

BITSAT 2019 Question Paper with Answer Key

Birla Institute of Technology and Science Admission Test

BITSAT : SOLVED PAPER 2019

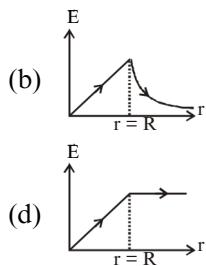
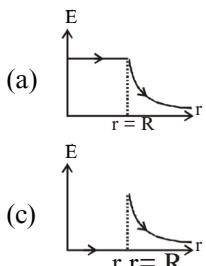
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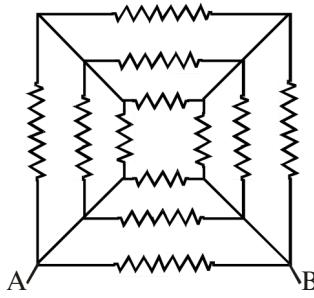
INSTRUCTIONS

- This question paper contains total 52 questions divided into four parts
Part I Physics Q No 1 to 15
Part II Chemistry Q No 16 to 25
Part III English Proficiency Q No 26 to 35
Part IV Logical Reasoning Q No 36 to 52
 - Part II questions are multiple choice questions with four options only one of them is correct
 - Each correct answer awarded 1 mark and -1 for each incorrect answer
 - Duration of paper 2 hours

PART - I : PHYSICS

1. Which one of the following graphs represents the variation of electric field with distance r from the centre of a charged spherical conductor of radius R ?

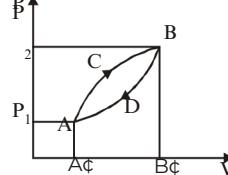
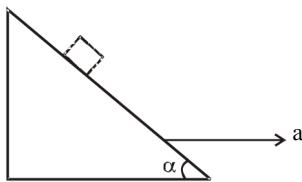


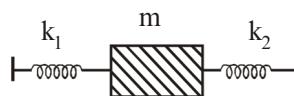
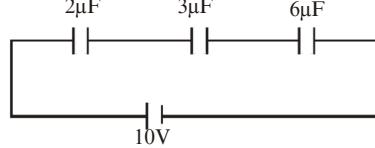


The time period of a satellite of earth is 5 hours. If the separation between the earth and the satellite is increased to 4 times the previous value, the new time period will become

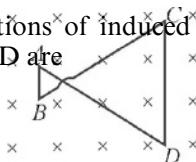
Two trains are moving towards each other with speeds of 20 m/s and 15 m/s relative to the ground. The first train sounds a whistle of frequency 600 Hz. The frequency of the whistle heard by a passenger in the second train before the train meets, is (the speed of sound in air is 340 m/s)

