FINAL JEE-MAIN EXAMINATION - JANUARY, 2024

(Held On Wednesday 31 January, 2024)

TIME: 9:00 AM to 12:00 NOON

M ATHEM ATICS

SECTION-A

- For $\cdot > c > b > a$, let $(a + b \gamma c)x + (b + c \gamma a)x$ + $(c + a - \gamma b) = \cdot$ and $\square \square \gamma$ be one of its root. Then , among the two statements
 - mean of a and c
 - (II) If $\Box\Box\Box\cdot \cap\Box$, then b may be the geometric mmon root, where c, d, e \Box R, then d : c : e mean of a and c
 - (1) Both (I) and (II) are true
 - (Y) Neither (I) nor (II) is true
 - (٣) Only (II) is true
 - (٤) Only (I) is true

Ans.(1)

Sol.
$$f(x) = (a + b - \gamma c) x + (b + c - \gamma a) x + (c + a - \gamma b)$$

 $f(x) = a + b - \gamma c + b + c - \gamma a + c + a - \gamma b = \cdot$
 $f(\gamma) = \cdot$

$$\Box \frac{c \Box a \Box^{\mathsf{Y}} b}{a \Box b \Box^{\mathsf{Y}} C}$$

$$\Box 1 \Box \frac{c \Box a \Box r}{a \Box b \Box b} \Box \cdot$$

$$b+c> \forall a \text{ and } b$$

therefore, b cannot be G.M. between a and c. If. .□□□\

$$0 \Box \frac{c \Box a \Box b}{a \Box b \Box c} \Box v$$

$$b < c$$
 and $b \square \frac{a \square c}{Y}$

TEST PAPER WITH SOLUTION

Let a be the sum of all coefficients in the expansion of $(1 - YX + YX)^{-YYYY} (Y - \xi X + YX)$

If the equations

 $cx^{r}+dx+e=\cdot$ and $rbx+ax^{r}+\xi=\cdot$ have a equals

- (1) 7:1: 8
- (1) 2:1:2
- (٣) 1 : 7 : 8
- (٤) 1:1: ٤

Ans. (ξ)

Sol. Put x = 1

ΠaΠν

Using L'HOPITAL Rule

Now, $c\dot{x} + dx + e = \cdot$, $x + \dot{x} + \dot{\xi} = \cdot$ $(D > \cdot)$

$$\begin{array}{c|c} c & d & e \\ \hline \end{array}$$

If the foci of a hyperbola are same as that of the ellipse $\frac{X^{\intercal}}{4} \Box_{\Upsilon A}^{\Upsilon T} \Box_{\Upsilon}$ and the eccentricity of the

hyperbola is $\frac{1}{\Lambda}$ times the eccentricity of the ellipse, then the smaller focal distance of the point, ξ \uparrow on the hyperbola, is equal to

Therefore, b may be the G.M. between a and c. Ans. (1)

Sol.
$$\frac{X^{\tau}}{q} \Box \frac{Y^{\tau}}{r_0} \Box \cdot$$

$$a = r, b = 0$$

$$e \Box \sqrt{\Box \frac{q}{r_0}} \Box \frac{\xi}{r_0} \Box \text{foci} \Box \Box \cdot \Box \text{be} \Box = (\cdot, \pm \xi)$$

$$\Box e H \Box \frac{\xi}{r_0} \Box \frac{r_0}{r_0} \Box \frac{r_0}{r_0$$

Let equation hyperbola

$$A_{x} \cap B_{x} \cap A_{1}$$

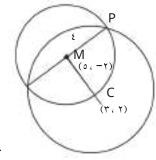
Directrix :y
$$\square \stackrel{B}{\underset{r \in \mathbb{R}}{\square}} \square \stackrel{1}{\underset{q}{\longrightarrow}}$$

$$PS = e \square PM \square \stackrel{r}{\underset{r}{\longrightarrow}} \square \stackrel{7}{\underset{q}{\longrightarrow}} \square \stackrel{1}{\underset{q}{\longrightarrow}}$$

$$\square \vee \sqrt{\stackrel{\mathsf{Y}}{\circ}} \square \frac{\wedge}{\mathsf{Y}}$$

If one of the diameters of the circle x + y - y + x + y٤y + ١٣ = • is a chord of another circle C، whose center is the point of intersection of the lines $\forall x + y \in \mathbb{R}$ ry = 17 and rx - ry = 0, then the radius of the circle C is

Ans. (٣)



Sol.

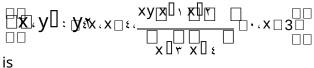
$$TX + TY = 1T$$

 $TX - TY = 0$

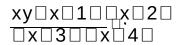
Center of given circle is (o, -Y)

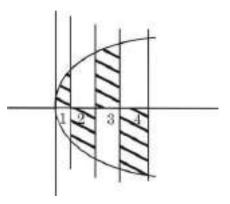
Radius Tolls 1405

The area of the region

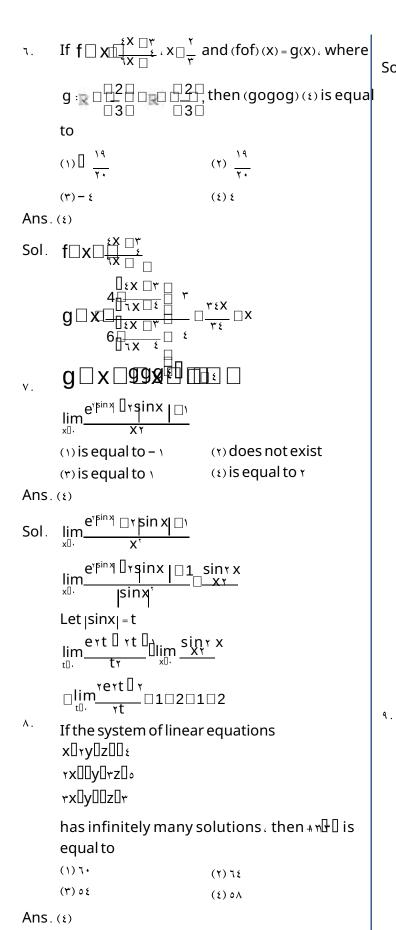


Ans.(٤) Sol.





