

FINAL JEE–MAIN EXAMINATION – APRIL, 2024

(Held On Friday 05 April, 2024)

TIME : 3 : 00 PM to 6 : 00 PM

MATHEMATICS

SECTION-A

1. Let $f: [-1, 2] \rightarrow \mathbb{R}$ be given by $f(x) = 2x + [x] + \{x\}$, where $[t]$ denotes the greatest integer less than or equal to t . The number of points, where f is not continuous, is :

- (1) 6 (2) 3
(3) 4 (4) 5

Ans. (3)

Sol. Doubtful points : $-1, 0, 1, 2$

at $x = -1$:
 $f(x) = 2x + [x] + \{x\}$
 $f(-1) = -2 + [-1] + \{-1\} = -2 - 1 + 0 = -3$
 $f(-1^+) = 2(-1^+) + [-1^+] + \{-1^+\} = -2 - 1 + 0 = -3$
 $f(-1^-) = 2(-1^-) + [-1^-] + \{-1^-\} = -2 - 1 + 1 = -2$
 $f(-1^-) \neq f(-1)$ Discontinuous

at $x = 0$:

RHL $f(x) = 2x + [x] + \{x\}$
 $f(0) = 0 + [0] + \{0\} = 0$
 $f(0^+) = 2(0^+) + [0^+] + \{0^+\} = 0 + 0 + 0 = 0$
 $f(0^-) = 2(0^-) + [0^-] + \{0^-\} = 0 + 0 + 0 = 0$
 $f(0^-) = f(0) = f(0^+)$ Continuous

at $x = 1$:

LHL $f(x) = 2x + [x] + \{x\}$
 $f(1) = 2 + [1] + \{1\} = 2 + 1 + 0 = 3$
 $f(1^+) = 2(1^+) + [1^+] + \{1^+\} = 2 + 1 + 0 = 3$
 $f(1^-) = 2(1^-) + [1^-] + \{1^-\} = 2 + 0 + 1 = 3$
 $f(1^-) = f(1) = f(1^+)$ Continuous

at $x = 2$:

LHL $f(x) = 2x + [x] + \{x\}$
 $f(2) = 4 + [2] + \{2\} = 4 + 2 + 0 = 6$
 $f(2^+) = 2(2^+) + [2^+] + \{2^+\} = 4 + 2 + 0 = 6$
 $f(2^-) = 2(2^-) + [2^-] + \{2^-\} = 4 + 1 + 1 = 6$
 $f(2^-) = f(2) = f(2^+)$ Continuous

at $x = -1$:

LHL $f(x) = 2x + [x] + \{x\}$
 $f(-1) = -2 + [-1] + \{-1\} = -2 - 1 + 0 = -3$
 $f(-1^+) = 2(-1^+) + [-1^+] + \{-1^+\} = -2 - 1 + 0 = -3$
 $f(-1^-) = 2(-1^-) + [-1^-] + \{-1^-\} = -2 - 1 + 1 = -2$
 $f(-1^-) \neq f(-1)$ Discontinuous

RHL $f(x) = 2x + [x] + \{x\}$
 $f(1) = 2 + [1] + \{1\} = 2 + 1 + 0 = 3$
 $f(1^+) = 2(1^+) + [1^+] + \{1^+\} = 2 + 1 + 0 = 3$
 $f(1^-) = 2(1^-) + [1^-] + \{1^-\} = 2 + 0 + 1 = 3$
 $f(1^-) = f(1) = f(1^+)$ Continuous

2. The differential equation of the family of circles passing the origin and having center at the line $y = x$ is :

- (1) $(x^2 - y^2 + 2xy)dx = (x - y + 2xy)dy$
 (2) $(x^2 + y^2 + 2xy)dx = (x + y - 2xy)dy$
 (3) $(x^2 - y^2 + 2xy)dx = (x - y - 2xy)dy$
 (4) $(x^2 + y^2 - 2xy)dx = (x + y + 2xy)dy$

Ans. (3)

TEST PAPER WITH SOLUTION

Sol. $C: x^2 + y^2 + 2gx + 2fy = 0$ (1)

$2x + 2yy' + 2g + 2fy' = 0$

$g + yy' = -x - fy'$

Put in (1)

$x^2 + y^2 + 2x(-x - fy') + 2f(-x - fy') = 0$

$(x^2 - y^2 - 2x^2 - 2xyf') = x - y^2 + 2xyf'$

3. Let $S = \{z \in \mathbb{C} : |z| \leq 1\}$

$S_1 = \{z \in \mathbb{C} : \text{Im}(z) = 1\}$ and $S_2 = \{z \in \mathbb{C} : \text{Re}(z) = 1\}$

$S_3 = \{z \in \mathbb{C} : \text{Re}(z) = 0\}$ Then

(1) 120° (2) 120°
 (3) 120° (4) 120°

(3) 120° (4) 120°
 (3) 120° (4) 120°

Ans. (4)

Sol. $S_1: x^2 + y^2 = 1$ (1)

$S_2: \text{Im}(z) = 1$

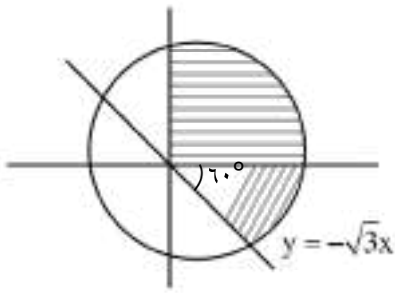
$\text{Im}(z) = 1$

$\text{Im}(z) = 1$

$x^2 + y^2 = 1$

$S_3: x = 1$ (2)

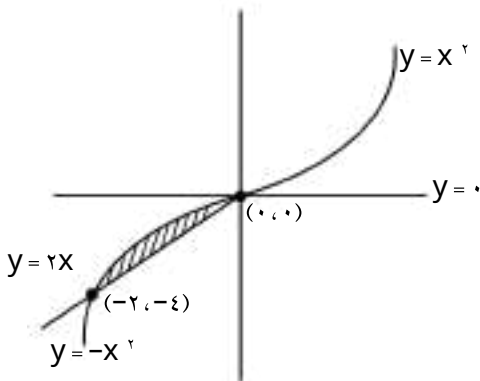
Area = $\frac{1}{2} \times \text{base} \times \text{height}$ (3)



8. The area enclosed between the curves $y = x|x|$ and $y = x - |x|$ is :

- (1) $\frac{1}{3}$ (2) $\frac{2}{3}$
 (3) 1 (4) $\frac{4}{3}$
 Ans. (4)

Sol.



$$A = \int_{-1}^0 (x - (-x)) dx = \int_{-1}^0 2x dx = 2 \left[\frac{x^2}{2} \right]_{-1}^0 = 1$$

9. words can be made using all the letters of the word BHBJO, with or without meaning. If these words are written as in a dictionary, then the word is :

- (1) OBBHJ (2) HBBJO
 (3) OBBJH (4) JBBOH
 Ans. (3)

Sol. B B H J O

B _____ 4! / 2! = 12

H _____ 3! / 2! = 3

J _____ 3! / 2! = 3

OBBHJ
 OBBJH is rankth

10. Let a, b, c be three vectors such that

$c \cdot i = a \cdot b \cdot i$ and $c \cdot j = a \cdot c \cdot i$ then $c \cdot j$ is equal to

- (1) 1 (2) 0
 (3) 1/2 (4) 1/3
 Ans. (2)

Sol. Let's assume $a = b = c = i$

$$i \cdot j = i \cdot k$$

and $c \cdot i = 0$

So,

$$i \cdot j = i \cdot k$$

$$i \cdot j = i \cdot k$$

$$i \cdot j = i \cdot k$$

$$i \cdot j = i \cdot k$$

$$i \cdot j = i \cdot k$$

$$i \cdot j = i \cdot k$$

$$i \cdot j = i \cdot k$$

$$i \cdot j = i \cdot k$$

$$c \cdot j = c \cdot k$$

$$c \cdot j = c \cdot k$$

11. Consider three vectors a, b, c . Let $a \cdot b = 1$ and $a \cdot c = 1$. If θ is the angle between the vectors b and c , then the minimum value of $|b - c|$ is equal to :

- (1) 1 (2) 1/2
 (3) 1/3 (4) 1/4
 Ans. (3)

Sol. $|c| \leq \frac{1}{\sqrt{a^2 + b^2}}$

$|c| \leq \frac{1}{\sqrt{a^2 + b^2}}$

$|a| \leq |b|$

$|a| \leq |b|$

$r \leq r \cos \theta$

$|c| \leq \frac{1}{\sqrt{a^2 + b^2}}$

$\frac{1}{\sqrt{a^2 + b^2}}$

$|c|_{\min} = \frac{1}{\sqrt{a^2 + b^2}}$

$\cos \theta = \frac{1}{\sqrt{a^2 + b^2}}$

$|c|_{\min} = \frac{1}{\sqrt{a^2 + b^2}} = \frac{1}{\sqrt{4 + 1}} = \frac{1}{\sqrt{5}}$

8.

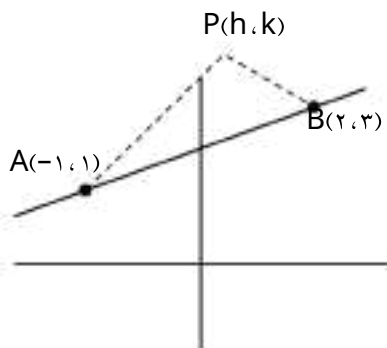
Let $A(-1, 1)$ and $B(2, 3)$ be two points and P be a variable point above the line AB such that the area of $\triangle PAB$ is 10. If the locus of P is $ax + by = 10$, then $5a + 2b$ is :

(1) $\frac{12}{5}$

(2) $\frac{7}{5}$

Ans. (1)

Sol.



$\frac{1}{2} \begin{vmatrix} h & k & 1 \\ -1 & 1 & 1 \\ 2 & 3 & 1 \end{vmatrix} = 10$

$-2x + 3y = 20$

$-2x + 3y = 20$

$a = -2, b = 3$

$5a + 2b = -10 + 6 = -4$

9.

Let (x, y, z) be the point (λ, μ, ν) in the line $\frac{x-1}{2} = \frac{y-1}{3} = \frac{z-2}{5}$. Then $\lambda + \mu + \nu$ is equal to

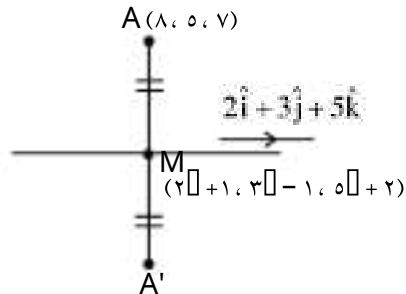
(1) 16 (2) 18

18 Ans.

(3) 20

(3)

Sol.