7. You are asked to design a shaving mirror assuming that a person keeps it 10 cm from his face and views the magnified image of the face at the closest comfortable distance of 25 cm. The radius of curvature of the mirror would then be : (b) -24 cm
 (c) -60 cm (d) 24 cm
8. A block is kept on a frictionless inclined surface with angle of inclination ' α '. The incline is given an acceleration ' a ' to keep the block stationary. Then ' a ' is equal to
- (a) $g \sec \alpha$
 (b) $g/\tan \alpha$
 (c) $g \tan \alpha$
 (d) g
9. With the increase in temperature, the angle of contact
- (a) decreases
 (b) increases
 (c) remains constant
 (d) sometimes increases and sometimes decreases
10. Forward biasing is that in which applied voltage
- (a) increases potential barrier
 (b) cancels the potential barrier
 (c) is equal to 1.5 volt
 (d) None of these
11. Number of significant figures in expression $\frac{4.327 \text{ g}}{2.51 \text{ cm}^3}$ is
- (a) 2 (b) 4 (c) 3 (d) 5
12. The ratio of the specific heats $\frac{C_p}{C_v} = g$ in terms of degrees of freedom (n) is given by
- (a) $\frac{g}{e} \div \frac{n}{3\phi}$ (b) $\frac{g}{e} \div \frac{n}{\phi}$
 (c) $\frac{g}{e} + \frac{n}{2\phi}$ (d) $\frac{g}{e} + \frac{n}{\phi}$
13. A stone is thrown with a velocity u making an angle q with the horizontal. The horizontal distance covered by its fall to ground is maximum when the angle q is equal to
- (a) 0° (b) 30° (c) 45° (d) 90°
14. A ball of mass 150 g, moving with an acceleration 20 m/s^2 , is hit by a force, which acts on it for 0.1 sec. The impulsive force is
- (a) 0.5 N (b) 0.1 N (c) 0.3 N (d) 1.2 N
15. A man drags a block through 10 m on rough surface ($\mu = 0.5$). A force of 3 kN acting at 30° to the horizontal. The work done by applied force is (a) zero (b) 7.5 kJ (c) 5 kJ (d) 10 kJ
16. A force of $2i + 3j + 4k$ N acts on a body for 4 second, produces a displacement of $(3i + 4j + 5k)$ m. The power used is
- (a) 9.5 W (b) 7.5 W (c) 6.5 W (d) 4.5 W
17. The Earth is assumed to be a sphere of radius R . A platform is arranged at a height R from the surface of the Earth. The escape velocity of a body from this platform is $\frac{v}{2}$, where v is its escape velocity from the surface of the Earth. The areal velocity of a planet is a consequence of the law of conservation of (a) (b) (c) (d) Energy
 Angular momentum
 Linear momentum
 None of these
18. According to the law of conservation of energy, the escape velocity of a planet is a consequence of the law of conservation of (a) (b) (c) (d) Energy
 Angular momentum
 Linear momentum
 None of these
19. Water is flowing through a horizontal tube having cross-sectional areas of its two ends being A and A' such that the ratio A/A' is 5. If the pressure difference of water between the two ends is $3 \times 10^5 \text{ N/m}^2$, the velocity of water with which it enters the tube will be (neglect gravity effect) (a) 10 m s^{-1} (b) $10 \sqrt{m} \text{ s}^{-1}$
 (c) 25 m s^{-1} (d) $50 \sqrt{m} \text{ s}^{-1}$
20. A thermodynamic system is taken from state A to B along ACB and is brought back to A along BDA as shown in the PV diagram. The net work done during the complete cycle is given by the area
- (a) $P_1 A C B P_2 P_1$
 (b) $A C B B' A' A$
 (c) $A C B D A' A$
 (d) $A D B B' A' A$



