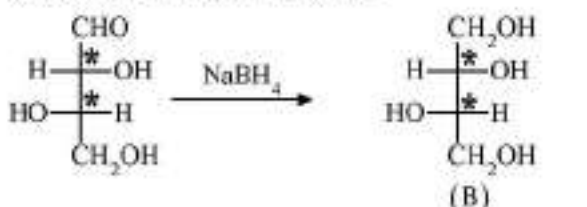


Official Ans. by NTA (2)

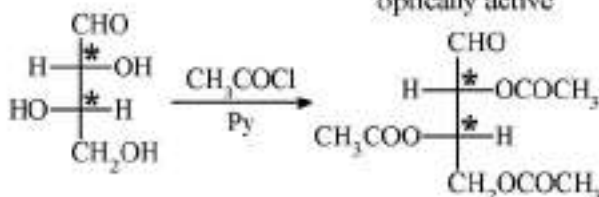
Sol.



L-tetrose with two chiral centre



optically active



(x) gives positive schiff's test due -CHO group

(x) is L-tetrose.

62. The polymer X - consists of linear molecules and is closely packed. It is prepared in the presence of triethylaluminium and titanium tetrachloride under low pressure. The polymer X is -

- (1) Polyacrylonitrile
- (2) Low density polythene
- (3) Polytetrafluoroethane
- (4) High density polythene

Official Ans. by NTA (4)

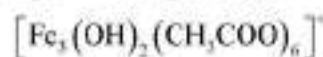
Sol. Ethene undergoes addition polymerisation to high density polythene in the presence of catalyst such as $AlEt_3$ and $TiCl_4$ (Ziegler - Natta catalyst) at a temperature of 333 K to 343 K and under a pressure of 6-7 atmosphere.

63. When a solution of mixture having two inorganic salts was treated with freshly prepared ferrous sulphate in acidic medium, a dark brown ring was formed whereas on treatment with neutral $FeCl_3$, it gave deep red colour which disappeared on boiling and a brown red ppt was formed. The mixture contains

- (1) CH_3COO^- & NO_3^-
- (2) $C_2O_4^{2-}$ & NO_3^-
- (3) SO_3^{2-} & CH_3COO^-
- (4) SO_3^{2-} & $C_2O_4^{2-}$

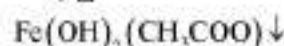
Official Ans. by NTA (1)

Sol. $CH_3COO^- + FeCl_3 \rightarrow Fe(CH_3COO)_3$ or

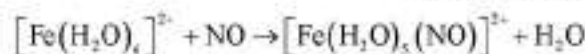
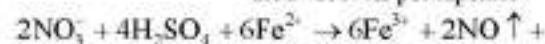


Blood red colour

$\downarrow \Delta$



Red-brown precipitate



Brown

64. Given below are two statements : one is labelled as Assertion A and the other is labelled as Reason R :

Assertion A : In the photoelectric effect, the electrons are ejected from the metal surface as soon as the beam of light of frequency greater than threshold frequency strikes the surface.

Reason R : When the photon of any energy strikes an electron in the atom, transfer of energy from the photon to the electron takes place.

In the light of the above statements, choose the most appropriate answer from the options given below :

- (1) Both A and R are correct but R is NOT the correct explanation of A
- (2) A is correct but R is not correct
- (3) Both A and R are correct and R is the correct explanation of A
- (4) A is not correct but R is correct

Official Ans. by NTA (2)

Sol. There is a characteristic minimum frequency, or "threshold frequency," for each metal below which the photoelectric effect is not seen. The ejected electrons leave with a specific amount of kinetic energy at a frequency $\nu > \nu_0$ with an increase in light frequency of these electron kinetic energies also rise.