$AM = 2\hat{i} + 3\hat{j} + 5\hat{k}$

$(2-1)^2 + (3-0)^2 + (5-2)^2 = 10$

$3\lambda = 10$

$\lambda = \frac{10}{3}$

$\mu = \frac{10}{3}$

$\nu = \frac{10}{3}$

$A'(1, 0, 2)$

10.

If the constant term in the expansion of $(\frac{1}{x} + \frac{1}{x^2})^{10}$ is 2^x , then x is

$\frac{1}{x} + \frac{1}{x^2}$

equal to :

(1) 139

(2) 124

(3) 143

(4) 142

Ans. (3)

Sol.

$T_{r+1} = {}^{10}C_r \left(\frac{1}{x}\right)^{10-r} \left(\frac{1}{x^2}\right)^r$

$r = 10$

$T_{10} = {}^{10}C_{10} \left(\frac{1}{x}\right)^0 \left(\frac{1}{x^2}\right)^{10} = \frac{1}{x^{20}}$

$2^x = \frac{1}{x^{20}}$

11. Let $f, g: \mathbb{R} \rightarrow \mathbb{R}$ be defined as: $f(x) = |x - 1|$ and $g(x) = \begin{cases} e^x & x \leq 1 \\ x & x > 1 \end{cases}$. Then the function $f(g(x))$ is

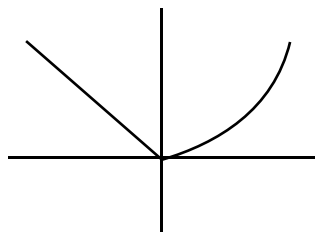
- (1) neither one-one nor onto.
- (2) one-one but not onto.
- (3) both one-one and onto.
- (4) onto but not one-one.

Ans. (1)

Sol. $f(g(x)) = |g(x) - 1|$

$$f(g(x)) = \begin{cases} |e^x - 1| & x \leq 1 \\ |x - 1| & x > 1 \end{cases}$$

$$f(g(x)) = \begin{cases} e^x - 1 & x \leq 1 \\ x - 1 & x > 1 \end{cases}$$



12. Let the circle $C_1: x^2 + y^2 - 2(x + y) + 1 = 0$ and C_2 be a circle having centre at $(-1, 0)$ and radius $\sqrt{2}$. If the line of the common chord of C_1 and C_2 intersects the y-axis at the point P, then the square of the distance of P from the centre of C_1 is:

- (1) 2
- (2) 1
- (3) 6
- (4) 8

Ans. (1)

Sol. $S_1: x^2 + y^2 - 2x - 2y + 1 = 0$

$$S_2: x^2 + y^2 + 2x - 2 = 0$$

$$\text{Common chord} = S_1 - S_2 =$$

$$-4x - 2y + 3 = 0$$

$$2x + y = \frac{3}{2} \Rightarrow P(0, \frac{3}{2})$$

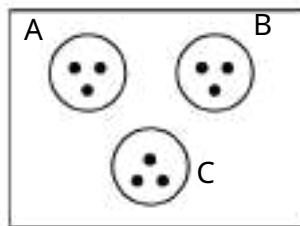
$$d(C_1, P) = (1 - 0)^2 + (2 - \frac{3}{2})^2 = 1$$

13. Let the set $S = \{2, 3, 4, 5, 6, \dots, 12\}$ be partitioned into 3 sets A, B, C with equal number of elements such that $A \cap B \cap C = S$ and $A \cap B = B \cap C = A \cap C = \emptyset$. The maximum number of such possible partitions of S is equal to:

- (1) 1680
- (2) 1020
- (3) 1710
- (4) 1640

Ans. (1)

Sol.



$$\frac{12!}{3!3!3!} = 1680$$

14. The values of m, n for which the system of equations

$$x + y + z = 4,$$

$$2x + 5y + 6z = 17,$$

$$x + 2y + mz = n$$

has infinitely many solutions, satisfy the equation:

- (1) $m^2 + n - 2m - n = 8$
- (2) $m^2 + n + 2m + n = 18$
- (3) $m^2 + n + mn = 18$
- (4) $m^2 + n - mn = 39$

Ans. (4)

Sol. $D = \begin{vmatrix} 1 & 1 & 1 \\ 2 & 5 & 6 \\ 1 & 2 & m \end{vmatrix} = 0 \Rightarrow m = 2$

$$D_3 = \begin{vmatrix} 1 & 1 & 4 \\ 2 & 5 & 17 \\ 1 & 2 & n \end{vmatrix} = 0 \Rightarrow n = 7$$

15. The coefficients a, b, c in the quadratic equation $ax^2 + bx + c = 0$ are from the set $\{1, 2, 3, 4, 5, 6\}$.

If the probability of this equation having one real root bigger than the other is p, then $100p$ equals:

- (1) 57
- (2) 38
- (3) 19
- (4) 76

Ans. (2)

Sol. $D < \cdot$

$$b' < \xi ac$$

$$b = 3 : (a, c) = (1, 1)(1, 2)(2, 1)$$

$$b = \xi : (a, c) = (1, 1)(1, 2)(2, 1)(1, 3)(3, 1)$$

$$b = 0 : (a, c) = (1, 1)(1, 2)(2, 1)(1, 3)(3, 1)(1, \xi)(\xi, 1) \\ (1, 0)(0, 1)(1, 6)(6, 1)(2, 3)(3, 2)(2, 2)$$

$$b = 6 : (a, c) = (1, 1)(1, 2)(2, 1)(1, 3)(3, 1)(1, \xi)(\xi, 1) \\ (1, 0)(0, 1)(1, 6)(6, 1)(2, 3)(3, 2)(2, \xi)(\xi, 2)(2, 2)$$

fav. cases = 38

Prob. $\frac{38}{1000}$

16. Let ABCD and AEFG be squares of side ξ and γ units, respectively. The point E is on the line segment AB and the point F is on the diagonal AC. Then the radius r of the circle passing through the point F and touching the line segments BC and CD satisfies :

$$(1) r = 1$$

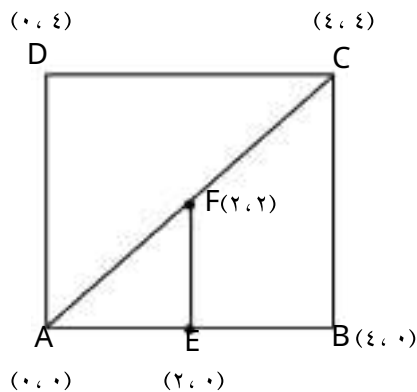
$$(2) r^2 - \lambda r + \lambda = 0$$

$$(3) 2r^2 - \xi r + 1 = 0$$

$$(4) 2r^2 - \lambda r + \gamma = 0$$

Ans. (2)

Sol.



$$OF = r^2$$

$$(\gamma - r)^2 + (\gamma - r)^2 = r^2$$

$$r^2 - \lambda r + \lambda = 0$$

17. Let $\int_0^1 (mx + n) e^{x^2} dx = a + b \cdot c$, then $\frac{1}{a+b+c}$ equals _____.

$$\int_0^1 (mx + n) e^{x^2} dx = a + b \cdot c$$

equals _____.

$$(1) \frac{1}{2}$$

$$(2) \frac{1}{3}$$

$$(3) \frac{1}{4}$$

$$(4) \frac{1}{5}$$

Ans. (4)

Sol. $\int_0^1 (mx + n) e^{x^2} dx = a + b \cdot c$

$$\int_0^1 (mx + n) e^{x^2} dx = a + b \cdot c$$

$$dx = \frac{1}{2} t^{-1/2} dt$$

$$\int_0^1 (mx + n) e^{x^2} dx = a + b \cdot c$$

$$\int_0^1 (mx + n) e^{x^2} dx = a + b \cdot c$$

$$a = \frac{1}{2}, b = \frac{1}{2}, c = 2$$

18.

Let $A = \begin{bmatrix} 1 & 2 & 3 \\ 2 & 3 & 4 \\ 3 & 4 & 5 \end{bmatrix}$ and $A^{-1} = \begin{bmatrix} a & b & c \\ d & e & f \\ g & h & i \end{bmatrix}$

If $B = \begin{bmatrix} 1 & 2 & 3 \\ 2 & 3 & 4 \\ 3 & 4 & 5 \end{bmatrix}$ is the matrix of cofactors

of the elements of A , then $\det(AB)$ is equal to :

$$(1) 343$$

$$(2) 120$$

$$(3) 74$$

$$(4) 216$$

Ans. (4)

Sol. Equating co-factor for A^{-1}

$$(a + b + c) = 1$$

$$a = 1, b = 2, c = 3$$

$$\text{Now, } a + b + c = 1$$

$$a = 1, b = 2, c = 3$$

$$|AB| = |A \text{ cof}(A)| = |A|^{-1}$$

$$A = \begin{bmatrix} 1 & 2 & 3 \\ 2 & 3 & 4 \\ 3 & 4 & 5 \end{bmatrix} \Rightarrow \det(A) = 1(15-12) - 2(10-12) + 3(10-9) = 3$$

19. If $y = \frac{\gamma \cos \alpha \cos \beta}{\cos \gamma + \cos \alpha \cos \beta}$,

then at $x = 1$, $y'' + y' + y$ is equal to:


$$\begin{array}{ccc} \square & & \\ (1) & - & (2) \end{array}$$

$$\frac{1}{(r)} = \quad (2)$$


Ans. (ε)

Sol. $y = \frac{r \cos \theta \cdot r \cos \theta}{r^2 \cos^2 \theta + r^2 \sin^2 \theta} = \frac{r^2 \cos^2 \theta}{r^2 (\cos^2 \theta + \sin^2 \theta)} = \frac{r^2 \cos^2 \theta}{r^2} = \cos^2 \theta$

$$y = \frac{1}{\cos x}$$


 $y = 1$

$$y' = \frac{1}{\sqrt{1 - \cos^2 x}}$$

 $y = \frac{1}{2}$

$$y'' = \frac{1}{r} \frac{d}{d\varphi} \left(\cos \varphi \frac{dr}{d\varphi} \right) + \sin \varphi \left(\frac{1}{r} \frac{dr}{d\varphi} \right)^2$$

$$\begin{array}{r} \square \\ \square \\ \square \\ \square \\ \hline y \end{array} \quad y =$$

For $x \in \mathbb{R}$, the least value of K , for which $\|f\|_X \leq \|f\|_Y$, is

$\frac{K}{\gamma}$, $\frac{K}{\gamma} + 1$ are three consecutive terms of an

A.P. is equal to :

- (1) 1, (2) 2
(3) 3, (4) 4

Ans. (1)

Sol. $k \frac{\partial^2 x}{\partial \xi^2} = \frac{\partial^2 x}{\partial \tau^2}$

k_{\square}

SECTION-B

۲۱. Let the mean and the standard deviation of the probability distribution

X	$\square\square$	1	.	-3
P(X)	$\frac{1}{3}\square$	K	$\frac{1}{6}$	$\frac{1}{2}$

be \square and \square , respectively. If $\square - \square = \square$, then $\square + \square$ is equal to _____.