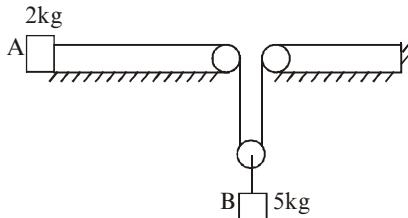
21. A boat crosses a river from port A to port B, which are just on the opposite side. The speed of water is V_w and that of boat is V_B relative to water. What is the time taken by the boat if it has to cross the river directly on the AB line [D = width of the river]
27. Two bodies A and B are placed in an evacuated vessel maintained at a temperature of 27°C . The temperature of A is 327°C and that of B is 227°C . The ratio of heat loss from A and B is about (d) 1 : 4
- If a rigid body is rotating about an axis with a constant velocity, then
- (a) Velocity, Angular velocity of all particles will be same
 (b) Velocity, Angular velocity of all particles will be different
 (c) Velocity of all particles will be different but angular velocity will be same.
 (d) Angular velocity of all particles will be different but velocity will be same.
- The fundamental frequency of an open organ pipe is 300 Hz. The first overtone of this pipe has same frequency as first overtone of a closed organ pipe. If speed of sound is 330 m/s, then the length of closed organ pipe is
- (a) 41 cm (b) 30 cm (c) 45 cm (d) 35 cm
- In Young's experiment, the distance between the slits is reduced to half and the distance between the slit and screen is doubled, then the fringe width
- (a) will not change
 (b) will become half
 (c) will be doubled
 (d) will become four times
- If a rolling body's angular momentum changes by 20 SI units in 3 seconds, by a constant torque. Then find the torque on the body
- (a) $20/3$ SI units (b) $100/3$ SI units
 (c) 20 SI units (d) 5 SI units
- Charge Q is distributed to two different metallic spheres having radii x and $2x$ such that both spheres have equal surface charge density, then charge on large sphere is
- (a) $\frac{4Q}{5}$ (b) $\frac{Q}{5}$ (c) $\frac{3Q}{5}$ (d) $\frac{5Q}{4}$
- In an LR circuit $f = 50$ Hz, $L = 2$ H, $E = 5$ volts, $R = 1$ W then energy stored in inductor is
- (a) 50 J (b) 25 J
 (c) 100 J (d) None of these
- A straight wire of length 0.5 metre and carrying a current of 1.2 ampere is placed in uniform magnetic field of induction 2 tesla. The magnetic field is perpendicular to the length of the wire. The force on the wire is
- (a) 2.4 N (b) 1.2 N (c) 3.0 N (d) 2.0 N
22. Two springs, of force constants k_1 and k_2 are connected to a mass m as shown. The frequency of oscillation of the mass is f . If both k are made four times their original values, the frequency of oscillation becomes
- 
- (a) $2f$ (b) $f/2$ (c) $f/4$ (d) $4f$
23. When a potential difference V is applied across a conductor at a temperature T , the drift velocity of electrons is proportional to
- (a) \sqrt{V} (b) V (c) \sqrt{T} (d) T
24. The amplitude of a damped oscillator becomes $\frac{1}{2}$ in 2 seconds. If its amplitude after 6 seconds is $\frac{1}{n}$ times the original amplitude, the value of n is
- (a) 32 (b) 33 (c) $3\sqrt{3}$ (d) 23
25. The angular speed of the electron in the n th orbit of Bohr hydrogen atom is
- (a) directly proportional to n
 (b) inversely proportional to \sqrt{n}
 (c) inversely proportional to n^2
 (d) inversely proportional to n^3
26. In the given figure, the charge on $3 \mu\text{F}$ capacitor is
- 
- (a) 10 μC (b) 15 μC
 (c) 30 μC (d) 5 μC
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The directions of induced current in wires AB and CD are



- (a) B to A and D to C (b) A to B and C to D
(c) A to B and D to C (d) B to A and C to D

39. Find the acceleration of block A and B. Assume pulley is massless.



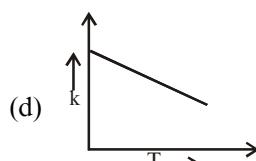
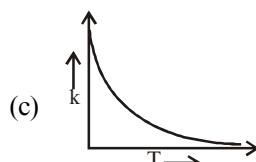
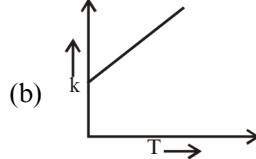
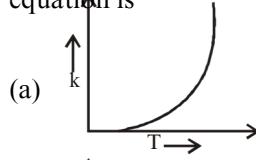
- (a) $\frac{10}{13}$ g, $\frac{5}{13}$ g (b) $\frac{1}{13}$ g, $\frac{5}{13}$ g
 (c) $\frac{9}{13}$ g, $\frac{11}{13}$ g (d) $\frac{10}{5}$ g, $\frac{13}{5}$ g

40. The nuclei of which one of the following pairs of nuclei are isotones?

 - (a) $^{34}_{\text{Se}}$, $^{71}_{\text{Ga}}$
 - (b) $^{84}_{\text{Sr}}$, $^{86}_{\text{Sr}}$
 - (c) $^{92}_{\text{Mo}}$, $^{92}_{\text{Zr}}$
 - (d) $^{40}_{\text{Ca}}$, $^{32}_{\text{S}}$