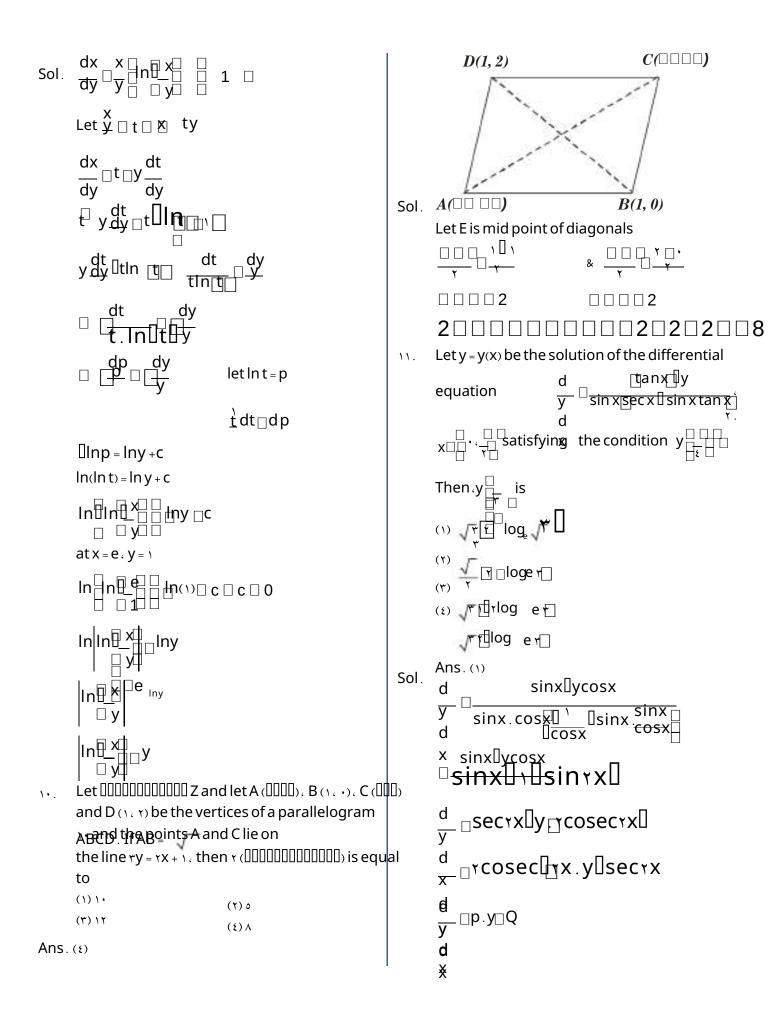
$$\Box 2 \Box_{3}^{7} \Box \ddot{x}^{/7} \Box \Box 0 \Box \ddot{x}^{77}$$

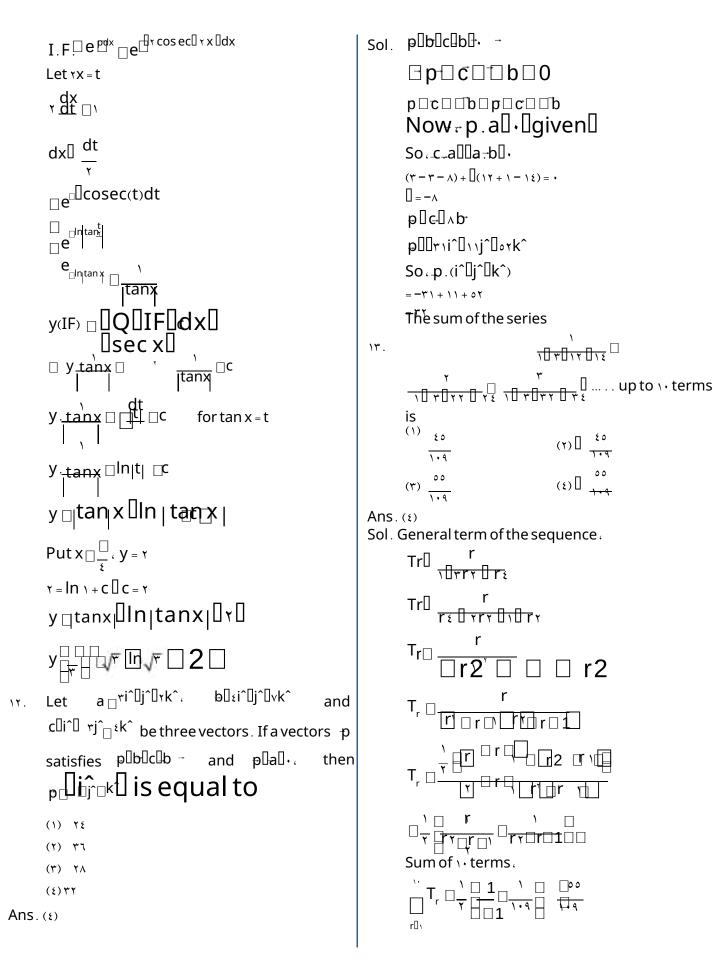


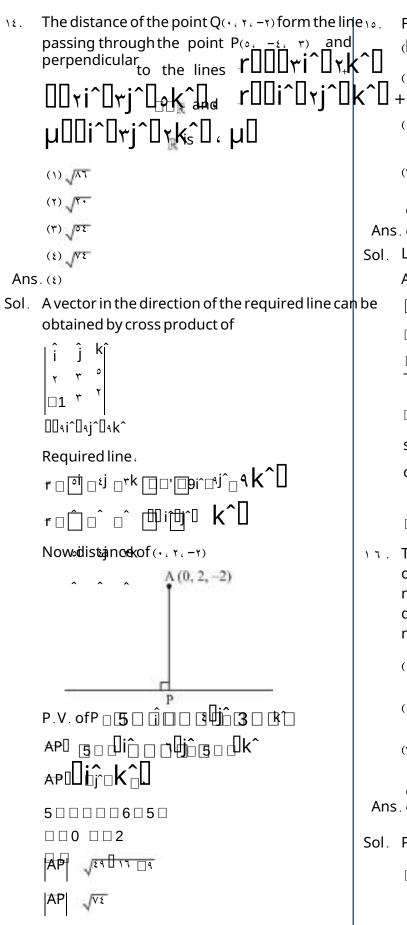
Sol. D □ r □ r □ r □ 1 □ $= 1(\boxed{\boxed{}} + 7) + 7(7\boxed{} - 9) + 1(-7 - 7\boxed{})$ $= \boxed{\boxed{} + \Psi + \underbrace{} \boxed{} - 1 \Lambda - Y - \Psi \boxed{}$ For infinite solutions $D = \cdot \cdot D = \cdot \cdot D = \cdot \cdot and$ D = • 14 4 4 4 4 $" \square \square$ $" \square$ $" \square$ putin(\) 10 0 0 <u>r</u> Now, 17 10 17 17 17 17 17 $= \xi + 0 \xi = 0 \Lambda$ The solution curve of the differential equation $y \frac{dx}{dy} []x [] logey [] \setminus [], x < \cdot, y < \cdot passin$ through the point (e, 1) is (1) $\log_{\chi} \frac{\chi}{\chi} \square_X$ (1) $\log_{\chi} \frac{\chi}{\chi} \square_{\chi}$

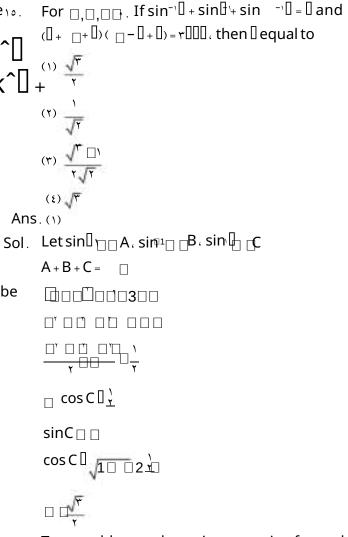
 $(r) \log_{e} | g \rangle | g \rangle$

Ans. (۳)