Ans. (o)

Sol. $\frac{1}{x^2} \cdot \frac{1}{k} = \frac{1}{kx^2}$

$$\begin{array}{r} \square \square \square \\ - 2 \\ \hline \end{array}$$

$$\frac{1}{4} \left(\frac{1}{3} - \frac{1}{4} \right) = \frac{1}{12}$$

$$\begin{array}{r} \square \square \sqrt{\begin{array}{r} \gamma \square \gamma \\ \square \gamma \square \Sigma \end{array}} \end{array}$$

$$\square = \square + 2$$

$$\square 2 \square \square \square \square \overset{y}{2} \square \square \overset{y}{\square} \overset{y}{\square} \square \overset{y}{\square} \square \overset{q}{\square} \square \overset{q}{\square} 2 \square \overset{q}{\square} \square \square$$

$$\frac{\square^2}{9} - \frac{2\square}{3} \square$$

 $\alpha = .05$ (reject) or $\alpha = .1$

(.. x_{\square} is already given)

$$\square\square + \square = 2\square + 2$$

$$= 0$$

۲۲. Let $y = y(x)$ be the solution of the differential

$$\text{equation } \frac{dy}{dx} = \frac{y^2 - x^2}{1 + x^2}$$

Then the area enclosed by the curve

$f(x) = y(x)e^{x^2}$ and the line $y - x = \xi$ is _____.

Ans. (11)

Sol. IF = $e^{\int \frac{1}{x} dx} = e^{\ln x} = x$

$y' e^{\int \frac{1}{x} dx} = \frac{1}{x^2} e^{\int \frac{1}{x} dx}$

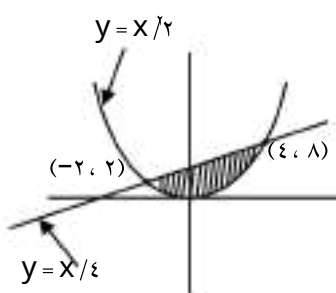
$y' e^{\ln x} = \frac{1}{x^2} e^{\ln x}$

$y' x = \frac{1}{x}$

$(0, 0) \rightarrow (1, 1)$

$y(x) = \frac{x}{2} e^{\frac{1}{x}}$

$f(x) = \frac{x}{2}$



$A = \int_0^1 (x e^{\frac{1}{x}} - \frac{x}{2}) dx = \frac{1}{2}$

23. The number of solutions of $\sin x + (2 + x \cos x) \sin x - 2(x - 1) = 0$, where $0 \leq x \leq \pi$, is

Ans. (2)

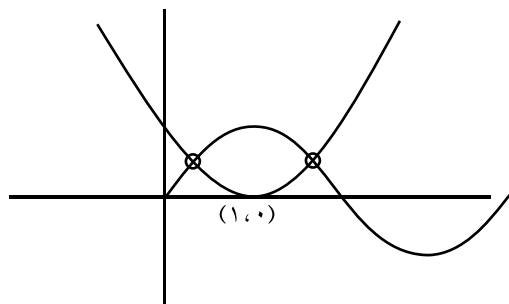
Sol. $\sin x - (x - 1) \sin x - 2(x - 1) = 0$

$\sin x - (x - 1) \sin x - 2(x - 1) = 0$

roots: -2 and $(x - 1)^2$

$\sin x = -2$ (reject) or $(x - 1)^2 = 0$

$\sin x = (x - 1)^2$



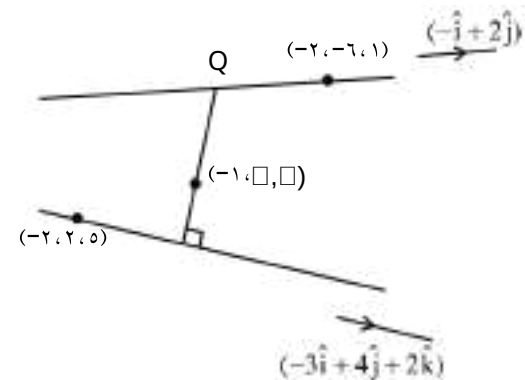
24.

Let the point $(-1, \alpha, \beta)$ lie on the line of the shortest distance between the lines $\frac{x-2}{3} = \frac{y-1}{\alpha} = \frac{z-0}{\beta}$ and $\frac{x-1}{1} = \frac{y-2}{\beta} = \frac{z-1}{\beta}$.

Then $(\alpha - \beta)$ is equal to _____.

Ans. (20)

Sol.



$P(-2, 2, 0)$

$Q(-2, -1, 1)$

DRS of $PQ = (x - 2, y - 2, z - 0)$

DRS of $PQ = \begin{vmatrix} i & j & k \\ -1 & 2 & 0 \\ -3 & \alpha & \beta \end{vmatrix}$

OR

$(2, 1, 1)$

$\frac{2}{\alpha} = \frac{1}{\beta} = \frac{1}{\beta}$

$\alpha = 2$ and $\beta = 1$

$\alpha - \beta = 1$

$Q(-2, -1, 1)$

$LPQ = \frac{x-2}{\alpha} = \frac{y-2}{\beta} = \frac{z-1}{\beta}$

$(-1, \alpha, \beta) = \frac{2-2}{\alpha} = \frac{2-2}{\beta} = \frac{2-1}{\beta}$

$\alpha = -2, \beta = 2$

$(\alpha - \beta) = 4$

20. If

$$1 + \frac{\sqrt{3}}{2\sqrt{2}} + \frac{0\sqrt{2}}{18} + \frac{9\sqrt{112}}{36\sqrt{2}} + \frac{89\sqrt{20}}{180} + \dots$$

upto $= \sqrt{\frac{b}{a}} \log \frac{a}{b}$, where a and b are

integers with $\gcd(a, b) = 1$, then $11a + 18b$ is equal to _____.

Ans. (76)

Sol. $S = 1 + \frac{x}{2\sqrt{2}} + \frac{x^2}{18} + \frac{x^3}{36\sqrt{2}} + \frac{x^4}{180} + \dots$

Put $\frac{x}{\sqrt{2}} = t$, where $x = \sqrt{2} - \sqrt{2}$

$$S = 1 + \frac{t}{2} + \frac{t^2}{18} + \frac{t^3}{36} + \frac{t^4}{180} + \dots$$

$$S = 1 + \frac{1}{2}t + \frac{1}{18}t^2 + \frac{1}{36}t^3 + \frac{1}{180}t^4 + \dots$$

$$S = 1 + \frac{t}{2} + \frac{t^2}{18} + \frac{t^3}{36} + \frac{t^4}{180} + \dots$$

$$S = 1 + \frac{t}{2} + \frac{t^2}{18} + \frac{t^3}{36} + \frac{t^4}{180} + \dots$$

$$S = 2 - \log \frac{1+t}{1-t} = 2 - \log \frac{1+t}{1-t}$$

$$S = 2 - \log \frac{1+\sqrt{2}-\sqrt{2}}{1-\sqrt{2}-\sqrt{2}} = 2 - \log \frac{1+\sqrt{2}-\sqrt{2}}{1-\sqrt{2}-\sqrt{2}}$$

$$S = 2 - \log \frac{1+\sqrt{2}-\sqrt{2}}{1-\sqrt{2}-\sqrt{2}} = 2 - \log \frac{1+\sqrt{2}-\sqrt{2}}{1-\sqrt{2}-\sqrt{2}}$$

$$S = 2 - \log \frac{1+\sqrt{2}-\sqrt{2}}{1-\sqrt{2}-\sqrt{2}} = 2 - \log \frac{1+\sqrt{2}-\sqrt{2}}{1-\sqrt{2}-\sqrt{2}}$$

$$a = 2, b = 2$$

$$11a + 18b = 11 \times 2 + 18 \times 2 = 76$$

26. Let a < b be a root of the equation $\sqrt{x} + x - 2 = 0$.

If $\lim_{x \rightarrow \frac{1}{a}} \frac{16 \cos(\sqrt{x} - \sqrt{2})}{(1 - ax)^2} = \frac{1}{17}$, where

$a, b \in \mathbb{Z}$ then $a + b$ is equal to _____.

Ans. (170)

Sol. $\sqrt{x} + x - 2 = 0$

$$\sqrt{x} + x - 2 = 0$$

$$\lim_{x \rightarrow \frac{1}{a}} \frac{16 \cos(\sqrt{x} - \sqrt{2})}{(1 - ax)^2} = \frac{1}{17}$$

$$\frac{16}{a^2} \cos(\sqrt{a} - \sqrt{2}) = \frac{1}{17}$$

$$\frac{16}{a^2} \cos(\sqrt{a} - \sqrt{2}) = \frac{1}{17}$$

$$\frac{16}{a^2} \cos(\sqrt{a} - \sqrt{2}) = \frac{1}{17}$$

$$\frac{16}{a^2} \cos(\sqrt{a} - \sqrt{2}) = \frac{1}{17}$$

$$a + b = 103 + 17 = 120$$

27. If $f(t) = \int_0^t \frac{x dx}{\cos t \sin x}$, $0 < t < \frac{\pi}{2}$, then the value

of $\frac{df(t)}{dt}$ equals _____.