PART - II : CHEMISTRY

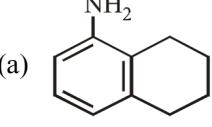
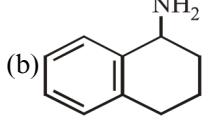
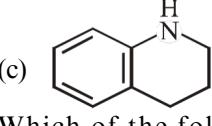
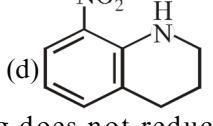
41. Plots showing the variation of the rate constant (k) with temperature (T) are given below. The plot that follows Arrhenius equation is



42. 3.6 g of oxygen is adsorbed on 1.2 g of metal powder. What volume of oxygen adsorbed per gram of the adsorbent at 1 atm and 173 K? (b)
None of these)

43. In the purification of impure nickel by Mond's process, metal is purified by :
(a) Electrolytic reduction
) Vapour phase thermal decomposition
(b) Thermite reduction
) Carbon reduction
(c)
(d)
)

44. When chlorine water is added to an aqueous solution of sodium iodide in the presence of chloroform, a violet colouration is obtained. On adding more of chlorine water and vigorous shaking, the violet colour disappears. This shows the conversion of into HI
) $\text{HI}, \text{H}_2\text{IO}_3$) I_2, HOI
45. In the clathrates of xenon with water, the nature of bonding between xenon and water
 (a) covalent
 (d) hydrogen bonding
 The covalent electronic configurations of Eu(Atomic induced dipole No. 63), Gd(Atomic No. 64) and Tb (Atomic No. 65) are
 (a) $[\text{Xe}]4f^7 6s^2$, $[\text{Xe}]4f^8 6s^2$ and $[\text{Xe}]4f^8 5d^1 6s^2$
 (b) $[\text{Xe}]4f^7 5d^1 6s^2$, $[\text{Xe}]4f^7 5d^1 6s^2$ and $[\text{Xe}]4f^9 6s^2$
 (c) $[\text{Xe}]4f^6 5d^1 6s^2$, $[\text{Xe}]4f^7 5d^1 6s^2$ and $[\text{Xe}]4f^8 5d^1 6s^2$
 (d) $[\text{Xe}]4f^7 6s^2$, $[\text{Xe}]4f^7 5d^1 6s^2$ and $[\text{Xe}]4f^9 6s^2$
46. Which of the following carbonyls will have the strongest C – O bond ?
 (a) $[\text{Mn}(\text{CO})_6]^+$ (b) $[\text{Cr}(\text{CO})_6]$
 (c) $[\text{V}(\text{CO})_6]^-$ (d) $[\text{Fe}(\text{CO})_6]$
47. How many chiral compounds are possible on monochlorination of 2- methyl butane ?
 (a) 8 (b) 2 (c) 4 (d) 6
48. Which of the following are intermediates in the reaction of excess of CHMgBr with $\text{CH}_3\text{COOCH}_2$ to make 2-phenyl - 2-propanol ?
 OMgBr
- A. $\text{C}_6\text{H}_5 - \overset{\div}{\underset{\text{C}}{\text{C}}} - \text{OCH}_2\text{CH}_3$
 \div
 OCH_3
- B. $\text{C}_6\text{H}_5 - \overset{\div}{\underset{\text{C}}{\text{CH}_3}}$
 OMgB
- C. $\text{C}_6\text{H}_5 - \overset{\div}{\underset{\text{C}}{\text{CH}_3}}$
 (a) A and CH_3 (b) A, B and C
) B A) B and C
 (c) and C) (d)