65. 25 mL of silver nitrate solution (1 M) is added dropwise to 25 mL of potassium iodide (1.05 M) solution. The ion(s) present in very small quantity in the solution is/are

- (1) NO_3^- only
- (2) K^+ only
- (3) Ag^+ and I^- both
- (4) I^- only

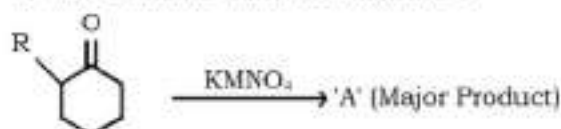
Official Ans. by NTA (3)

Sol. $\text{AgNO}_3 + \text{KI} \rightarrow \text{AgI} \downarrow + \text{KNO}_3$

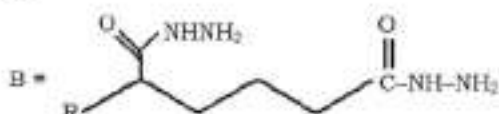
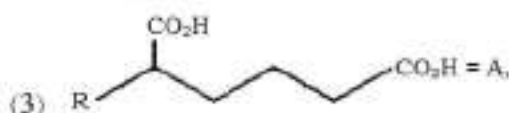
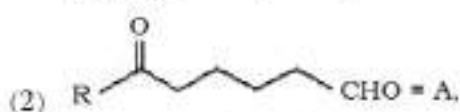
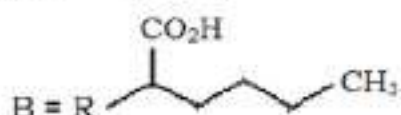
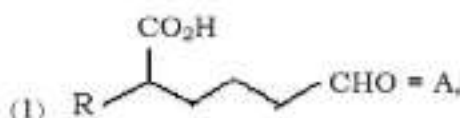
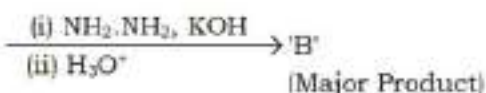


AgI is a insoluble salt so concentration Ag^+ and I^- will be negligible.

66. 'A' and 'B' in the below reactions are :

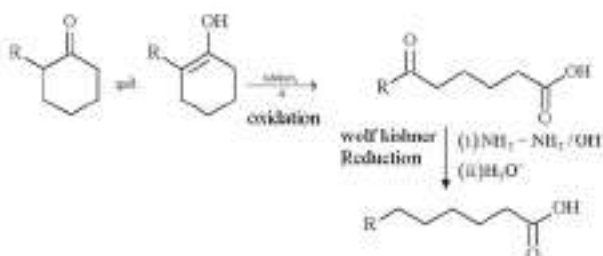


(R = alkyl)



Official Ans. by NTA (4)

Sol.

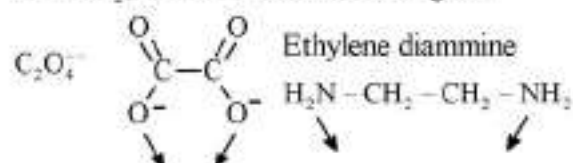


67. The set which does not have ambidentate ligand(s) is

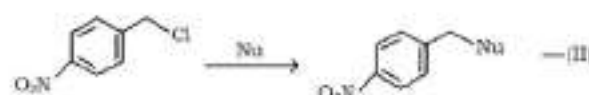
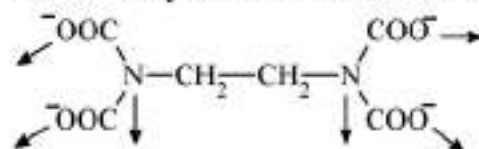
- (1) $\text{C}_2\text{O}_4^{2-}$, ethylene diammine, H_2O
- (2) EDTA^{4-} , NCS^- , $\text{C}_2\text{O}_4^{2-}$
- (3) NO_2^- , $\text{C}_2\text{O}_4^{2-}$, EDTA^{4-}
- (4) $\text{C}_2\text{O}_4^{2-}$, NO_2^- , NCS^-

Official Ans. by NTA (1)

Sol. NO_2^- , NCS^- are ambidentate ligand



EDTA Ethylene diamine tetra acetate



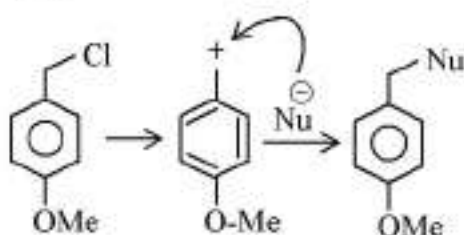
Where Nu = Nucleophile

Find out the correct statement from the options given below for the above 2 reactions.

- (1) Reaction (I) is of 2nd order and reaction (II) is of 1st order
- (2) Reaction (I) and (II) both are of 2nd order
- (3) Reaction (I) is of 1st order and reaction (II) is of 2nd order
- (4) Reactions (I) and (II) both are of 1st order

Official Ans. by NTA (3)

Sol.