Ans. (1)

Sol. $f(t) = \int_0^t \frac{x dx}{1 - \cos t \sin x} dx \dots (1)$

$$\frac{1}{2} \int \frac{(1-x)dx}{1-\cos^2 t \sin^2 x} \dots (2)$$

$$f(t) = \int \frac{dx}{1-\cos^2 t \sin^2 x}$$

$$f(t) = \int \frac{dx}{1-\cos^2 t \sin^2 x}$$

divide by $\cos x$

$$f(t) = \int \frac{\sec x dx}{1-\cos^2 t \tan^2 x}$$

$$f(t) = \int \frac{\sec x dx}{1-\cos^2 t \tan^2 x}$$

$$\tan x = z$$

$$\sec x dx = dz$$

$$f(t) = \int \frac{dz}{1-\sin^2 t z^2}$$

$$\frac{1}{2} \int \frac{dz}{\sin^2 t}$$

$$\text{Then } \int f(t) dt$$

$$\int \frac{1}{\sin^2 t} dt$$

$$= 1$$

28. Let the maximum and minimum values of

$$\sqrt{1-x^2} + \sqrt{1-y^2}, \quad x \in \mathbb{R} \text{ be } M \text{ and } m$$

respectively. Then $M - m$ is equal to _____.

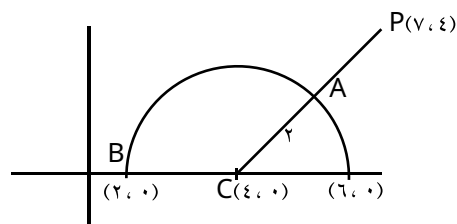
Ans. (1600)

$$\text{Sol. } (x-y)^2 + (y-x)^2 = 2$$

$$y = \sqrt{1-x^2}$$

$$y^2 = 1-x^2 \Rightarrow 1-x^2 = 1-y^2$$

$$x^2 - y^2 = 0 \Rightarrow y = x$$



$$m = \gamma$$

$$M = \xi$$

$$M - m = \xi - \gamma = 1 - \gamma$$

29. Let a line perpendicular to the line $\gamma x - y = 1$,

touch the parabola $y = \xi(x - \gamma)$ at the point P. The distance of the point P from the centre of the circle

$$x^2 + y^2 - \xi x - \gamma y + 1 = 0 \text{ is } \underline{\hspace{2cm}}.$$

Ans. (10)

$$\text{Sol. } y = \xi(x - \gamma)$$

$$\text{slope of tangent} = \frac{1}{\gamma}$$

$$\text{Point of contact } P = \left(\frac{1}{\gamma}, \frac{1}{\gamma} \right)$$

$$P(1/\gamma, -\xi)$$

$$\text{center of circle } C(\gamma, \xi)$$

$$\text{distance } CP = \sqrt{\left(\frac{1}{\gamma} - \gamma \right)^2 + (-\xi - \xi)^2} = 10$$

30. The number of real solutions of the equation

$$x|x+5| + 2|x+7| - 2 = 0 \text{ is } \underline{\hspace{2cm}}.$$

Ans. (3)

The number of real solutions of the equation

$$x|x+5| + 2|x+7| - 2 = 0 \text{ is } \underline{\hspace{2cm}}.$$

30.

Allen Ans. (3)

$$\text{Case I : } x \leq -5$$

Sol.

$$x^2 + 5x + 2x + 12 = 0$$

$$x^2 + 7x + 12 = 0$$

$$x = -3, -4$$

$$\text{Case II : } -5 < x < -3$$

$$-x^2 - 5x + 2x + 12 = 0$$

$$-X'' - 3X + 12 = 0$$

$$X = \frac{-3 \pm \sqrt{9 - 12}}{2}$$

$$= \frac{-3 \pm \sqrt{-3}}{2}$$

$$X = \frac{-3 \pm \sqrt{-3} - 3 \pm \sqrt{-3}}{2} \text{ (rejected)}$$

Case III : $x = -v$

$$-X'' + 3X - 12 = 0$$

$$X'' - 3X + 12 = 0$$

$$D = 9 - 12 < 0$$

No solutions

No. of solutions = 3

PHYSICS

SECTION-A

31. Given below are two statements :
- Statement I : When the white light passed through a prism, the red light bends lesser than yellow and violet.
- Statement II : The refractive indices are different for different wavelengths in dispersive medium. In the light of the above statements, choose the correct answer from the options given below :
- (1) Both Statement I and Statement II are true. 34.
- (2) Statement I is true but Statement II is false.
- (3) Both Statement I and Statement II are false.
- (4) Statement I is false but Statement II is true.

Ans. (1)

Sol. As $\mu_{\text{red}} < \mu_{\text{yellow}} < \mu_{\text{violet}}$
Light ray with longer wavelength bends less.

32. Which of the following statement is not true about stopping potential (V_0)?
- (1) It depends on the nature of emitter material.
- (2) It depends upon frequency of the incident light.
- (3) It increases with increase in intensity of the incident light.
- (4) It is $1/e$ times the maximum kinetic energy of electrons emitted. 35.

Ans. (3)

Sol. $K_{\text{Emax}} = h\nu - \phi_0 = eV_0$

33. The angular momentum of an electron in a hydrogen atom is proportional to : (Where r is the radius of orbit of electron)

(1) \sqrt{r}

(2) $\frac{1}{r}$

(3) r

(4) $\frac{1}{\sqrt{r}}$

Ans. (1)

TEST PAPER WITH SOLUTION

Sol. $FC = \frac{mv^2}{r}$

$\frac{Kqql}{r^2} = \frac{mv^2}{r}$

$mv^2 = Kqqlqr$

$L^2 = Kqqlr$

$L \propto r$

A galvanometer of resistance 100Ω when connected in series with 1000Ω measures a voltage of upto 10 V . The value of resistance required to convert the galvanometer into ammeter to read upto 10 A is $x \times 10^3 \Omega$. The value of x is :

(1) 2

(2) 100

(3) 20

(4) 200

Ans. (3)

Sol. $i_g = 400 \times 10^{-3} = 0.4 \text{ A}$

For ammeter

Let shunt resistance = S

$i_g R = (i - i_g) S$

$20 \times 10^{-3} \times 1000 = (10 - 20 \times 10^{-3}) S$

$S = 20 \times 10^3 \Omega$

The vehicles carrying inflammable fluids usually have metallic chains touching the ground :

- (1) To conduct excess charge due to air friction to ground and prevent sparking.
- (2) To alert other vehicles.
- (3) To protect tyres from catching dirt from ground.
- (4) It is a custom.

Ans. (1)

Sol. Static charge is developed due to air friction. This can result in combustion. So, metallic chains is used to discharge excess charge.

36. If n is the number density and d is the diameter of the molecule, then the average distance covered by a molecule between two successive collisions (i.e. mean free path) is represented by :

- (1) $\frac{1}{\sqrt{2}nd^2}$ (2) $\frac{1}{\sqrt{2}nd}$
 (3) $\frac{1}{\sqrt{2}nd^2}$ (4) $\frac{1}{\sqrt{2}nd}$

Ans. (3)

Sol. n = number of molecule per unit volume

d = diameter of the molecule

$$\lambda = \frac{1}{\sqrt{2}nd^2} \quad (\text{By Theory})$$

37. A particle moves in x-y plane under the influence of a force F such that its linear momentum is $\vec{P}(t) = P_0(\hat{i} \cos(kt) + \hat{j} \sin(kt))$. If k is constant, the angle between F and P will be :

- (1) $\frac{\pi}{2}$ (2) $\frac{\pi}{4}$
 (3) $\frac{\pi}{3}$ (4) $\frac{\pi}{6}$

Ans. (1)

Sol. $\vec{P} = P_0 \cos(kt)\hat{i} + P_0 \sin(kt)\hat{j}$; $|\vec{P}| = P_0$

$$\therefore \vec{P} = m\vec{v}$$

$$\hat{P} = \hat{v}$$

$$\hat{v} = \cos(kt)\hat{i} + \sin(kt)\hat{j}$$

$$\hat{a} = \frac{d\hat{v}}{dt} = -k \sin(kt)\hat{i} + k \cos(kt)\hat{j}$$

$$\hat{a} = k \sin(kt)\hat{i} - k \cos(kt)\hat{j}$$

$$\therefore \vec{F} = m\hat{a} = m k \sin(kt)\hat{i} - m k \cos(kt)\hat{j}$$

$$\cos \theta = \frac{\vec{F} \cdot \vec{P}}{FP} = \frac{m k \sin^2(kt) - m k \cos^2(kt)}{m k} = 0$$

38. The electrostatic force and magnetic force acting on a charge q moving with velocity \vec{v} (F)

can be written :

- (1) $F_1 = q\vec{v} \cdot \vec{E}$, $F_2 = q(\vec{B} \cdot \vec{v})$
 (2) $F_1 = q\vec{B}$, $F_2 = q(\vec{B} \times \vec{v})$
 (3) $F_1 = q\vec{E}$, $F_2 = q(\vec{v} \times \vec{B})$
 (4) $F_1 = q\vec{E}$, $F_2 = q(\vec{B} \times \vec{v})$

Ans. (3)

Sol. $F_1 = q\vec{E}$ (Theory)

$$F_2 = q(\vec{v} \times \vec{B})$$

39. A man carrying a monkey on his shoulder does cycling smoothly on a circular track of radius 4 m and completes 120 revolutions in 3 minutes. The magnitude of centripetal acceleration of monkey is (in m/s^2) :

- (1) $16\pi \text{ ms}^{-2}$
 (2) $4\pi \text{ ms}^{-2}$
 (3) $576\pi \text{ ms}^{-2}$

Ans. (2)

Sol. Given : $R = 4\text{ m}$.

120 revolution in 3 min

$$\omega = \frac{120 \text{ Rev}}{3 \text{ min}} = \frac{120 \times 2\pi \text{ rad}}{3 \times 60 \text{ sec}} = \frac{4\pi}{3} \text{ rad/s}$$

$$a_{\text{centripetal}} = \omega^2 R = \left(\frac{4\pi}{3}\right)^2 \times 4 = 16\pi \text{ m/s}^2$$

40. A series LCR circuit is subjected to an AC signal of 200 V , 50 Hz . If the voltage across the inductor ($L = 10\text{ mH}$) is 21.4 V , then the current in this circuit is _____ :

- (1) 18 A (2) 13 A
 (3) 10 A (4) 10 mA

Ans. (3)

Sol. Voltage across inductor $V_L = I X_L$

$$21.4 = I \times 2\pi f L$$

$$21.4 = I \times 2\pi \times 50 \times 10 \times 10^{-3}$$

$$21.4 = I \times 3.14 \times 10 \times 10^{-3} \times 1000$$

$$I = 10\text{ A}$$

ε 1. What is the dimensional formula of ab in the equation $\frac{1}{2}PV^2(V-b) = RT$, where letters have their usual meaning.