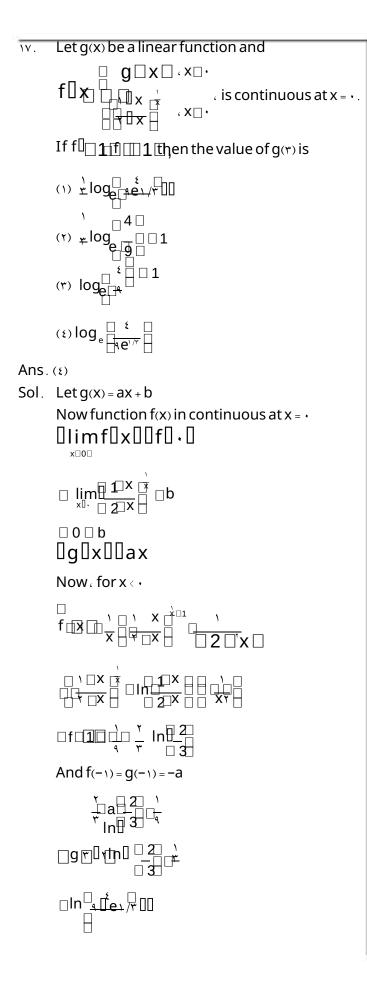


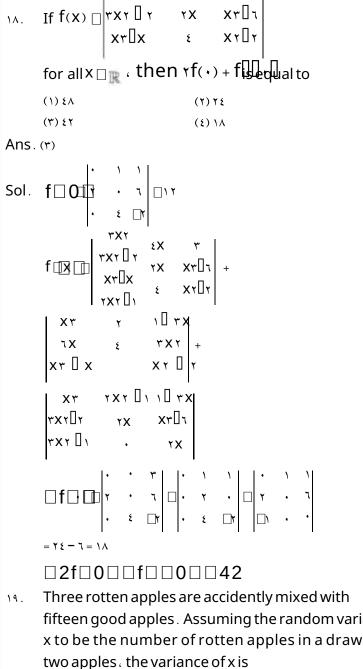
Two marbles are drawn in succession from a box containing \(\cdot\) red \(\tau\) white \(\cdot\) blue and \(\cdot\) orange marbles \(\cdot\) with replacement being made after each drawing . Then the probability \(\cdot\) that first drawn marble is red and second drawn marble is white \(\cdot\) is

- ر ہہ (۳)
- (٤) <u>(</u>5 (٤) (٤)

Sol. Probability of drawing first red and then white

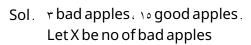
$$\Box \frac{\vee \circ}{\vee \circ} \Box \frac{\circ}{\vee \circ} \Box \frac{\circ}{\vee \circ}$$





fifteen good apples. Assuming the random variable x to be the number of rotten apples in a draw of two apples ι the variance of x is

Ans. (ξ)



Then
$$P(X = \cdot) \begin{bmatrix} & & & \\ & &$$

Var^[]X^[] **E** □**X 2** □**X** □ □

$$\Box \frac{\circ \vee}{\circ \circ} \Box \frac{1}{9} \Box \frac{\circ}{\circ \circ}$$

Let S be the set of positive integral values of a for

which

Then, the number of elements in S is:

- (1)1
- (٢) •
- (٣)
- (٤) ٣

Ans. (Y)

SECTION-B

If the integral

Ans. (177)

Put cosx = $t \sqcap \sin x dx = -\tau t dt$

Put
$$1 + t = k$$

otdt = Yk dk

$$I = \bigcap_{0}^{\Lambda} \bigcap_{i=1}^{q^{r}} k^{\gamma} \cap \forall k \in \mathbb{I} k \forall dk$$

$$I \underset{\circ}{\square} \frac{\sqrt{Y}}{\sqrt{Y}} \underset{\circ}{\square} \frac{\sqrt{Y}}{\sqrt{Y}} \underset{\tau}{\square} \frac{\sqrt{Y}}{\sqrt{Y}} \underset{\sigma}{\square} \frac{\sqrt{Y}}{\sqrt{$$

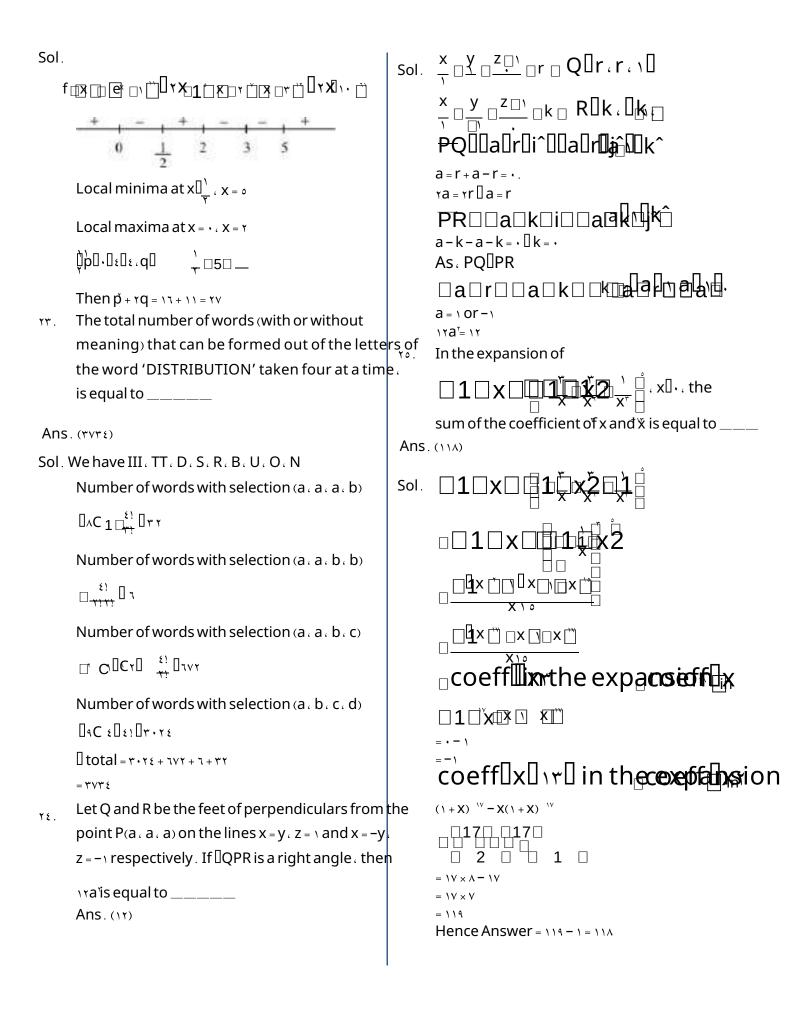
TY. Let
$$S = 1$$
 and $f: S = \mathbb{R}$ be defined as

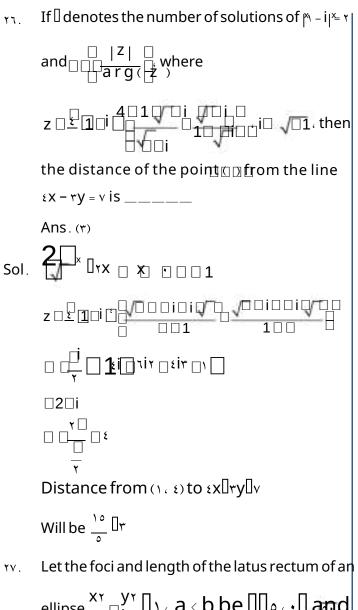
$$f(x) = \underbrace{\overset{x}{|e|}}_{t=1} \underbrace{(x+t)}_{t=1} \underbrace{(x+t)}$$

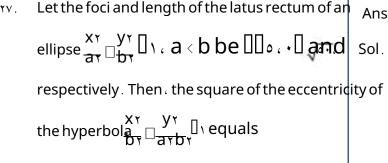
Let p = Sum of square of the values of x, where f(x) attains local maxima on S. and g = Sum of the values of x, where f(x) attains local minima on S.