Electron Donating group
 S_N^1 Mech. : 1st order



Electron withdrawing group
 S_N^2 Mech. : 2nd order

69. For elements B, C, N, Li, Be, O and F the correct order of first ionization enthalpy is

- (1) $\text{Li} < \text{Be} < \text{B} < \text{C} < \text{N} < \text{O} < \text{F}$
- (2) $\text{B} > \text{Li} > \text{Be} > \text{C} > \text{N} > \text{O} > \text{F}$
- (3) $\text{Li} < \text{B} < \text{Be} < \text{C} < \text{O} < \text{N} < \text{F}$
- (4) $\text{Li} < \text{Be} < \text{B} < \text{C} < \text{O} < \text{N} < \text{F}$

Official Ans. by NTA (3)

Sol. First I.E.

$\text{F} > \text{N} > \text{O} > \text{C} > \text{Be} > \text{B} > \text{Li}$

Li – 520 kJ/mol

Be – 899 kJ/mol

B – 801 kJ/mol

C – 1086 kJ/mol

N – 1402 kJ/mol

O – 1314 kJ/mol

F – 1681 kJ/mol

70. Match List-I with List-II :

List-I Species	List-II Geometry/Shape
A. H_3O^+	I. Tetrahedral
B. Acetylide anion	II. Linear
C. NH_4^+	III. Pyramidal
D. ClO_2^-	IV. Bent

Choose the correct answer from the options given below :

- (1) A-III, B-II, C-I, D-IV
- (2) A-III, B-I, C-II, D-IV
- (3) A-III, B-IV, C-I, D-II
- (4) A-III, B-IV, C-II, D-I

Official Ans. by NTA (1)

Sol. Molecule/ion Hybridisation Shape

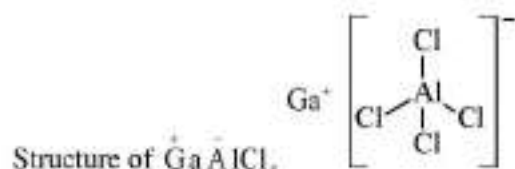
H_3O^+	sp^3	Pyramidal	
Acetylide	sp	linear	$\text{C} \equiv \text{C}$
NH_4^+	sp^3	tetrahedral	
ClO_2^-	sp^3	Bent	

71. For compound having the formula GaAlCl_4 , the correct option from the following is

- Ga is more electronegative than Al and is present as a cationic part of the salt GaAlCl_4
- Oxidation state of Ga in the salt GaAlCl_4 is +3.
- Cl forms bond with both Al and Ga in GaAlCl_4
- Ga is coordinated with Cl in GaAlCl_4

Official Ans. by NTA (1)

Sol. Gallous tetrachloro aluminate $\text{Ga}^+ \text{AlCl}_4^-$



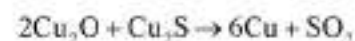
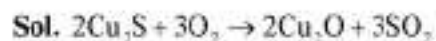
Ga is cationic part of salt GaAlCl_4 .

72. In the extraction process of copper, the product obtained after carrying out the reactions

- $2\text{Cu}_2\text{S} + 3\text{O}_2 \rightarrow 2\text{Cu}_2\text{O} + 2\text{SO}_2$
- $2\text{Cu}_2\text{O} + \text{Cu}_2\text{S} \rightarrow 6\text{Cu} + \text{SO}_2$ is called

- Blister copper
- Copper scrap
- Reduced copper
- Copper matte

Official Ans. by NTA (1)



Blister copper

Due to evolution of SO_2 , the solidified copper formed has a blistered look and is referred to as blister copper.

73. Match List-I with List-II :

List-I	List-II
A. K	I. Thermonuclear reactions
B. KCl	II. Fertilizer
C. KOH	III. Sodium potassium pump
D. Li	IV. Absorbent of CO_2

Choose the correct answer from the options given below :

- A-III, B-II, C-IV, D-I
- A-IV, B-I, C-III, D-II
- A-IV, B-III, C-I, D-II
- A-III, B-IV, C-II, D-I

Official Ans. by NTA (1)

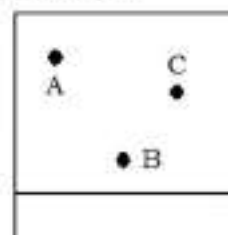
Sol. K^+ – Sodium – Potassium Pump

KCl – Fertiliser

KOH – absorber of CO_2

Li – used in thermonuclear reactions

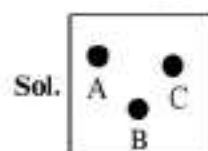
74. Thin layer chromatography of a mixture shows the following observation :



The correct order of elution in the silica gel column chromatography is

- A, C, B
- B, C, A
- C, A, B
- B, A, C

Official Ans. by NTA (1)



According to the observation, A is more mobile and interacts with the mobile phase more than C, and C is more drawn to the mobile phase than B.