- (1) MLT (2) MLT

Ans. (3) MLT (4) MLT

Sol. $\therefore PV = b$
 Dimension of $b = \frac{[P][V]}{[L]}$

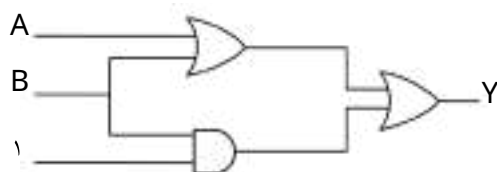
& $[P] = \frac{[a]}{[V^2]}$

$[a] = \frac{[P][V]}{[V^2]} = \frac{[MLT^{-2}][L]}{[L^2]}$

Dimension of $a = \frac{[MLT^{-2}][L]}{[L^2]}$

$ab = \frac{[MLT^{-2}][L]}{[L^3]}$

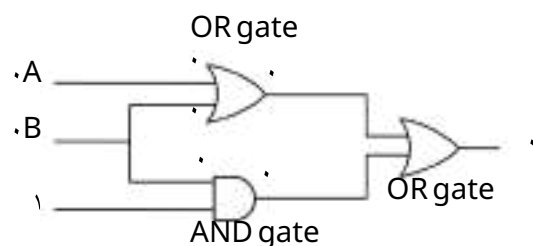
ε 2. The output (Y) of logic circuit given below is only when :



- (1) $A = 1, B = 0$ (2) $A = 0, B = 0$ (3) $A = 1, B = 1$ (4) $A = 0, B = 1$

Ans. (2)

Sol.



ε 3. A body is moving unidirectionally under the influence of a constant power source. Its displacement in time t is proportional to :

- (1) t^2 (2) $t^{3/2}$
 (3) $t^{2/3}$

Ans. (3)

Sol. $P = \text{constant} \Rightarrow FV = \text{constant}$

$m \frac{dV}{dt} V = \text{constant}$

$\int_0^V V dV = \int_0^t C dt$

$\frac{1}{2} V^2 = Ct$
 $V = \sqrt{2Ct}$

$\frac{d}{dt} t^{1/2}$

$\frac{d}{dt} K t^{1/2} = \frac{1}{2} K t^{-1/2}$

$S \propto K t^{3/2}$

$S \propto t^{3/2}$

ε 4. Displacement is proportional to $t^{3/2}$

Match List-I with List-II :-

	List-I EM-Wave		List-II Wavelength Range
(A)	Infra-red	(I)	$> 10 \text{ nm}$
(B)	Ultraviolet	(II)	$10 \text{ nm to } 1 \text{ mm}$
(C)	X-rays	(III)	$1 \text{ mm to } 10 \text{ nm}$
(D)	Gamma rays	(IV)	$1 \text{ nm to } 10 \text{ nm}$

Choose the correct answer from the options given below :

- (1) (A)-(II), (B)-(I), (C)-(IV), (D)-(III)
 (2) (A)-(III), (B)-(II), (C)-(IV), (D)-(I)
 (3) (A)-(IV), (B)-(III), (C)-(II), (D)-(I)
 (4) (A)-(I), (B)-(III), (C)-(II), (D)-(IV)

Ans. (2)

Sol. Infrared is the least energetic thus having biggest wavelength (I) & gamma rays are most energetic thus having smallest wavelength (IV).

ε 5. During an adiabatic process, if the pressure of a gas is found to be proportional to the cube of its absolute temperature, then the ratio of the

gas is :

- (1) $\frac{5}{3}$ (2) $\frac{9}{7}$
(3) $\frac{3}{2}$ (4) $\frac{7}{5}$

Ans. (3)

Sol. $P \propto T^\gamma$

$$PT^{-\gamma} = \text{constant}$$

$$\therefore \frac{PV}{T} \propto nR = \text{constant from ideal gas equation}$$

$$(P)(PV) \propto \text{constant}$$

$$P^2V = \text{constant} \dots (1)$$

\therefore Process equation for adiabatic process is

$$PV^\gamma = \text{constant} \dots (2)$$

Comparing equation (1) and (2)

$$\frac{C_1}{C_2} \propto \gamma \propto \frac{3}{2}$$

ε 6.

Match List-I with List-II :

	List-I		List-II
(A)	A force that restores elastic body unit area to its original state	(I)	Bulk modulus
(B)	Two equal and opposite forces parallel to opposite faces	(II)	Young's modulus
(C)	Forces perpendicular everywhere over the surface unit area same everywhere	(III)	Stress
(D)	Two equal and opposite forces perpendicular to opposite faces	(IV)	Shear modulus

Choose the correct answer from the options given below : (1) (A)-(II), (B)-(IV), (C)-(I), (D)-(III) (2) (A)-(IV), (B)-(II), (C)-(III), (D)-(I) (3) (A)-(III), (B)-(IV), (C)-(I), (D)-(II) (4) (A)-(III), (B)-(I), (C)-(II), (D)-(IV)

Ans. (3)

Sol. (A) stress $\propto \frac{F_{\text{restoring}}}{A}$

$$\text{Stress} = \frac{F_{\text{restoring}}}{A}$$

(A)-(III)

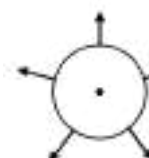
(B)



□ Shear stress

(B)-(IV)

(C)



□ Volumetric stress

(C)-(I)

(D)



□ Longitudinal stress

(D)-(II)

ε 7.

A vernier callipers has 20 divisions on the vernier scale, which coincides with 19 divisions on the main scale. The least count of the instrument is 0.1 mm. One main scale division is equal to _____ mm.

(1) 1

(2) 0.5

(3) 2

(4) 0

Ans. (3)

Sol. 20 VSD = 19 MSD

$$1 \text{ VSD} \propto \frac{19}{20} \text{ MSD}$$

$$\text{L.C.} = 1 \text{ MSD} - 1 \text{ VSD}$$

$$\therefore 0.1 \text{ mm} = 1 \text{ MSD} \propto \frac{19}{20} \text{ MSD}$$

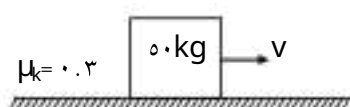
$$0.1 \propto \frac{1}{20} \text{ MSD}$$

$$1 \text{ MSD} = 2 \text{ mm}$$

8. A heavy box of mass 50 kg is moving on a horizontal surface. If co-efficient of kinetic friction between the box and horizontal surface is 0.3 then force of kinetic friction is :
 (1) 14.7 N (2) 147 N (3) 1.47 N (4) 1470 N

Ans. (2)

Sol.



$$F_k = \mu_k N = 0.3 \times 50 \times 9.8 = 147 \text{ N}$$

9. A satellite revolving around a planet in stationary orbit has time period 1 hours. The mass of planet is one-fourth the mass of earth. The radius orbit of planet is : (Given = Radius of geo-stationary orbit for earth is $4.2 \times 10^4 \text{ km}$)
 (1) $1.4 \times 10^4 \text{ km}$
 (2) $8.4 \times 10^4 \text{ km}$
 (3) $1.68 \times 10^4 \text{ km}$
 (4) $1.05 \times 10^4 \text{ km}$

Ans. (4)

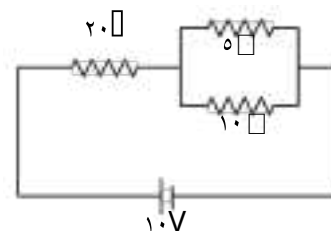
Sol. $T \propto \frac{2\pi R^3}{\sqrt{GM}}$

$$\frac{T_1}{T_2} = \left(\frac{R_1}{R_2} \right)^{3/2} \sqrt{\frac{M_2}{M_1}}$$

$$\frac{6}{24} = \left(\frac{r_1}{4.2 \times 10^4} \right)^{3/2} \sqrt{\frac{M}{M/4}}$$

$$r_1 = 1.05 \times 10^4 \text{ km}$$

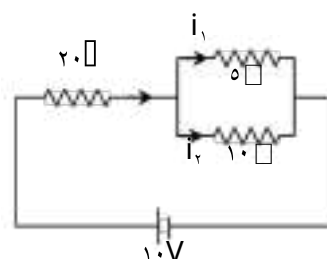
The ratio of heat dissipated per second through the resistance 20 Ω and 10 Ω in the circuit given below is :



- (1) 1 : 2 (2) 2 : 1
 (3) 4 : 1 (4) 1 : 1

Ans. (2)

Sol.



$$\frac{i_1}{i_2} = \frac{10}{5} = \frac{2}{1}$$

$$\frac{P_1}{P_2} = \frac{i_1^2 R_1}{i_2^2 R_2} = \frac{2^2 \times 20}{1^2 \times 10} = \frac{5}{1}$$

SECTION-B

10. A solenoid of length 1.0 m has a radius of 1 cm and is made up of 'm' number of turns. It carries a current of 5A. If the magnitude of the magnetic field inside the solenoid is $1.28 \times 10^{-2} \text{ T}$, then the value of m is :

Ans. (500)

Sol. $\mu_0 n i = B$ n = number of turns per unit length

$$\mu_0 \frac{m}{l} i = B$$

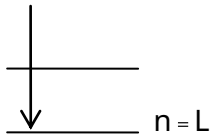
$$m = \frac{B \cdot l}{\mu_0 i} = \frac{1.28 \times 10^{-2} \times 1}{12.56 \times 10^{-7} \times 5}$$

$$m = 500$$

Q2. The shortest wavelength of the spectral lines in the Lyman series of hydrogen spectrum is 915 \AA . The longest wavelength of spectral lines in the Balmer series will be _____ \AA .

Ans. (6588)

Sol. Lyman Series



$$\text{Shortest } \frac{hc}{\lambda} = 3.6 \times \frac{1}{n^2} - \frac{1}{1^2}$$

$$\lambda = \frac{hc}{3.6} = 915 \text{ \AA}$$

Balmer Series :

$$n = 2$$

$$n = 2$$

$$\frac{hc}{\lambda} = 3.6 \times \frac{1}{2^2} - \frac{1}{2^2}$$

$$\frac{hc}{\lambda} = 3.6 \times \frac{1}{4} - \frac{1}{4}$$

$$\frac{hc}{\lambda} = 3.6 \times \frac{1}{4} - \frac{1}{4}$$

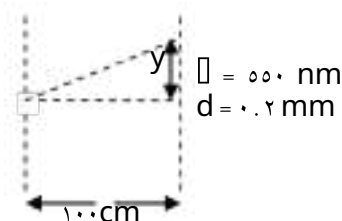
$$\lambda = \frac{hc}{3.6 \times \frac{1}{4} - \frac{1}{4}} = 6588 \text{ \AA}$$

Q3.