- O
50. $\text{CH}_3 - \overset{\div}{\underset{\text{C}}{\text{C}}} \text{CH}_2 - \text{CH}_3 + \text{CH}_3\text{MgBr} \xrightarrow[170^\circ\text{C}]{\frac{3}{4}\text{H}_2\text{O}_2, \frac{1}{4}\text{S}_2\text{O}_8}$
 What is Z?
- OH
- (a) $\text{CH}_3 - \overset{\div}{\underset{\text{C}}{\text{C}}} - \text{CH}_2 - \text{CH}_3$
 \div
- (b) $\text{CH}_3 - \overset{\text{CH}_3}{\underset{\div}{\text{C}}} = \text{CH} - \text{CH}_3$
 \div
 CH_3
 $\overset{\text{C}_2\text{H}_5}{\div} \overset{\text{C}_2\text{H}_5}{\div}$
- (c) $\text{CH}_3 - \overset{\div}{\underset{\text{CH}_3}{\text{C}}} - \text{O} - \overset{\div}{\underset{\text{CH}_3}{\text{C}}} - \text{CH}_3$
 $\div \quad \div$
- (d) $\text{CH}_2 = \overset{\div}{\underset{\text{CH}_3}{\text{C}}} - \text{CH}_2 - \text{CH}_3$
51. Which of the following is the strongest base ?
- (a) 
- (b) 
- (c) 
- (d) 
52. Which of the following does not reduce Benedict's solution?
 (a) Glucose (b) Fructose
 (c) Sucrose (d) Aldehyde
53. General formula of solid in zinc blende structure is:
 (a) AB^2 (b) AB_3 (c) AB (d) A_2B
54. Glycine in alkaline solution exists as _____ and migrates to _____.
 (a) Cation, cathode
) Neutral, anode
 (b) Zwitter ion, cathode
) anion, anode
 (c)
 (d)
)

55. Product on reaction of ethanamide with phosphorus pentoxide is:
- ethanamine
 - acetonitrile
 - ethanol
 - ethane isonitrile
56. K_a of HX is 10^{-5} , then find concentration of HO^- when equal volumes of 0.25M HX and 0.05M NaOH are mixed.
- $4 \times 10^{-5}\text{M}$
 - $6 \times 10^{-5}\text{M}$
 - $8 \times 10^{-3}\text{M}$
 - $2 \times 10^{-5}\text{M}$
57. Net cell reaction of $\text{Pt}_2 | \text{H} (640\text{ mm}) | \text{H}_2\text{Cl} | \text{H} (510\text{ mm}) | \text{Pt}$.
 $\text{A}^{3/4} \text{X} @ \text{B}^{3/4} \text{reag}^{3/4} \text{yer}^{3/4} \text{reag}^{3/4}$
- 0.89V
 - 0.93V
 - $2.91 \times 10^{-3}\text{V}$
 - $2.5 \times 10^{-2}\text{V}$
58. Which of the following has zero net dipole moment? (a) XeF_4 (b) BrF_3 (c) ClF_3 (d) SF_4
59. Which of the following element has the highest ionisation enthalpy?
- Boron
 - Aluminium
 - Germanium
 - Thallium
60. Out of the elements with atomic number 7, 8, 9, 13 which has the smallest size and highest ionization enthalpy?
- 7
 - 8
 - 9
 - 13
61. Which one is classified as a condensation polymer?
- Dacron
 - Neoprene
 - Teflon
 - Acrylonitrile
62. Which of the following compounds is not an antacid?
- Phenelzine
 - Ranitidine
 - Aluminium hydroxide
 - Cimetidine
63. Mole fraction of the solute in a 1.00 molal aqueous solution is
- 0.1770
 - 0.0177
 - 0.0344
 - 1.7700
64. The IUPAC name of the following compound is
-
- trans-2-chloro-3-iodo-2-pentene
 - cis-3-iodo-4-chloro-3-pentene
 - trans-3-iodo-4-chloro-3-pentene
 - cis-2-chloro-3-iodo-2-pentene
65. Most stable carbocation among the following is:
- -
 -
 -
- Which is correct for the following changes?
-
66. The stability of +1 oxidation state among Al, Ga, In and Tl increases in the sequence :
- $\text{Ga} < \text{In} < \text{Al} < \text{Tl}$
 - $\text{Al} < \text{Ga} < \text{In} < \text{Tl}$
 - $\text{Tl} < \text{In} < \text{Ga} < \text{Al}$
 - $\text{In} < \text{Tl} < \text{Ga} < \text{Al}$
67. Which of the following alkaline earth metal hydroxides is amphoteric in character?
- $\text{Be}(\text{OH})_2$
 - $\text{Ca}(\text{OH})_2$
 - $\text{Sr}(\text{OH})_2$
 - $\text{Ba}(\text{OH})_2$
68. Which reaction shows oxidising nature of O_2 ?
- $\text{H}_2\text{O}_2 + 2\text{KI} \rightarrow 2\text{KOH} + \text{I}_2$
 - $\text{Cl}_2 + \text{H}_2\text{O}_2 \rightarrow \text{HCl} + \text{O}_2$
 - $2\text{O}_2 + \text{Ag}_2\text{O} \rightarrow 2\text{Ag} + \text{H}_2\text{O} + \text{O}_2$
 - $\text{NaClO} + \text{H}_2\text{O}_2 \rightarrow \text{NaCl} + \text{H}_2\text{O} + \text{O}_2$
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 - $2\text{O}_2 + \text{Ag}_2\text{O} \rightarrow 2\text{Ag} + \text{H}_2\text{O} + \text{O}_2$
 - $\text{NaClO} + \text{H}_2\text{O}_2 \rightarrow \text{NaCl} + \text{H}_2\text{O} + \text{O}_2$
70. a $\text{K}_2\text{Cr}_2\text{O}_7$ b KCl c H_2SO_4 d CrO_2
 $\text{Cl}_2 + \text{y KHSO}_4 + \text{z H}_2\text{O}$
- The above equation balances when
- $a = 2, b = 4, c = 6$ and $x = 2, y = 6$,
 - $z = 3$ $a = 4, b = 2, c = 6$ and $x = 6, y = 2$,
 - $a = 2, z = 3$ $a = 6, b = 4, c = 2$ and $x = 6, y = 3$,
 - $a = 1, b = 4, c = 6$ and $x = 2, y = 6, z = 3$