Hence, the correct order of elution in the silica gel column chromatography is – $B < C < A$

75. Which of the following complex has a possibility to exist as meridional isomer?

- (1) $[\text{Co}(\text{NH}_3)_3(\text{NO}_2)_3]$
- (2) $[\text{Co}(\text{en})_3]$
- (3) $[\text{Co}(\text{en})_2\text{Cl}_2]$
- (4) $[\text{Pt}(\text{NH}_3)_2\text{Cl}_2]$

Official Ans. by NTA (1)

Sol. $[\text{MA}_3\text{B}_3]$ type of compound exists as facial and meridional isomer.



76. Given below are two statements :

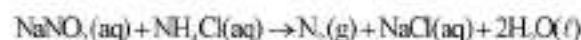
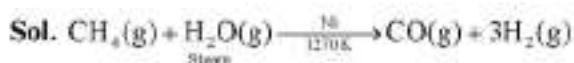
Statement-I : Methane and steam passed over a heated Ni catalyst produces hydrogen gas.

Statement-II : Sodium nitrite reacts with NH_4Cl to give H_2O , N_2 and NaCl .

In the light of the above statements, choose the most appropriate answer from the options given below :

- (1) Both the statements I and II are correct
- (2) Both the statements I and II are incorrect
- (3) Statement I is incorrect but Statement II is correct
- (4) Statement I is correct but Statement II is incorrect

Official Ans. by NTA (1)



77. Given below are two statements :

Statement I : If BOD is 4 ppm and dissolved oxygen is 8 ppm, then it is a good quality water.

Statement II : If the concentration of zinc and nitrate salts are 5 ppm each, then it can be a good quality water.

In the light of the above statements, choose the most appropriate answer from the options given below :

- (1) Both the statements I and II are incorrect
- (2) Statement I is incorrect but Statement II is correct
- (3) Both the statements I and II are correct
- (4) Statement I is correct but Statement II is incorrect

Official Ans. by NTA (3)

Sol. Clean water would have BOD value of less than 5 ppm.

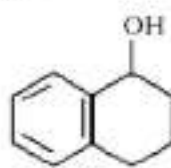
Maximum limit of Zn in clean water

$$= 5.0 \text{ ppm or } \text{mg dm}^{-3}$$

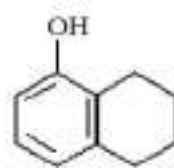
Maximum limit of NO_3^- in clean water

$$= 50 \text{ ppm or } \text{mg dm}^{-3}$$

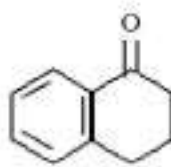
78. Arrange the following compounds in increasing order of rate of aromatic electrophilic substitution reaction



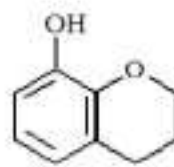
(a)



(b)



(c)

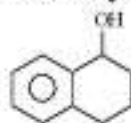


(d)

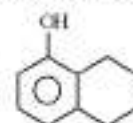
- (1) d, b, c, a
- (2) b, c, a, d
- (3) c, a, b, d
- (4) d, b, a, c

Official Ans. by NTA (3)

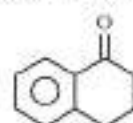
Sol. Benzene becomes more reactive towards EAS when any substituent raises the electron density.