Then, the value of p + rq is _____

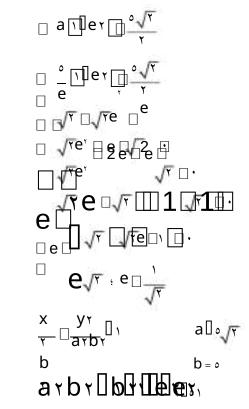
Ans. (YV)





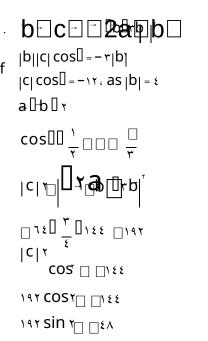


Ans. (۵۱)



Let a and b be two vectors such that $|a| \Box + |b| \Box \epsilon - and a \cdot b \Box \tau_{b} = b$ and the angle between b and c is \Box . then

Ans. (٤٨)



۲٩.	Let $A = \langle (1, 1), (1, 2) \rangle$ and $R = \langle (1, 1), (1, 2), (1, 2) \rangle$ be	۳٠.	Let f : □
	a relation on A . Let S be the equivalence		$f(x) \prod \underbrace{x \prod y}$ and
	relation on A such that R \square S and the number		_
	of		$M = \prod_{i=1}^{f(\sqrt{a})} x \sin^i x x x dx$
	elements in S is n . Then , the minimum value		f(a) f(\ _[a)
Ans	ု ရုf်ဂွာ is		$N = \prod_{f(a)}^{N \setminus g} sin \{ x \} \setminus x = dx : a = \frac{1}{x}$. If
Sol. /	All elements are included		$\square M \square \square N, \square, \square \square \square \cap \iota$ then the least value of
	Answer is 13		רַ ווֹץ (בּוֹיִי is equal to
		Ans	. (٥)
		Sol.	f(a) + f(y - a) = y.
			$\mathbf{M} = \prod_{f(\mathbf{a})}^{f(\mathbf{a})} \mathbf{a} \cdot \mathbf{a} \cdot \mathbf{a} \cdot \mathbf{b} \cdot \mathbf{a} \cdot \mathbf{a} \cdot \mathbf{a}$
			M = N - M $M = N$
			□ = Y : □ = Y : Ans. ∘
			, 1113.

PHYSICS

SECTION-A

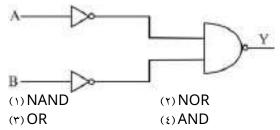
- The parameter that remains the same for molecules $is \cot A/r$. The angle of minimum deviation is: of all gases at a given temperature is :
 - (1) kinetic energy
- (Y) momentum
- (٣) mass
- (٤) speed

Ans.(1)

Sol.
$$KE = \int_{\mathbf{v}}^{\mathbf{f}} kT$$

Conceptual

Identify the logic operation performed by the given circuit.



Ans. (۳)

SOI YOABBADBDABB

(De-Morgan's law)

- The relation between time 't' and distance 'x' is t =٣٣. []x + []x, where [] and [] are constants. The relation, between acceleration (a) and velocity (v) is:
 - $(1) a = -Y \nabla^{-1}$
- (Y) a = 0 | V
- $\nabla \nabla = a \nabla \nabla$
- $(\xi) a = -\xi [v]$

Ans.(1)

$$\frac{\mathsf{dt}}{\mathsf{dx}} \square 2 \square \mathsf{x} \square \square$$

$$\frac{1}{V}\Box 2\Box x\Box\Box$$

(differentiating wrt time)

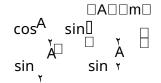
$$\Box \frac{1}{VY} \frac{dV}{dt} \Box 2 \Box \frac{dX}{dt}$$

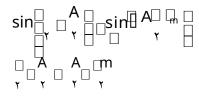
TEST PAPER WITH SOLUTION

The refractive index of a prism with apex angle A

Ans. (ξ)

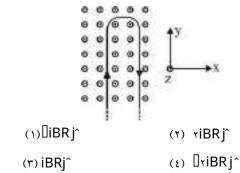
٣٤.



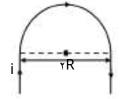


_m000 ∧A

A rigid wire consists of a semicircular portion of radius R and two straight sections. The wire is partially immerged in a perpendicular magnetic field B = B j as shown in figure. The magnetic force on the wire if it has a current i is:



Ans. (ξ)

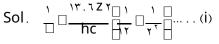


Note : Direction of magnetic field is $in^{k\hat{}}$

If the wavelength of the first member of Lyman

> series of hydrogen is \Box . The wavelength of the $_{\gamma_{\Lambda}}$ $\sec \frac{\partial n}{\partial x}$ member will be $\frac{\pi \tau}{x v}$

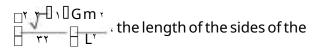
Ans.(1)



$$\frac{1}{100} \frac{1}{100} \frac{1}{100} \frac{1}{100} \frac{1}{100} \frac{1}{100} \frac{1}{100} \frac{1}{100} \frac{1}{100} \dots (ii)$$

On dividing (i) & (ii)

Four identical particles of mass m are kept at the four corners of a square . If the gravitational force exerted on one of the masses by the other masses is



square is

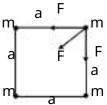
- (1) T
- (Y) £L

(m) mL

(٤) YL

Ans. (Y)

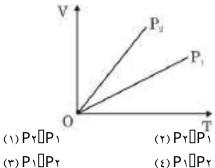
Sol.



$$F \square \frac{Gm^{\gamma}}{a\gamma}$$
 and $F' \square \frac{Gm^{\gamma}}{\sqrt{\gamma}a}$

$$\mathsf{F}_{\mathsf{net}} \, \square \, \sqrt{\mathsf{r}} \, \, \, \, \mathsf{ar} \, \, \, \, \square \, \, \mathsf{rar}$$

The given figure represents two isobaric processes for the same mass of an ideal gas, then



Ans. (ξ)

PV = nRT Sol.

Slope =
$$\frac{nR}{P}$$

(Slope) < (Slope)

If the percentage errors in measuring the length and the diameter of a wire are . . v/. each. The percentage error in measuring its resistance will be: (١) ٠. ٢٪.(٣) ٠. ١٪.

Ans.(Y)

Sol.
$$R \square \frac{\square}{\text{d} \Upsilon}$$

$$\frac{\Box L}{L}$$
 $\Box \cdot \cdot \cdot \cdot \%$ and $\frac{\Box d}{d}$ $\Box \cdot \cdot \cdot \cdot \%$

In a plane EM wave, the electric field oscillates sinusoidally at a frequency of o x Y. Hz and an amplitude of o. Vm: The total average energy density of the electromagnetic field of the wave is sol.