In a single slit experiment, a parallel beam of green light of wavelength 500 nm passes through a slit of width 0.2 mm . The transmitted light is collected on a screen 100 cm away. The distance of first order minima from the central maximum will be $x \text{ m}$. The value of x is :

Ans. (270)

Sol.

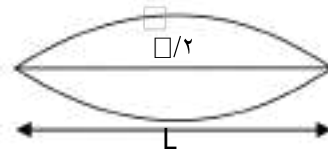


$$y = \frac{\lambda D}{d} = \frac{500 \times 10^{-9} \times 100 \times 10^{-2}}{0.2 \times 10^{-3}} = 270$$

Q4. A sonometer wire of resonating length 40 cm has a fundamental frequency of 400 Hz when kept under some tension. The resonating length of the wire with fundamental frequency of 600 Hz under same tension

Ans. (60) _____ cm .

Sol.



$$f = 400 \text{ Hz} ; v = \sqrt{\frac{T}{\mu}} = \text{constant}$$

$$2L = v = f \cdot \lambda$$

$$\frac{v}{2f_0} = L ; v = 2Lf_0$$

$$L' = \frac{v}{2f} = \frac{2Lf_0}{2f}$$

$$L' = \frac{L f_0}{f} = \frac{40 \times 400}{600} = 60$$

Q5. A hollow sphere is rolling on a plane surface about its axis of symmetry. The ratio of rotational kinetic

energy to its total kinetic energy is $\frac{x}{5}$. The value of x is _____.

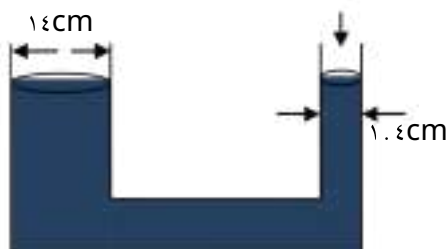
Ans. (2)

$$\text{Sol. } \frac{\frac{1}{2} I \omega^2}{\frac{1}{2} I \omega^2 + \frac{1}{2} m v^2} = \frac{\frac{1}{2} \times \frac{2}{3} m R^2 \omega^2}{\frac{1}{2} \times \frac{2}{3} m R^2 \omega^2 + \frac{1}{2} m (R \omega)^2}$$

$$= \frac{\frac{2}{3}}{\frac{2}{3} + 1} = \frac{2}{5}$$

$$x = 2$$

56. A hydraulic press containing water has two arms with diameters as mentioned in the figure. A force of 10 N is applied on the surface of water in the thinner arm. The force required to be applied on the surface of water in the thicker arm to maintain equilibrium of water is $\underline{100 \text{ N}}$ N .



Ans. (100 N)

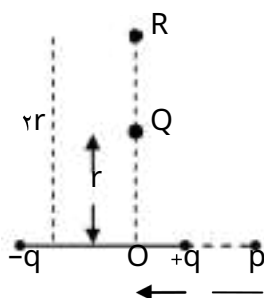
Sol. $\frac{F_1}{A_1} = \frac{F_2}{A_2}$

$$\frac{F}{(1)^2} = \frac{10}{(1.5)^2}$$

$$F = 100 \text{ N}$$

57. The electric field at point p due to an electric dipole is E . The electric field at point R on

equatorial line will be $\frac{E}{x}$. The value of x :



Ans. (16)

Sol. $E_P = \frac{2KP}{r^3} = E$

$$E_R = \frac{KP}{(2r)^3} = \frac{E}{16}$$

$$x = 16$$

58. The maximum height reached by a projectile is 16 m . If the initial velocity is halved, the new maximum height of the projectile is $\underline{4 \text{ m}}$.
Ans. (4)

Sol. $H_{\text{max}} = \frac{u^2 \sin^2 \theta}{2g}$

$$\frac{H_{1\text{max}}}{H_{2\text{max}}} = \frac{u_1^2}{u_2^2}$$

$$\frac{64}{H_{2\text{max}}} = \frac{u^2}{(u/2)^2}$$

59. $H_{2\text{max}} = 4 \text{ m}$

A wire of resistance 20Ω is divided into 10 equal parts. A combination of two parts are connected in parallel and so on. Now resulting pairs of parallel combination are connected in series. The equivalent resistance of final combination is $\underline{2 \Omega}$.

Ans. (2)

Sol.

$$20 \Omega \rightarrow 10 \text{ equal part}$$

Each part has resistance = 2Ω

2 parts are connected in parallel so, $R = 1 \Omega$

Now, there will be 5 parts each of resistance 1Ω , they are connected in series.

$$R_{\text{eq}} = 5R, R_{\text{eq}} = 5 \Omega$$

60. The current in an inductor is given by $I = (rt + \lambda)$ where t is in second. The magnitude of induced emf produced in the inductor is 12 mV . The self-inductance of the inductor $\underline{1 \text{ mH}}$.

Ans. (1)

Sol. $I = rt + \lambda$

$$\epsilon = 12 \text{ mV}$$

$$|\epsilon| = \left| \frac{dI}{dt} \right|$$

$$12 = L \times r$$

$$L = 1 \text{ mH}$$

CHEMISTRY

SECTION-A

61. Match List - I with List - II.

List - I

List - II

(A) ICl

(I) T-Shape

(B) ICl₃

(II) Square pyramidal

(C) ClF₅

(III) Pentagonal bipyramidal

(D) IF₇

(IV) Linear

Choose the correct answer from the options given below:

(1) (A)–(I), (B)–(IV), C–(III), D–(II)

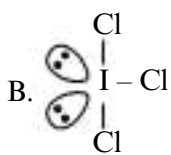
(2) (A)–(I), (B)–(III), C–(II), D–(IV)

(3) (A)–(IV), (B)–(I), C–(II), D–(III)

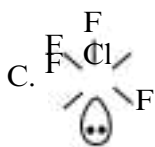
(4) (A)–(IV), (B)–(III), C–(II), D–(I)

Ans. (3)

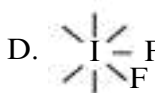
Sol. A. I – Cl (iv) linear



(I) T-shape



(II) Square pyramidal



(III) Pentagonal bipyramidal

62. While preparing crystals of Mohr's salt, dil. H₂SO₄ is added to a mixture of ferrous sulphate and ammonium sulphate, before dissolving this mixture in water, dil. H₂SO₄ is added here to:

(1) prevent the hydrolysis of ferrous sulphate

(2) prevent the hydrolysis of ammonium sulphate

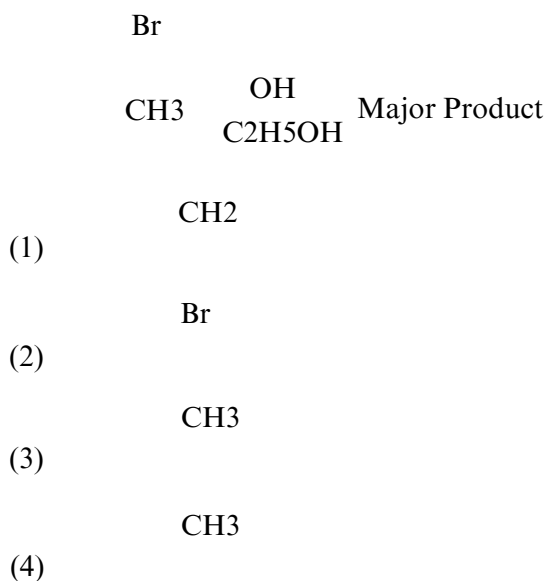
(3) make the medium strongly acidic

(4) increase the rate of formation of crystals

Ans. (1)

TEST PAPER WITH SOLUTION

Sol. Fe⁺² ions undergoes hydrolysis, therefore while preparing aqueous solution of ferrous sulphate and ammonium sulphate in water dilute sulphuric acid is added to prevent hydrolysis of ferrous sulphate. Identify the major product in the following reaction.

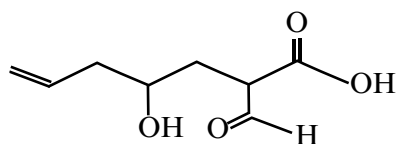


Ans. (3)

Sol.



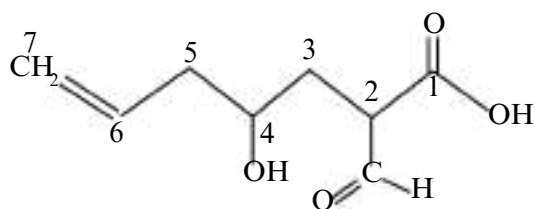
64. The correct nomenclature for the following compound is:



(1) 2-carboxy-4-hydroxyhept-6-enal (2) 2-carboxy-4-hydroxyhept-7-enal (3) 2-formyl-4-hydroxyhept-6-enoic acid (4) 2-formyl-4-hydroxyhept-7-enoic acid

Ans. (3)

Sol.



2-formyl-4-hydroxyhept-6-enoic acid Given

65. below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A) : NH_3 and NF_3 molecule have pyramidal shape with a lone pair of electrons on nitrogen atom. The resultant dipole moment of NH_3 is greater than that of NF_3 .

Reason (R) : In NH_3 , the orbital dipole due to lone pair is in the same direction as the resultant moment of the N–H bonds. F is the most electronegative element.

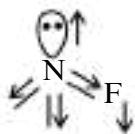
In the light of the above statements, choose the correct answer from the options given below:

- (1) Both (A) and (R) are true and (R) is the correct explanation of (A)
 (2) (A) is false but (R) is true
 (3) (A) is true but (R) is false

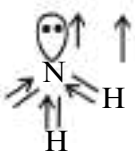
(4) Both (A) and (R) are true but (R) is NOT the correct explanation of (A)

Ans. (1)

Sol.



Resultant dipole moment = $0.80 \times 10^{-30} \text{Cm}$



Resultant dipole moment = $4.90 \times 10^{-30} \text{cm}$

66. Given below are two statements: Statement I : On passing HCl(g) through a saturated solution of BaCl_2 , at room temperature white turbidity appears.

Statement II : When HCl gas is passed through a saturated solution of NaCl , sodium chloride is precipitated due to common ion effect. In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Statement I is correct but Statement II is incorrect
 (2) Both Statement I and Statement II are incorrect
 (3) Statement I is incorrect but Statement II is correct
 (4) Both Statement I and Statement II are correct

Ans. (1)

Sol. BaCl_2 , NaCl are soluble but on adding HCl(g) to BaCl_2 , NaCl solutions, Sodium or Barium chlorides may precipitate out, as a consequence of the law of mass action.