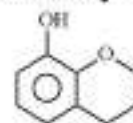
a
-CH₃ has
+H effect



b
-OH has +R
while -CH₃-
group has +H
effect.



c
-C(=O)- has -R
effect



d
-OH and -O- both
show +R effect

Correct order

$$c < a < b < d$$

79. The complex that dissolves in water is

- (1) $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$
- (2) $[\text{Fe}_3(\text{OH})_2(\text{OAc})_6]\text{Cl}$
- (3) $\text{K}_3[\text{Co}(\text{NO}_2)_6]$
- (4) $(\text{NH}_4)_3[\text{As}(\text{Mo}_3\text{O}_{10})_4]$

Official Ans. by NTA (2)

Allen Ans. (2)

Sol. $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$ Prussian Blue—water insoluble

$\text{K}_3[\text{Co}(\text{NO}_2)_6]$ very poorly water soluble

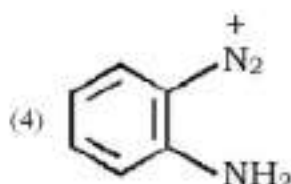
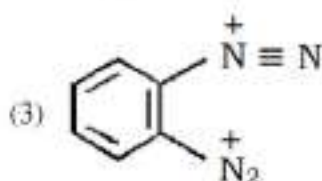
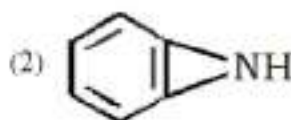
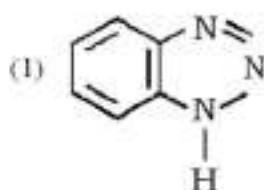
$(\text{NH}_4)_3[\text{As}(\text{MO}_3\text{O}_{10})_3]$ water insoluble

ammonium arseno molybdate

$[\text{Fe}_3(\text{OH})_2(\text{OAc})_6]\text{Cl}$ is water soluble.

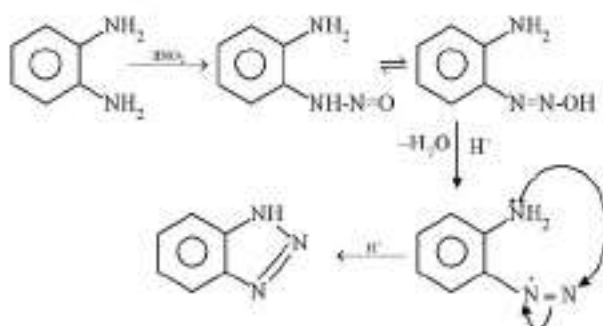


'X' is



Official Ans. by NTA (1)

Sol. Orthophenyl amine.



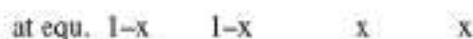
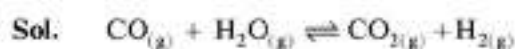
SECTION-B

81. A mixture of 1 mole of H_2O and 1 mole of CO is taken in a 10 litre container and heated to 725 K. At equilibrium 40% of water by mass reacts with carbon monoxide according to the equation :



The equilibrium constant $K_C \times 10^2$ for the reaction is _____. (Nearest integer)

Official Ans. by NTA (44)



at equilibrium 40% by mass water reacts with CO

$$x = 0.4 \quad 1 - x = 0.6$$

$$K_C = \frac{[\text{CO}_2][\text{H}_2]}{[\text{CO}][\text{H}_2\text{O}]} = \frac{0.4 \times 0.4}{0.6 \times 0.6} = 0.44$$

$$K_C \times 10^2 = 44$$

82. The ratio of spin-only magnetic moment values $\mu_{\text{eff}}[\text{Cr}(\text{CN})_6]^{3-} / \mu_{\text{eff}}[\text{Cr}(\text{H}_2\text{O})_6]^{3+}$ is _____.

Official Ans. by NTA (1)

Sol. Spin magnetic moment of $[\text{Cr}(\text{CN})_6]^{3-} (t_{2g}^3 e_g^0)$

$$\mu_1 = \sqrt{3(3+2)} = \sqrt{15} \text{ BM}$$

Spin magnetic moment of $[\text{Cr}(\text{H}_2\text{O})_6]^{3+} (t_{2g}^3 e_g^0)$

$$\mu_2 = \sqrt{3(3+2)} = \sqrt{15} \text{ BM}$$

$$\frac{\mu_1}{\mu_2} = \frac{\sqrt{15}}{\sqrt{15}} = 1$$

83. An atomic substance A of molar mass 12 g mol^{-1} has a cubic crystal structure with edge length of 300 pm. The no. of atoms present in one unit cell of A is _____. (Nearest integer)

Given the density of A is 3.0 g mL^{-1} and $N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$

Official Ans. by NTA (4)