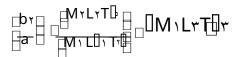
Ans.(1)

A force is represented by F = ax + btWhere x = distance and t = time. The dimensions of by /a are:

- شَيِّلِيَّةً MLT (يَوْلِيُّنِيُّ (٢)
- (٣) ٤٤٤ **MLT**

Ans.(1)

Sol. F = ax + bt

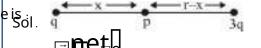


Two charges q and rq are separated by a distance 'r' in air. At a distance x from charge q, the resultant electric field is zero. The yalue of x is :

$$r = \frac{r}{r}$$

$$(r) \frac{r}{r(1 \square r)}$$

Ans. (۳)

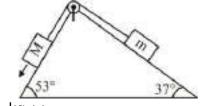


$$\frac{\mathsf{kq}}{\mathsf{x}^{\mathsf{y}}} \frac{\mathsf{k} \mathsf{l}^{\mathsf{y}} \mathsf{q}}{(\mathsf{r} \mathsf{l}^{\mathsf{y}} \mathsf{x})^{\mathsf{y}}}$$

$$(\mathbf{r} - \mathbf{X})^{\frac{1}{2}} \mathbf{r} \mathbf{X}$$

$$x \square r$$

In the given arrangement of a doubly inclined plane two blocks of masses M and m are placed. The blocks are connected by a light string passing over an ideal pulley as shown. The coefficient of friction between the surface of the plane and the blocks is The value of m, for which M = 1. kg will move down with an acceleration of mm/s. is : $(take g = 1 \cdot m / \$ and tan \pi v^o = \pi / \$)$



(1) 4 kg (Y)

٤.٥ kg (٣)

٦.٥ kg (٤)

7.70 kg

Ans. (۲)

Sol. $a = rm/s^{x}$

For M block

 $1 \cdot q \sin \alpha r^{\circ} - \mu (1 \cdot q) \cos \alpha r^{\circ} - T = 1 \cdot x r$

 $T = \Lambda \bullet - 10 - 7 \bullet$

T = £ 0 N

For m block

 $T - mg \sin \pi v^{\circ} - \mu mg \cos \pi v^{\circ} = m \times r$

m = ٤.0 kg A coil is placed perpendicular to a magnetic field of or T. When the field is ٤٤. changed to * · · · T in vs. an induced emf of vv V is produced in the coil. If the diameter of the coil is . . y m, then the number of turns in the coil is:

(1)V

(Y) V·

(٣) ٣٥

(٤) ١٤٠

Ans. (Y)



$\boxtimes\boxtimes(\boxtimes B)A$

 $B_i = 0 \cdot \cdot \cdot T_i$

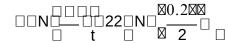
 $B_{c} = \forall \cdots T$

d = • . • y m

 $r = \cdot \cdot \cdot \cdot m$

$\boxtimes\boxtimes(\boxtimes B)A$

$$= (\Upsilon \cdot \cdot \cdot) \square (\cdot \cdot \cdot \cdot) = ^{\Upsilon} \cdot \cdot \Upsilon \square \square$$



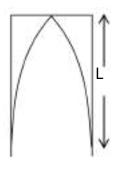
 $N = V \cdot$

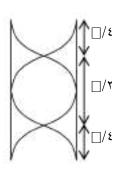
The fundamental frequency of a closed organ pipe is equal to the first overtone frequency of an open organ pipe. If length of the open pipe is $\tau \cdot cm$, the length of the closed pipe will be:

- (1) 7 · cm
- (Y) & cm
- (٣) ٣ · cm
- (ξ) \ ο CM

Ans. (ξ)

Sol.





 ${ }^\square_4 \, \square \, L_1$



 $V = f \square \square$

$$f2 \ \Box \frac{2v}{2L2} \ \Box$$

v = f1(4L1)

$$f2 \boxtimes \frac{v}{L2}$$

$$f_1 \square \frac{v}{4L1}$$

$$f_1 = f_2$$

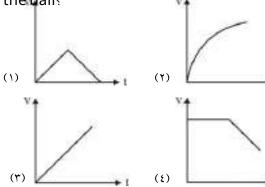
$$\frac{v}{4L1} \square \frac{v}{L2}$$

$$\boxtimes$$
 L2 = 4L1

$$60 = 4 \times L1$$

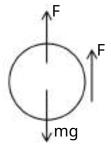
$$L_{1} = 15 \text{ cm}$$

A small steel ball is dropped into a long cylinder containing glycerine. Which one of the following is the correct representation of the velocity time graph for the transit of the balls

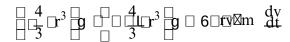


Ans.(Y)

Sol.



 $mg - F_R - F_{=}ma$



Let $rac{4}{3m}$ $\square R3g$ $\square \square L \!\!\! \square K_{_I}$ and $rac{6 \square L \!\!\! \square K_{_I}}{m}$ $\square K_{_2}$

$$\frac{dv}{dt} \, \Box K \mathbf{1} \boxtimes K 2v$$

$$\bigcap_{k \in \mathbb{N}} dv \atop 2 = \bigcap_{k \in \mathbb{N}} dt$$

 $\square_{\mathbb{K}_{2}}^{1} \ell n \boxtimes \mathbb{K}_{2} k 2 v \boxtimes \square t$

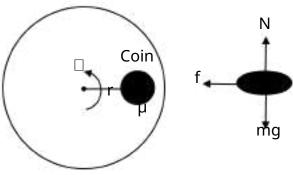
 $K1 \boxtimes K2 v \boxtimes K 1 e^{\boxtimes K \frac{t}{2}}$

$$v \boxtimes \frac{K_1 \boxtimes 1 \boxtimes e}{K_2} \boxtimes K_{2t}$$

friction between the coin and the disc is μ . If the distance of the coin from the center of the disc is r, the maximum angular velocity which can be given to the disc, so that the coin does not slip away, is

Ans. (۳)

Sol.



N = mg f = mUr

f = µN µmg = mr[]



Two conductors have the same resistances at °C but their temperature coefficients of resistance are and and are a coefficients. The respective temperature coefficients for their series and parallel combinations are :

$$(Y) \frac{\Box 1 \Box \Box_{Y}}{Y}, \frac{Y}{Y}$$

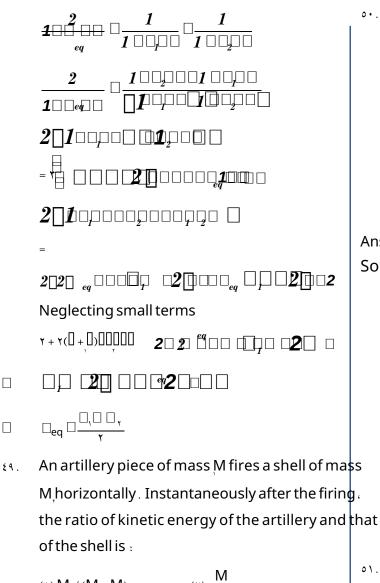
(٤)
$$\frac{\Box_{\scriptscriptstyle 1}\Box \Box_{\scriptscriptstyle 1}}{{}^{\scriptscriptstyle 1}}$$
 $\Box 1\Box \Box 2$ Ans. (1)

Sol. Series:

Parallel:

$$\frac{1}{R_{eq}} \square \frac{1}{R_1} \square \frac{1}{R_2}$$

$$\frac{R_{eq}}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square \frac{R_1}{R_1} \square \frac{R_1}{R_2} \square$$



(1) $M \neq M$ (2) $(M_{+}M) \neq M$ (1) $M \neq M$ (2) $M \neq M$ (3) $M \neq M$ (4) $M \neq M$

Ans. (Y)

Sol. $|p|^{\square}p|^{-}$ M $KE = \frac{p2}{2M} : p \text{ same}$ $KE = \frac{n}{M}$

 $\frac{\mathsf{KE}_1}{\mathsf{KE}_1} \square \frac{\mathsf{p}_1 \setminus \mathsf{tM}}{\mathsf{p}_1 \setminus \mathsf{tM}} \square \frac{\mathsf{M}_1}{\mathsf{M}_1}$

When a metal surface is illuminated by light of wavelength □, the stopping potential is AV.