71. For the reactions



K_c for the reaction $A \rightleftharpoons D$ is

$$(a) \frac{2}{2} \times \frac{4}{4} \times \frac{6}{6}$$

$$(b) \frac{2}{4} \times \frac{4}{6}$$

$$\underline{\underline{2}}$$

72. Which of the following will always lead to a non-spontaneous change?

(a) DH and DS both +ve

) DH is -ve DS both +ve

(b) DH and DS both -ve

) DH is +ve DS both -ve

73. The densities of two gasses are in the ratio of 1:

{(a). The ratio of their rates of diffusion is

{(a) 16:1 (b) 4:1 (c) 1:4 (d) 1:16

74. In the reaction $2\text{PCl}_5 \rightarrow \text{PCl}_4 + \text{PCl}_6$, the

change in hybridisation is from

{(a) sp^3d to sp^3 and sp^3d^2

{(b) sp^3d to sp^2 and sp^3

{(c) sp^3d^2 to sp^3d^2 and sp^3d^3

{(d) sp^3d^2 to sp^3 and sp^3d

75. The group having isoelectronic species is:

(a) O_2^- , F^- , Na^+ , Mg^{2+}

(b) O_2^- , F^- , Na^- , Mg^+

(c) O_2^- , F^- , Na^- , Mg^{2+}

(d) O_2^- , F^- , Na^+ , Mg^{2+}

76. 100 mL O_2 and H_2 kept at same temperature and pressure. What is true about their number of molecules

(a) $N_{\text{O}_2} > N_{\text{H}_2}$

) $N_{\text{O}_2} < N_{\text{H}_2}$

) $N_{\text{O}_2} = N_{\text{H}_2}$

(c) $N_{\text{O}_2} + N_{\text{H}_2} = 1$ mole

77. If m_A gram of a metal A displaces m_B gram of another metal B from its salt solution and if the equivalent mass are E_A and E_B respectively then

equivalent mass of A can be expressed as:

$$(a) EA = \frac{m_A}{m_B} \cdot EB$$

$$(b) \frac{m_A}{E_A} = \frac{m_B}{EB}$$

$$E_A = \frac{m_B}{m_A \cdot EB}$$

$$(c) EA = \frac{m_B}{A} \cdot EB$$

$$(d) EA = \sqrt{\frac{m}{m_B}} \cdot EB$$

78. Which one of the following set of quantum numbers is not possible for 4p electron?

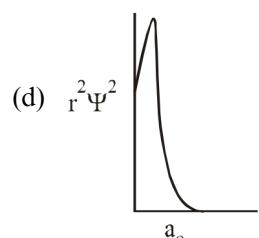
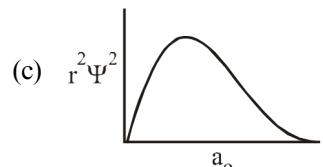
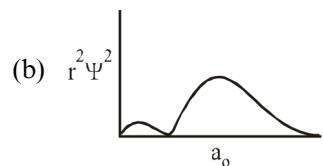
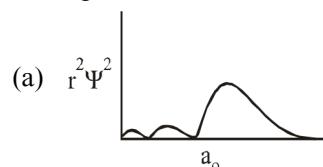
$$(a) n = 4, l = 1, m_s = -1, m_l = \frac{1}{2}$$

$$(b) n = 4, l = 1, m_s = 0, m_l = \frac{1}{2}$$

$$(c) n = 4, l = 1, m_s = 2, m_l = \frac{1}{2}$$

$$(d) n = 4, l = 1, m_s = -1, m_l = -\frac{1}{2}$$

79. Which of the following radial distribution graphs correspond to $l = 2$ for the H atom?



80. Which of the following is paramagnetic?

(a) B_2 (b) C_2 (c) N_2 (d) F_2

PART - III (A): ENGLISH

DIRECTIONS (Qs. 81-83) : In the following questions below, out of the four alternatives, choose the one which best expresses the meaning of the given word.

DIRECTIONS (Qs. 84-86) : In the following questions, choose the word opposite in meaning to the given word.

84. Knack :
(a) Talent (b) Dullness
(c) Dexterity (d) Balance

85. Pernicious :
(a) Prolonged (b) Ruinous
(c) Ruthless (d) Beneficial

86. Opulence :
(a) Luxury (b)
Transparency
(c) Weath (d) Poverty

DIRECTIONS (Qs. 87-90) : Read the passage carefully and choose the best answer to each question out of the four alternatives and mark it by blackening the appropriate circle :

Like watering a plant, we grow our friendships [and all our relationships) by nurturing them. Friendships need the same attention as other relationships. If they are to continue. These relationships can be delightful, non-judgemental, supportive, understanding and fun.

Sometimes a friendship can bring out the positive side that you never show in any other relationship. This may be because the pressure of playing a 'role' (daughter, partner or child) is removed. With a friend you are free to be yourself and free to change. Of course, you are free to do this in all other relationships as well, but in friendships you get to have lots of rehearsals and discussion about changes.

as you experience them. It is an unconditional experience where you receive as much as you give. You can explain yourself to a friend openly without the fear of hurting a family member. How do friendships grow ? The answer is simple. By revealing yourself; being attentive; remembering what is most showing empathy; seeing the world through the eyes of your friend, you will understand the value of friendship. All this means learning to accept a person from a completely different family to your own or perhaps someone from a completely different cultural background. This is the way we learn tolerance. In turn ⁸⁷ In good friendships we gain tolerance and acceptance for our own differences.

(b) neither give nor receive.
(c) only give.
(d) only receive.

88. Empathy means

 - (a) someone else's misfortunes
 -) the ability to share and understand another
 - (b) feelings.
 -) skill and efficiency
 - (c) ability to do something

89. Through strong friendships, we gain

 - {(a) only acceptance.
 - (b) only attention.
 - (c) acceptance and tolerance.
 - (d) only tolerance.

90. Friendships and relationships grow when they are

(a) compared	(b) divided
(c) favoured	(d) nurtured

DIRECTIONS (Qs. 91-92) : In the following questions, sentences are given with blanks to be filled with an appropriate word(s). Four alternatives are suggested for each question. Choose the correct alternative out of the four as your answer.