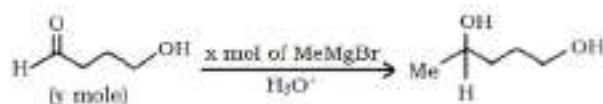
Sol. $d = 3 \text{ g/cc}$ $M = 12 \text{ g/mol}$

$a = 300 \text{ pm} = 3 \times 10^{-8} \text{ cm}$

$$Z = \frac{d \times N_A \times a^3}{M} = \frac{3 \times 6.02 \times 10^{23} \times (3 \times 10^{-8})^3}{12}$$

$= 4.06 \approx 4$

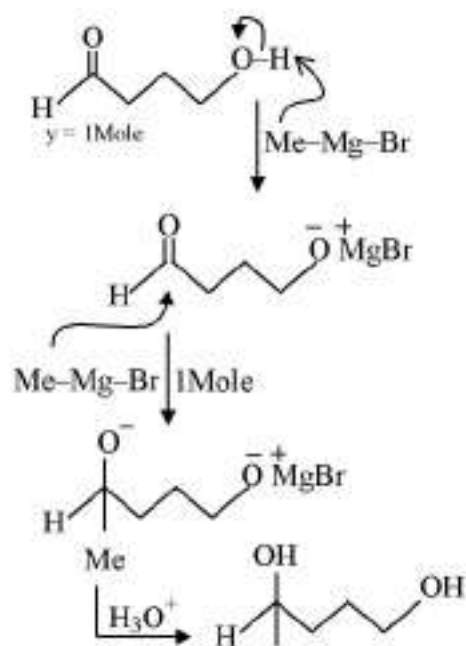
84.



The ratio x/y on completion of the above reaction is _____.

Official Ans. by NTA (2)

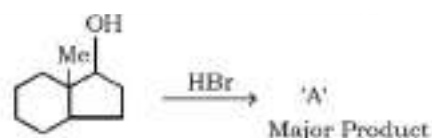
Sol.



$\therefore x = 2 \text{ mole}$

$\frac{x}{y} = \frac{2}{1} = 2$

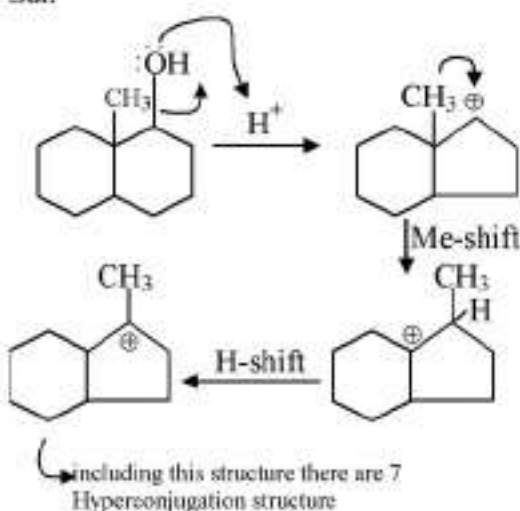
85.



The number of hyperconjugation structures involved to stabilize carbocation formed in the above reaction is _____.

Official Ans. by NTA (7)

Sol.



86.

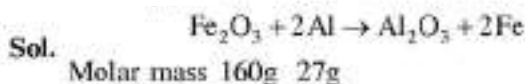
Solid fuel used in rocket is a mixture of Fe_2O_3 and Al (in ratio 1 : 2). The heat evolved (kJ) per gram of the mixture is _____ (Nearest integer)

Given : $\Delta H_f^\circ (\text{Al}_2\text{O}_3) = -1700 \text{ kJ mol}^{-1}$

$\Delta H_f^\circ (\text{Fe}_2\text{O}_3) = -840 \text{ kJ mol}^{-1}$

Molar mass of Fe, Al and O are 56, 27 and 16 g mol^{-1} respectively.