When the same surface is illuminated by light of wavelength r□, stopping potential is rV.

The threshold wavelength for this surface is:

(1) □

(r) r□

(r) q□

Ans. (*) ٤.□□

SECTION-B

B is _____ T.

Ans.(\circ)
Sol. F[q[v]B] $5ek^{\cap}[e[3i^{\circ}]5j^{\circ}][[]]Bi^{\circ}[2Bj^{\circ}]$ $5ek^{\cap}[e[6Bk]]5Bk^{\circ}$ [B[5T]

A parallel plate capacitor with plate separation omm is charged up by a battery. It is found that on introducing a dielectric sheet of thickness r mm، while keeping the battery connections intact, the capacitor draws Yo'/ more charge from the battery than before The dielectric constant of the sheet is _

Ans.(Y)

Sol. Without dielectric

$$Q \square \frac{A \square}{d} \theta V$$

with dielectric

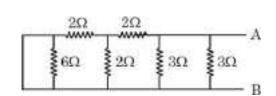
$$Q \, \Box \, \frac{A \, \Box \, V}{d \, \Box \, t \, \Box \, \overset{t}{K}}$$

given

$$\frac{A \, \square_{o}V}{d \, \square \, t \, \square_{K}^{t}} \, \square \, \square 1.25 \, \square_{d}^{A \, \square_{o}V}$$

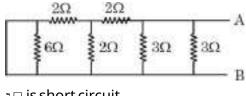
 $\sqcap K \sqcap 2$

Equivalent resistance of the following network is ٥٣. ╻┃.

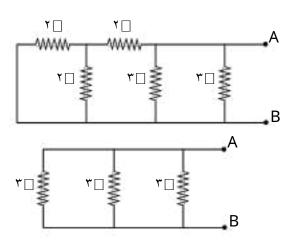


Ans. (1)

Sol.



ম⊓ is short circuit

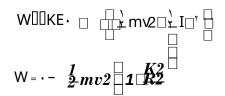


Req $\boxtimes 3 \boxtimes \frac{1}{3} \square 1 \square$

A solid circular disc of mass or kg rolls along a ٥٤. horizontal floor so that its center of mass has a speed of ... t m/s. The absolute value of work done

Ans. (٦)

Sol. Using work energy theorem



$$= \boxed{\frac{1}{2} \square 50 \square 0.4^2} \boxed{\frac{1}{2} \square \frac{1}{2}} = - 7$$

Absolute work = +1

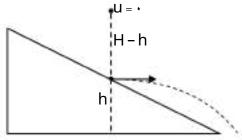
$$W = - \text{TJ} W \text{TT}$$

A body starts falling freely from height H hits an inclined plane in its path at height h. As a result of this perfectly elastic impact, the direction of the velocity of the body becomes horizontal. The value

of
$$\frac{H}{h}$$
 for which the body will take the maximum h time to reach the ground is _____.

Ans. (Y)

Sol.

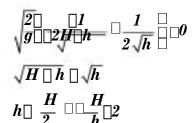


Total time of flight = T

$$T \Box \sqrt{\frac{rh}{g}} \Box \sqrt{\frac{r(H \ h)}{g}}$$

$$dT$$

For max. time =dh ...



Two waves of intensity ratio \ : 4 cross each other at a point. The resultant intensities at the point. when (a) Waves are incoherent is I(b) Waves are

coherent is Į and differ in phase by $\neg \cdot \circ$. $\frac{1}{1} \neg$

then x = ______

Ans. (14)

Sol. For incoherent wave I \ \[\lambda \lamb

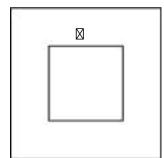
$$I = I + AI + 0 \quad 2\sqrt{9I_{\theta}^2} \cdot \frac{1}{2} = AVI$$

$$\frac{I_1}{I_7} \, \Box_{\prime \tau}^{\prime \tau}$$

ov. A small square loop of wire of side ℓ is placed inside a large square loop of wire of side L (L ¥). The loops are coplanar and their centers coinside. The value of the mutual inductance of the

Ans. (۱۲۸)

L



Flux linkage for inner loop.

Sol.

$$\Box_4 \Box \frac{\Box_L^i}{4\Box_-} \Box \sin 45 \operatorname{sin} 45 \Box^2$$

$$\Box \Box 2\sqrt{2} \frac{\Box_{\theta} i}{\Box L} \ell^2$$

$$M \; \Box_{\stackrel{\longleftarrow}{2}}^{\square} \; \Box^{\frac{2\sqrt{2}\square_{\theta}\ell^2}{\square L}} \; \Box 2\sqrt{2} \; \frac{\square_{\theta}}{\square}$$

$$\Box 2\sqrt{-\frac{4}{\Box}}\Box 10\Box 7$$

$$=82\square 10\square 7$$
 H

X = 11A

on. The depth below the surface of sea to which a rubber ball be taken so as to decrease its volume by ... v./. is ______ m.

(Take density of sea water $\stackrel{\cdot}{=} 1 \cdot \text{kgm}_{?}$ Bulk modulus of rubber = $9 \times 1^{\circ} \text{Nm}_{?}$ and $9 = 1 \cdot \text{ms}_{?}$

Ans. (۱۸)

$$\Box P \Box \Box V \over V$$

$$\Box$$
gh \Box \Box V

$$1 \cdot \frac{1}{2} \cdot 1 \cdot x \cdot h = \frac{9}{108} \frac{0.02}{100}$$

h = 1 A m

Ans. (v)

Sol.
$$v \square \square 4^{-2} \square x^2$$

 $2A$
at $x = 3$

New amplitude = A'

$$= \frac{3600 \square 10^7}{3600} kWh \square 1 \square 107kWh$$

CHEMISTRY

SECTION-A

- Give below are two statements:
 - Statement-I: Noble gases have very high boiling points.

Statement-II : Noble gases are monoatomic gases . They are held together by strong dispersion forces.

Because of this they are liquefied at very low

temperature. Hence, they have very high boiling points.

In the light of the above statements choose the

correct answer from the options given below:

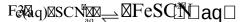
- () Statement I is false but Statement II is true.
- (y) Both Statement I and Statement II are true.
- (٣) Statement I is true but Statement II is false.
- (٤) Both Statement I and Statement II are false.

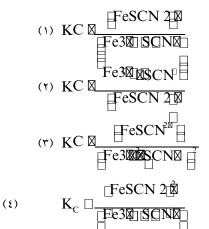
Ans. (ξ)

Sol. Statement I and II are False

Noble gases have low boiling points Noble gases are held together by weak dispersions. (Y) forces

For the given reaction, choose the correct ٦٢. expression of KC from the following:-





Ans.(1)

Sol. KC Products ion conc. Reactants ion conc.