67. The metal atom present in the complex MABXL (where A, B, X and L are unidentate ligands and M is metal) involves sp^3 hybridization. The number of geometrical isomers exhibited by the complex is:

- (1) 4 (2) 0
 (3) 2 (4) 3

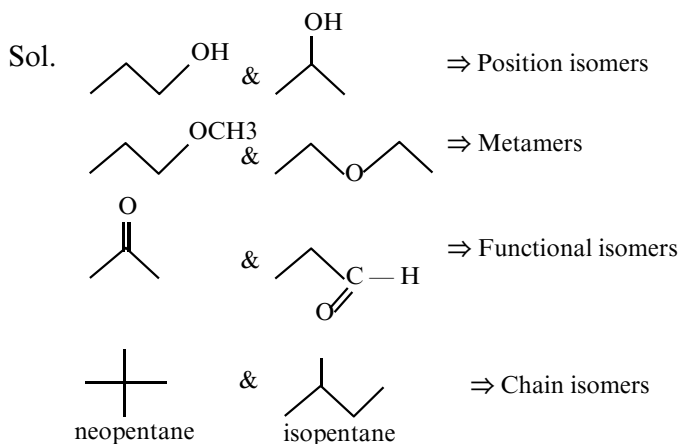
Ans. (2)

Sol. Tetrahedral complex does not show geometrical isomerism.

68. Match List - I with List - II.

List - I (Pair of Compounds)	List - II (Isomerism)
(A) n-propanol and Isopropanol	(I) Metamerism
(B) Methoxypropane and ethoxyethane	(II) Chain Isomerism
(C) Propanone and propanal	(III) Position Isomerism
(D) Neopentane and Isopentane	(IV) Functional Isomerism
(1) (A)–(II), (B)–(I), (C)–(IV), (D)–(III)	
(2) (A)–(III), (B)–(I), (C)–(II), (D)–(IV)	
(3) (A)–(I), (B)–(III), (C)–(IV), (D)–(II)	
(4) (A)–(III), (B)–(I), (C)–(IV), (D)–(II)	

Ans. (4)



69. The quantity of silver deposited when one coulomb charge is passed through AgNO_3 solution:
- (1) 0.1 g atom of silver
 - (2) 1 chemical equivalent of silver
 - (3) 1 g of silver
 - (4) 1 electrochemical equivalent of silver

Ans. (4)

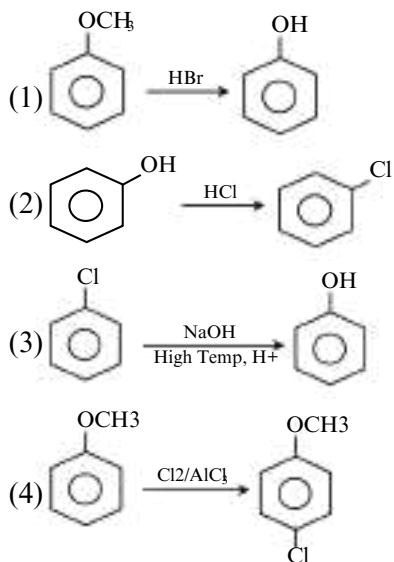
Sol. $W = ZIt$

$$W = ZQ$$

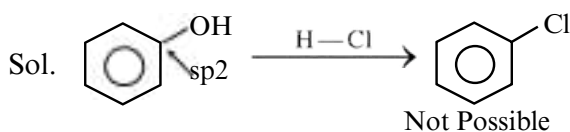
$$Q = \frac{W}{Z}$$

$$W = ZQ = (\text{electrochemical equivalent})$$

70. Which one of the following reactions is NOT possible?



Ans. (2)



71. Given below are two statements :

Statement I : The metallic radius of Na is 1.86 \AA° and the ionic radius of Na^+ is lesser than 1.86 \AA° .
Statement II : Ions are always smaller in size than the corresponding elements.

In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is correct but Statement II is false
- (2) Both Statement I and Statement II are true
- (3) Both Statement I and Statement II are false
- (4) Statement I is incorrect but Statement II is true

Ans. (1)

Sol. $r_{\text{Na}} > r_{\text{Na}^+}$

So, Statement (I) is correct but size of anions are greater than size of neutral atoms.

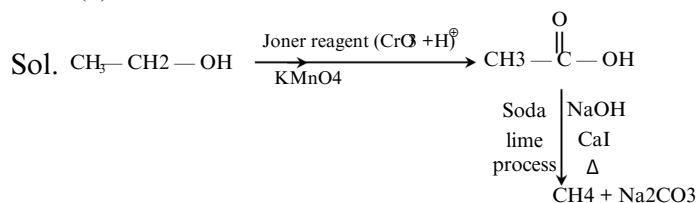
So statement (II) is incorrect.

72. $\text{CH}_3\text{CH}_2-\text{OH} \xrightarrow[\text{(iii) NaOH, CaO, } \Delta]{\text{(i) Jones's Reagent, (ii) KMnO}_4}$

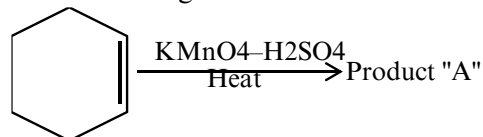
Consider the above reaction sequence and identify the major product P.

- (1) Methane
- (2) Methanal
- (3) Methoxymethane
- (4) Methanoic acid

Ans. (1)



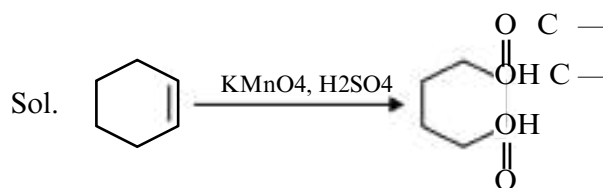
73. Consider the given chemical reaction :



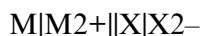
Product "A" is :

- (1) picric acid
- (2) oxalic acid
- (3) acetic acid
- (4) adipic acid

Ans. (4)



74. For the electro chemical cell



$$\text{If } E_{M^{2+}/M}^{\circ} = 0.46V \text{ and } E_{X^{2-}/X}^{\circ} = 0.34V.$$

Which of the following is correct ?

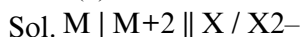
(1) $E_{\text{cell}} = -0.80 \text{ V}$

(2) $M + X \rightarrow M^{2+} + X^{2-}$ is a spontaneous reaction

(3) $M^{2+} + X^{2-} \rightarrow M + X$ is a spontaneous reaction

(4) $E_{\text{cell}} = 0.80 \text{ V}$

Ans. (3)



$$E_{\text{cell}} = E_{M^{2+}/M}^{\circ} - E_{X^{2-}/X}^{\circ}$$

$$= -0.46 + 0.34 = -0.12V$$

As E_{cell} is negative so anode becomes cathode and cathode become anode. Spontaneous reaction will be $M^{2+} + X^{2-} \rightarrow M + X$

75. The number of moles of methane required to produce

11g $\text{CO}_2(\text{g})$ after complete combustion is:

(Given molar mass of methane in g mol^{-1} : 16)

(1) 0.75

(2) 0.25

(3) 0.35

(4) 0.5

Ans. (2)



4gm

11gm

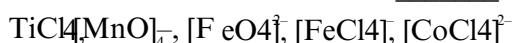
0.25 mole

0.25 mole

0.25 mole CH_4 gives 0.25 mole (or 11gm) CO_2

76. The number of complexes from the following with

no electrons in the t_2 orbital is _____.



(1) 3

(2) 1

(3) 4

(4) 2

Ans. (1)



Mn^{+7}



Fe^{+6}



Fe^{+2}



Co^{+2}



77. The number of ions from the following that have the ability to liberate hydrogen from a dilute acid is _____. Ti^{2+} , Cr^{2+} and V^{2+}

(1) 0

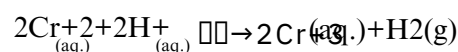
(2) 2

(3) 3

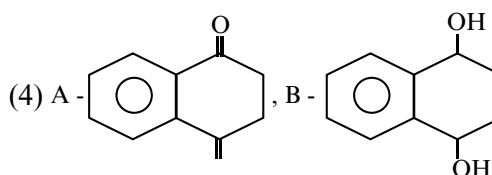
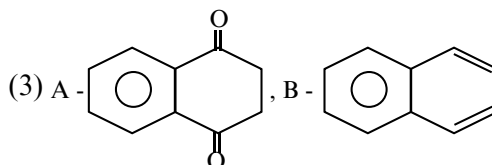
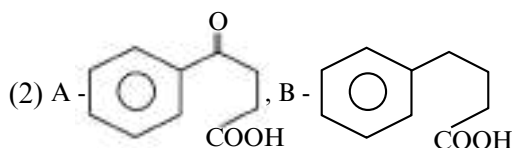
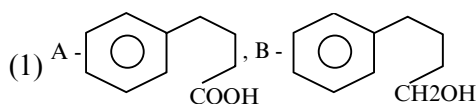
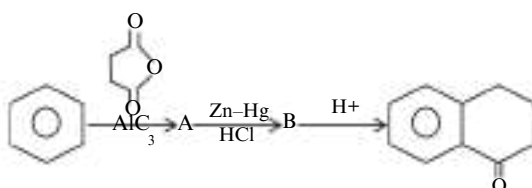
(4) 1

Ans. (3)

Sol. The ions Ti^{2+} , V^{2+} Cr^{2+} are strong reducing agents and will liberate hydrogen from a dilute acid, eg.

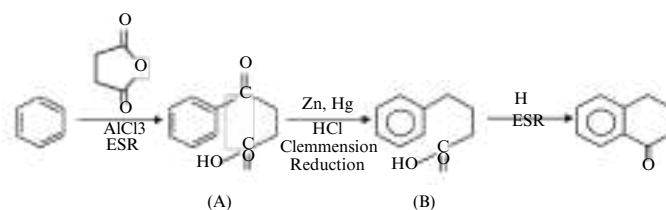


78. Identify A and B in the given chemical reaction sequence :-



Ans. (2)

Sol.



79. The correct statements from the following are :
- (A) The decreasing order of atomic radii of group 13 elements is $Tl > In > Ga > Al > B$.
- (B) Down the group 13 electronegativity decreases from top to bottom.
- (C) Al dissolves in dil. HCl and liberate H_2 but conc. HNO_3 renders Al passive by forming a protective oxide layer on the surface.
- (D) All elements of group 13 exhibits highly stable +1 oxidation state.
- (E) Hybridisation of Al in $[Al(H_2O)_6]^{3+}$ ion is sp^3d^2 .