Official Ans. by NTA (4)



$$\begin{aligned}
 (\Delta H_f^\circ)_{\text{reaction}} &= [(\Delta H_f^\circ)_{\text{Al}_2\text{O}_3} + 2(\Delta H_f^\circ)_{\text{Fe}}] - [(\Delta H_f^\circ)_{\text{Fe}_2\text{O}_3} + 2(\Delta H_f^\circ)_{\text{Al}}] \\
 &= [-1700 + 0] - [-840 + 0] \\
 &= -860 \text{ kJ/mol}
 \end{aligned}$$

Total mass of mixture = $\text{Fe}_2\text{O}_3 + \text{Al}$ (1 : 2 molar ratio)
 $= 160 + 2 \times 27$
 $= 214 \text{ g/mol}$

Heat evolved per gram = $\frac{860}{214} = 4 \text{ kJ/g}$

87. A solution of sugar is obtained by mixing 200 g of its 25% solution and 500 g of its 40% solution (both by mass). The mass percentage of the resulting sugar solution is _____. (Nearest integer)

Official Ans. by NTA (36)

Sol. Total mass of sugar in mixture of 25% of 200 and 40% of 500 g

$$\text{Sugar solution} = 0.25 \times 200 + 0.40 \times 500$$

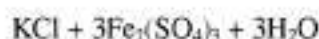
$$= 50 + 200 = 250 \text{ g}$$

$$\text{Total mass of solution} = 200 + 500 = 700 \text{ g}$$

$$\text{Mass of sugar in solution} = \frac{250}{700} \times 100 = 35.7\%$$

$$\approx 36\%$$

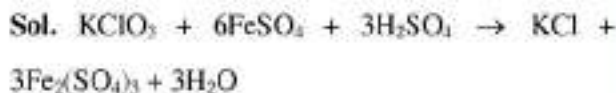
88. $\text{KClO}_3 + 6\text{FeSO}_4 + 3\text{H}_2\text{SO}_4 \rightarrow$



The above reaction was studied at 300 K by monitoring the concentration of FeSO_4 in which initial concentration was 10 M and after half an hour became 8.8 M. The rate of production of $\text{Fe}_2(\text{SO}_4)_3$ is _____ $\times 10^{-6} \text{ mol L}^{-1} \text{ s}^{-1}$.

(Nearest integer)

Official Ans. by NTA (333)



$$\begin{aligned} \text{ROR} &= -\frac{\Delta[\text{KClO}_3]}{\Delta t} = \frac{-1}{6} \frac{\Delta[\text{FeSO}_4]}{\Delta t} \\ &= \frac{+1}{3} \frac{\Delta[\text{Fe}_2(\text{SO}_4)_3]}{\Delta t} \end{aligned}$$

$$\begin{aligned} \frac{\Delta[\text{Fe}_2(\text{SO}_4)_3]}{\Delta t} &= \frac{1}{2} \frac{-\Delta[\text{FeSO}_4]}{\Delta t} \\ &= \frac{1}{2} \frac{(10 - 8.8)}{30 \times 60} \end{aligned}$$

$$= 0.333 \times 10^{-3}$$

$$= 333 \times 10^{-6} \text{ mol litre}^{-1} \text{ sec}^{-1}$$

89. 0.004 M K_2SO_4 solution is isotonic with 0.01 M glucose solution. Percentage dissociation of K_2SO_4 is _____. (Nearest integer)

Official Ans. by NTA (75)

Sol. Isotonic solutions,

$$\pi_{\text{K}_2\text{SO}_4} = \pi_{\text{Glucose}}$$

$$i \times 0.004 \times RT = 0.01 \times RT$$

$$i = 2.5$$

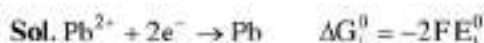
$$\text{For } \text{K}_2\text{SO}_4 \quad \{\text{for dissociation } i = 1 + (n - 1)\alpha\}$$

$$\text{DOD}(\alpha) = \frac{i - 1}{n - 1} = \frac{2.5 - 1}{3 - 1} = 0.75$$

$$\% \text{ dissociation} = 75$$

90. In an electrochemical reaction of lead, at standard temperature, if $E^\circ_{(\text{Pb}^{2+}/\text{Pb})} = m \text{ Volt}$ and $E^\circ_{(\text{Pb}^{4+}/\text{Pb})} = n \text{ Volt}$, then the value of $E^\circ_{(\text{Pb}^{2+}/\text{Pb}^{4+})}$ is given by $m - xn$. The value of x is _____. (Nearest integer)

Official Ans. by NTA (2)



$$\Delta G_3^\circ = \Delta G_1^\circ - \Delta G_2^\circ$$

$$-2FE_3^\circ = 2F(2n - m)$$

$$E_3^\circ = m - 2n = m - xn$$

$$\text{Hence } x = 2$$