TEST PAPER WITH SOLUTION

- Identify the mixture that shows positive deviations from Raoult's Law
 - (1) (CHr) (CO + C1H0NH)
 - (Y) CHClr + CIHI
 - (r) CHClr + (CHr)rCO
 - (٤) (CHr) (CO + CS)
- Sol. (CH+) CO + CS Exibits positive deviations from
 - The compound that is white in color is
 - (1) ammonium sulphide
 - (۲) lead sulphate
 - (٣) lead iodide

Raoult's Law

- (٤) ammonium arsinomolybdate
- Sol. Lead sulphate-white

Ammonium sulphide-soluble

Lead iodide-Bright yellow

Ammonium arsinomolybdate-yellow

- The metals that are employed in the battery industries are
 - A. Fe
 - B. Mn
 - C. Ni
 - D. Cr
 - E. Cd

Choose the correct answer from the options given below:

- (۱) B، C and E only
- (Y) A, B, C, D and E
- (۳) A، B، C and D only
- (٤) B، D and E only

Ans.(1)

Sol. Mn. Ni and Cd metals used in battery industries.

- A species having carbon with sextet of electrons 14. and can act as electrophile is called
 - (1) carbon free radical
 - (Y) carbanion
 - (٣) carbocation
 - (٤) pentavalent carbon

Ans. (۳)

Sol.



Six electron species

- Identify the factor from the following that does not affect electrolytic conductance of a solution.
 - (1) The nature of the electrolyte added.
 - (Y) The nature of the electrode used.
 - (٣) Concentration of the electrolyte.
 - (£) The nature of solvent used.

Ans.(Y)

- Sol. Conductivity of electrolytic cell is affected by concentration of electrolyte, nature of electrolyte and nature of solvent.
- The product (C) in the below mentioned reaction is:

 $\mathsf{CH}_3 \mathsf{\square} \mathsf{CH} \ \mathsf{\square} \mathsf{CH} \ \mathsf{\square} \mathsf{Br} \mathsf{\square} \mathsf{\square} \mathsf{\square} \mathsf{\square} \mathsf{CH} \mathsf{\square} \mathsf{Br} \mathsf{\square} \mathsf{\square} \mathsf{\square} \mathsf{C} \mathsf{C}$

- (1) Propan-1-ol
- (۲) Propene
- (۳) Propyne
- (٤) Propan-۲-ol

Ans. (ξ)

Sol.

$$CH3-CH2-Br\xrightarrow{KOH(a lc)}CH3-CH=CH2$$

$$\downarrow^{HBr}$$

$$CH3-CH-CH3\xrightarrow{\square}CH3-CH-CH3$$

$$OH$$

$$Rr$$

Given below are two statements: One is labelled as Assertion A and the other is labelled as Reason R:

Assertion A: Alcohols react both as nucleophiles and electrophiles.

Reason R: Alcohols react with active metals

such

as sodium, potassium and aluminum to yield ()) A is false but R is true corresponding alkoxides and liberate (r) A is true but R is false.

hydroff and Rare true and Ris the correct In the juntation of above statements, choose the

correct explanation of A

Ans. (ξ)

- Sol. As per NCERT, Assertion (A) and Reason (R) is correct but Reason (R) is not the correct explanation.
- The correct sequence of electron gain enthalpy of the elements listed below is
 - A Ar
 - B. Br
 - C.F
 - $\mathsf{D}.\mathsf{S}$

Choose the most appropriate from the options given below:

- $\text{(1)} \ C < B < D < A$
- (Y) A < D < B < C
- $\text{(\ref{eq:property})} \ A < D < C < B$
- $\textbf{($\xi$)}\ D < C < B < A$

Ans.(Y)

Sol. Element	[]egH(kJ/mol)
F	- ٣٣٣
S	-7
Br	-440
۸r	+ 4 ٦

- Identify correct statements from below:
 - A. The chromate ion is square planar.
 - B. Dichromates are generally prepared from chromates.
 - C. The green manganate ion is diamagnetic.
 - D. Dark green coloured KṛMnO ≀ disproportion ates in Pi X = Pi Pt + Pi a neutral or acidic medium to give permanganate
 - E. With increasing oxidation number of transition metal, ionic character of the oxides decreases.

Choose the correct answer from the options given below:

- (۱) B، C، D only
- (۲) A، D، Eonly
- (٣) A, B, Conly
- (٤) B, D, Eonly

Ans. (ξ)

- Sol. A. CrOsits tetrahedral
 - B. YNaYCrOs + YH+ NaYCrYOV + YNa ++HYO
 - per NCERT, green manganateis paramagnetic with \unpaired electron.
 - D. Statement is correct
 - E. Statement is correct
- 'Adsorption' principle is used for which of the following purification methods
 - (1) Extraction
 - (Y) Chromatography
 - (٣) Distillation
 - (٤) Sublimation

Ans.(Y)

Sol. Principle used in chromotography is adsorption

phase reaction is given by (where Pi is initial pressure and Pt is total pressure at time t)

(1) k
$$= \frac{2.303}{t} \log_{100} \frac{P_i}{1000}$$

(r)
$$k = \frac{2.303}{t} = \log \times 2Pi \times Pi$$
 2.303

$$\text{(r)} \quad k \, \Box \frac{t}{2.303} \, \Box \log \frac{\Box Pi \boxtimes Pt}{P_i} \, \boxtimes$$

$$\stackrel{\text{(1)}}{=} k \square \stackrel{t}{=} \stackrel{P}{\boxtimes} Pt \boxtimes Pt \boxtimes$$

Ans. (1)

$$Pt = Pi + x$$

- - = YPi Pt

$$K \boxtimes \frac{2.303}{t} log \frac{P}{2P! \boxtimes Pt}$$

Given below are two statements: One is labelled as ٧٤. Assertion A and the other is labelled as Reason R: Assertion A: pKa value of phenol is v... while that of ethanol is 10.4.

> Reason R: Ethanol is stronger acid than phenol. In the light of the above statements, choose the correct answer from the options given below:

- (1) A is true but R is false.
- (Y) A is false but R is true.
- (٣) Both A and R are true and R is the correct explanation of A.
- (£) Both A and R are true but R is NOT the correct explanation of A.

Ans.(1)

Sol. Phenol is more acidic than ethanol because conjugate base of phenoxide is more stable than ethoxide.

Given below are two statements:

Integrated rate law equation for a first order gas Statement I: IUPAC name of HO-CHt-(CHt)r-CHY-COCHY is v-hydroxyheptan-Y-one. Statement II: Y-oxoheptan-Y-ol is the correct IUPAC name for above compound.

> In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Statement Lis correct but Statement II is incorrect.
- (Y) Both Statement I and Statement II are incorrect.
- (٣) Both Statement I and Statement II are correct.
- (٤) Statement I is incorrect but Statement II is correct.