Choose the correct answer from the options given below :

- (1) (C) and (E) only
- (2) (A), (C) and (E) only
- (3) (A), (B), (C) and (E) only
- (4) (A) and (C) only

Ans. (1)

Sol. A. size order $Tl > In > Al > Ga > B$

B. Electronegativity order $B > Al < Ga < In < Tl$

D. B, Al are more stable in +3 oxidation state

So, only C, E statements are correct.

80. Coagulation of egg, on heating is because of :

- (1) Denaturation of protein occurs
- (2) The secondary structure of protein remains unchanged
- (3) Breaking of the peptide linkage in the primary structure of protein occurs
- (4) Biological property of protein remains unchanged

Ans. (1)

Sol. Coagulation of egg give primary structure of protein, which is known as denaturation of protein

SECTION-B

81. Combustion of 1 mole of benzene is expressed at

$$C_6H_6(l) + \frac{15}{2} O_2(g) \rightarrow C_6H_6O(l) + 3H_2O(l)$$

The standard enthalpy of combustion of 2 mol of benzene is $-x$ kJ.

$x =$ _____.

- (1) standard Enthalpy of formation of 1 mol of $C_6H_6(l)$, for the reaction
- $$6C(\text{graphite}) + 3H_2(g) \rightarrow C_6H_6(l) \text{ is } 48.5 \text{ kJ mol}^{-1}.$$
- (2) Standard Enthalpy of formation of 1 mol of $CO_2(g)$, for the reaction
- $$C(\text{graphite}) + O_2(g) \rightarrow CO_2(g) \text{ is } -393.5 \text{ kJ mol}^{-1}.$$
- (3) Standard and Enthalpy of formation of 1 mol of $H_2O(l)$, for the reaction



Ans. (6535)

Sol. $6C(\text{graphite}) + 3H_2(g) \rightarrow C_6H_6(l); \Delta H = 48.5 \text{ kJ/mol}$

$C(\text{graphite}) + O_2(g) \rightarrow CO_2(g); \Delta H = -393.5 \text{ kJ/mol}$

$H_2(g) + \frac{1}{2} O_2(g) \rightarrow H_2O(l); \Delta H = -286 \text{ kJ/mol}$

equation $-(1) \times 1 + (2) \times 6 + (3) \times 3$

$-48.5 - 6 \times 393.5 - 3 \times 286$

$= -3267.5 \text{ kJ for 1 mol}$

$= -6535 \text{ kJ for 2 mol}$

Ans. 6535 kJ

82. The fusion of chromite ore with sodium carbonate in the presence of air leads to the formation of products A and B along with the evolution of CO_2 . The sum of spin-only magnetic moment values of

A and B is ____ B.M. (Nearest integer)

(Given atomic number : C : 6, Na : 11, O : 8,

Fe : 26, Cr : 24]

Sol. $4FeCrO_4 + 8Na_2CO_3 + 7O_2$

$8Na_2CrO_4 + 2Fe_2O_3 + 8CO_2$

A B

Spin only magnetic moment

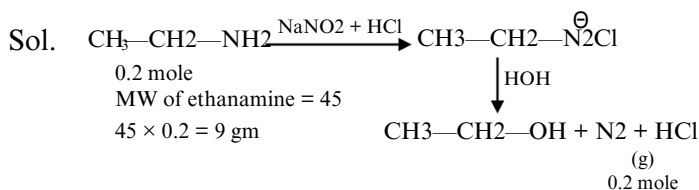
For $Na_2CrO_4 \mu_B = 0$

For $Fe_2O_3 \mu_B = 5.9$

sum = 5.9

83. X of ethanamine was subjected to reaction with NaNO_2/HCl followed by hydrolysis to liberate N_2 and HCl . The HCl generated was completely

Ans. (9) Neutralised by 0.2 moles of NaOH . X is ____ g.



84. In an atom, total number of electrons having quantum numbers $n = 4$, $|m_l| = 1$ and $m_s = \frac{1}{2}$ is

Ans. (6)

Sol. $n = 4$

\square m \square

0 0

1 -1, 0, +1

2 -2, -1, 0, +1, +2, +3

So number of orbital associated with

$n = 4$, $|m_l| = 1$ are 6

Now each orbital contain one e^- with $m_s = \frac{1}{2}$

85. Using the given figure, the ratio of R_f values of sample A and sample C is $x \times 10^{-2}$. Value of x is

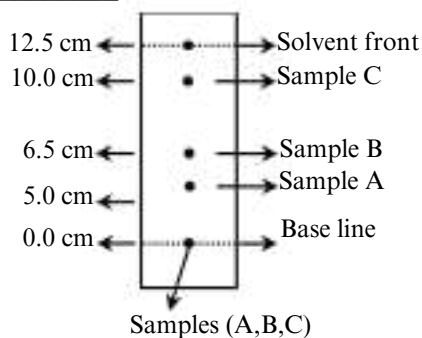


Fig : Paper chromatography of Samples

Ans. (50)

Sol. R_f of A = $\frac{5}{12.5}$ R_f of C = $\frac{10}{12.5}$

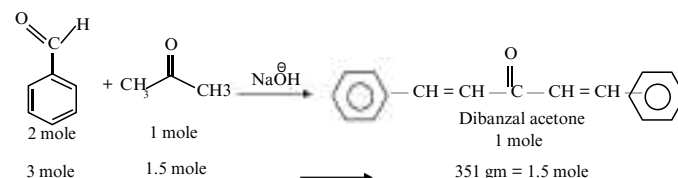
$$\text{Ratio} = \frac{R_{f(A)}}{R_{f(C)}} = \frac{1}{2} = 0.5 \text{ or } 50 \times 10^{-2}$$

86. In the Claisen-Schmidt reaction to prepare 351 g of dibenzalacetone using 87 g of acetone, the amount of benzaldehyde required is _____ g.

(Nearest integer)

Ans. (318)

Sol. Claisen Schmidt reaction



mw of benzaldehyde = 106

$106 \times 3 = 318 \text{ gm}$. Benzaldehyde is required to give 1.5 mole (or 351 gm) product

87. Consider the following single step reaction in gas

phase at constant temperature.

$2\text{A(g)} + \text{B(g)} \rightleftharpoons \text{C(g)}$ The initial rate of the reaction is recorded as r_1

when the reaction starts with 1.5 atm pressure of A

and 0.7 atm pressure of B. After some time,

the rate r_2 is recorded when the pressure of C becomes 0.5 atm. The ratio $r_1 : r_2$ is _____ ×

Ans. (315)

Sol. $2\text{A(g)} + \text{B(g)} \rightleftharpoons \text{C(g)}$
 r_1 1.5 atm 0.7 atm
 r_2 0.5 atm 0.2 atm 0.5 atm

$$r = K [\text{A}]^2 [\text{B}]$$

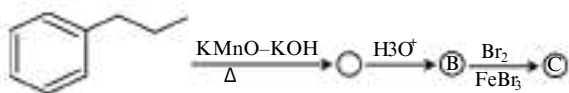
$$r_1 = K [1.5]^2 [0.7]$$

$$r_2 = K [0.5]^2 [0.2]$$

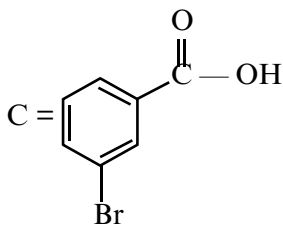
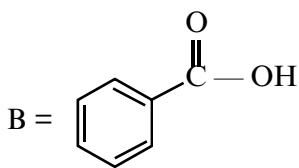
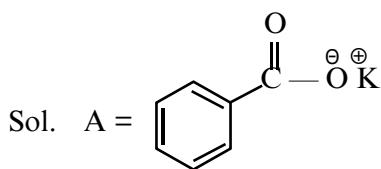
$$\frac{r_1}{r_2} = 9 \times \frac{7}{2} = 31.5 = 315 \times 10^{-1}$$

Ans. 315

88. The product © in the following sequence of reactions has _____ π bonds.



Ans. (4)



π bonds = 4

89. Considering acetic acid dissociates in water, its dissociation constant is 6.25×10^{-5} . If 5 mL of acetic acid is dissolved in 1 litre water, the solution will freeze at $-x \times 10^{-2} ^\circ\text{C}$, provided pure water freezes at $0 ^\circ\text{C}$.
 $x =$ _____. (Nearest integer)

Given : $(K_f)_{\text{water}} = 1.86 \text{ K kg mol}^{-1}$

density of acetic acid is 1.2 g mol^{-1}

molar mass of water = 18 g mol^{-1} .

molar mass of acetic acid = 60 g mol^{-1} .

density of water = 1 g cm^{-3}

Acetic acid dissociates as

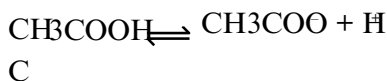


Ans. (19)

- Sol. Mass of $\text{CH}_3\text{COOH} = V \times d$
 $= 5 \text{ ml} \times 1.2 \text{ g/ml}$
 $= 6 \text{ gm}$

$$n_{\text{CH}_3\text{COOH}} = \frac{6}{60} = 0.1 \text{ mol}$$

$$m_{\text{CH}_3\text{COOH}} \approx M \quad \text{CH}_3\text{COOH} \rightleftharpoons \text{CH}_3\text{COO}^- + \text{H}^+ \quad 0.1 = 0.1 \text{ M}$$



$$K_a = \frac{\alpha^2}{1 - \alpha}$$

$$1 - \alpha \approx 1 \Rightarrow K_a = \alpha^2$$

$$\alpha = \sqrt{\frac{K_a}{C}} = \sqrt{\frac{6.25 \times 10^{-5}}{0.1}} = 25 \times 10^{-3}$$

$$\text{V.f. (i)} = 1 + \alpha(n - 1) = 1 + \alpha(2 - 1) = 1 + \alpha$$

$$= 1 + 25 \times 10^{-3} = 1.025$$

$$\Delta T_f = i K_f m$$

$$= (1.025)(1.86)(0.1)$$

$$= 0.19$$

$$= 19 \times 10^{-2}$$

90. Number of compounds from the following with zero dipole moment is _____.

HF , H_2 , H_2S , CO_2 , NH_3 , BF_3 , CH_4 , CHCl_3 , SiF_4 , H_2O , BeF_2

Ans. (6)

Sol. H_2 , CO_2 , BF_3 , CH_4 , SiF_4 , BeF_2

are symm. molecule so dipole moment is zero