Ans. (1)

Sol. v-Hydroxyheptan-y-one is correct IUPAC name

- The correct statements from following are:
 - A. The strength of anionic ligands can be explained by crystal field theory.
 - B. Valvence bond theory does not quantitative interpretation of kinetic stability of coordination compounds.
 - C. The hybridization involved in formation of [Ni(CN)]2–complex is dsp2
 - D. The number of possible isomer(s) of cis-[PtC1₂(en)2† is one

Choose the correct answer from the options given

below: (1) A, D only (2) A, C only (3) B, D only

(4) B, C only

Ans. (ξ)

- Sol. B. VBT does not explain stability of complex
 - C. Hybridisation of [Ni(CN)]-2 is dsp2
- The linear combination of atomic orbitals to form molecular orbitals takes place only when the combining atomic orbitals
 - A. have the same energy
 - B. have the minimum overlap
 - C. have same symmetry about the molecular axis
 - D. have different symmetry about the molecular axis

Choose the most appropriate from the options given below:

- (1) A, B, C only
- (2) A and C only
- (3) B, C, D only
- (4) B and D only

Ans.(Y)

- Sol. * Molecular orbital should have maximum overlap
 - * Symmetry about the molecular axis should be similar

VA. Match List I with List II

LIST-I		LIST-II
A. Glucose/NaHCO3/D	I.	Gluconic acid
Glucose/HNO3	II.	No reaction
Elucose/HI/ D	III.	n-hexane
D. Glucose/Bromine water]	IV.	Saccharic acid

Choose the correct answer from the options given below:

- (1) A-IV, B-I, C-III, D-II
- (2) A-II, B-IV, C-III, D-I
- (3) A-III, B-II, C-I, D-IV
- (4) A-I, B-IV, C-III, D-II

Ans.(Y)

Sol. Glucose 3/4N3/4aH3D3/4CO3/4® no reaction

Glucose ³/₄H³/₄NOD³/₄3® saccharic acid

Glucose ¾H¾ID® n-hexane

Glucose ³/₄B³/₄r²D® Gluconic acid

- Consider the oxides of group 14 elements SiO2, GeO2, SnO2, PbO2, CO and GeO. The amphoteric oxides are
 - (1) GeO, GeO2
 - (2) SiO2, GeO2
 - (3) SnO2, PbO2
 - (4) SnO2, CO

Ans. (۳)

Sol. SnO2 and PbO2 are amphoteric

Match List I with List II

LIS	T I (Technique)	LIST II (Application)
A.	Distillation	I.Separation of glycerol from spent-lye
B.	Fractional	IIAniline - Water
	distillation	mixture
C.	Steam	I IS eparation of crude
	distillation	oil fractions
D.	Distillation	IK:hloroform-
	under reduced pressure	Aniline

Choose the correct answer from the options given below:

- (1) A-IV, B-I, C-II, D-III
- (2) A-IV, B-III, C-II. D-I
- (3) A-I. B-II, C-IV, D-III
- (4) A-II, B-III. C-I, D-IV

Ans. (Y)

Sol. Fact (NCERT)

SECTION-B

Molar mass of the salt from NaBr، NaNO۳، KI and CaFr which does not evolve coloured vapours on heating with concentrated HrSO٤ is ___________ g mo (Molar mass in g mờl : Na : ۲۴، N : ١٤، K : ٣٩،

Ο: ١٦، Br: Α٠, Ι: ١٢٧, F: ١٩, Ca: ٤٠

Ans. (VA)

Sol. CaFr does not evolve any gas with concentrated HrSO £.

NaBr 🛮 evolve Br

NaNO_T evolve NO_T

KI 🛮 evolve I 🕆

The 'Spin only' Magnetic moment for Ni(NTHr) 1 is _________ ^\x 1 · BM.

(given = Atomic number of Ni : YA)

Ans. (YA)

Sol. NH₇ act as WFL with Ni **

Ni^{*±} rd ^

1 1 1 1 1

No. of unpaired electron = Y

 $\square n \square n \square 2 \square n \square 2 \square 8 \square 2.82 \text{ BM}$

 $X = Y \Lambda$

Number of moles of methane required to produce $\forall \forall g \ CO \forall (g) \ after \ combustion \ is \ x \times \vec{\lor} \cdot moles$. The value of x is

Ans.(۵۰)

Sol. CH4(g)\(\mathbb{Q}\)2O2(g)\(\mathbb{Q}\)CO2(g)\(\mathbb{Q}\)2H2O()

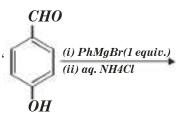
$$nCO \boxtimes \frac{22}{44} \square \cdot . \circ moles$$

So moles of CH ≀ required = •. ∘ moles

i.e. ٥٠×١٠ mole

X = 0 •

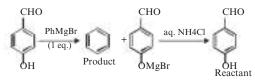
 $A \varepsilon$. The product of the following reaction is P.



The number of hydroxyl groups present in the product P is _____.

Ans. (•)

Sol. Product benzene has zero hydroxyl group



The number of species from the following in which the central atom uses sp hybrid orbitals in its bonding is______

NHT, SOT, SIOT, BECIT, COT, HTO, CHE, BFT

Ans. (ξ)

Sol. NH+ □sp *

SOT Spall

SiO 🛮 sp🗓

BeCl T D sp

COT [[]sp[]

H₇O∏sp ^r

CH: Dspl

BFr Osp

Ans. (1.)

Sol.
$$CH3CH2Br + NaOH$$
 $\xrightarrow{C2H3OH}$
 $CH3 = CH2 = CH2$
 $\xrightarrow{H2O}$
 $OH3 = CH2 = CH2$

Total number of hydrogen atom in A and B is \vee

Number of alkanes obtained on electrolysis of a mixture of CH+COONa and C+H+COONa is

۸ns. (۳)

Sol. CH3 COONa $\overset{\square}{C}H_3$ C2 H5 COONa $\overset{\square}{C}_2H_5$ 2C2H5\(\text{NC}\)H3\(\text{NC}\)H 2\(\text{NC}\)H3\(

Ans. (۱۲۳)

Sol.
$$\frac{3}{2}O_{2(g)} \rightleftharpoons O3(g.KP \boxtimes 2.47 \boxtimes 10 \boxtimes 29)$$

$$\Box G \Box = -RT \ln K \qquad P$$

$$= -\Lambda. \text{TIE} \times \text{IV} \quad \text{TYAN} \times \ln (\text{TIEV} \times \text{IV} -\text{TYA})$$

$$= -\Lambda. \text{TIE} \times \text{IV} \quad \text{TYAN} \times (-\text{TO.AV})$$

$$= \text{IT. IA} \text{ kJ}$$

The ionization energy of sodium in kJ mol⁻. If electromagnetic radiation of wavelength yet nm is just sufficient to ionize sodium atom is_____.

Ans. (ξ٩ξ)

Sol.
$$E \square \frac{1240}{\square \lceil nm \rceil} eV$$

$$\Box \frac{1240}{242} eV$$

$$= 0.17 \times 1.7 \times 1.$$
 -19

$$= 8.198 \times 10^{19} \text{ J/atom}$$

$$= 494 \text{ kJ/mol}$$

One Faraday of electricity liberates x × ہ gram atom of copper from copper sulphate ، x is_____

Ans.(0)

r Faraday 🗓 🗓 mol Cu

√ Faraday 🗓 🗓 🗓 mol Cu deposit

•. •
$$mol = •. • g atom = • × • • •$$

$$X = 0$$