91. There are not solitary, free-living creatures ; every form of life is _____ other forms.
(a) dependent on (b) parallel to
(d) segregated by

92. I'll take _____ now as I have another's appointment some where else.
(a) departure (b) your leave
(c) permission (d) leave from work

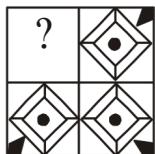
DIRECTIONS (Qs. 93-95) : In the following questions, some parts of the sentences have errors and some are correct. Find out which part of a sentence has an error. The number of that part is the answer. If a sentence is free from error, then your answer is (d). i.e., No error.

93. When one hears of the incident (a)/about the plane crash (b)/ he feels very sorry. (c)/ No error (d)
 94. I went there (a)/ with a view to survey (b)/ the entire procedure. (c)/ No error (d)
 95. It had laid (a)/ in the closet (b)/ for a week before we found it. (c)/ No error (d)

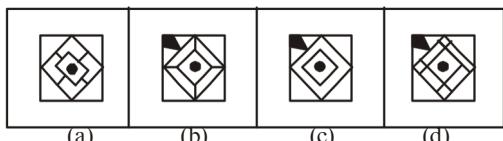
PART - III (B) : LOGICAL REASONIN

DIRECTIONS (Qs. 96 & 97) : In the following questions, which answer figure will complete the question figure?

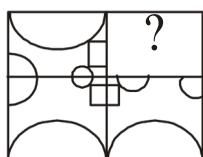
96. Question Figures :



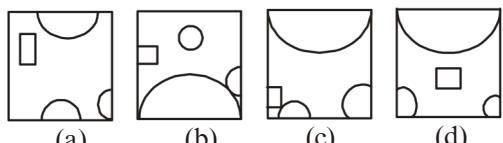
Answer figures :



97. Question Figure:

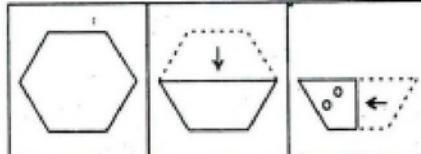


Answer Figure:

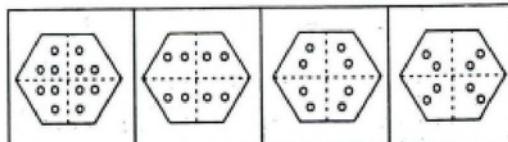


98. A piece of paper is folded and cut/punched as shown below in the question figures. From the given answer figures, indicate how it will appear when opened.

Question figures:



Answer figures:



(a) (b) (c) (d)

	7	6
	7	6
	77	88
	3773	3632

(a) 66 (b) 87 (c) 78 (d) 76

129. If A and B are two events, such that

$$P(A \cap B) = \frac{3}{4}, P(A \cup B) = \frac{1}{4}, P(A^c) = \frac{2}{3}$$

where A^c stands for the complementary event of A, then $P(B)$ is given by:

- | | |
|-------------------|-------------------|
| (a) $\frac{1}{3}$ | (b) $\frac{2}{3}$ |
| (c) $\frac{1}{9}$ | (d) $\frac{2}{9}$ |

130. If $f(x) = \frac{\frac{1}{2}e^{x/2} - e^{-x/2}}{e^x + e^{-x}}$, $x \neq 0$ then
 $\uparrow \quad \downarrow$
 $\frac{1}{2}k, \quad x=0$

- (a) f is continuous at $x=0$
- (b) f is not continuous at $x=0$ for any real k.
- (c) $\lim_{x \rightarrow 0} f(x)$ exist infinitely
- (d) None of these

131. ~~(a)~~ $\cos \frac{1}{2} \tan^{-1} \frac{\sqrt{1-x}}{\sqrt{1+x}}$

- (a) $\frac{1}{8}(x^2 - 1) + k$
- (b) $\frac{1}{2}x^2 + k$
- (c) $\frac{1}{2}x^2 + k$
- (d) None of these

132. The equation of chord of the circle $x^2 + y^2 = 8x$ bisected at the point (4, 3) is

- (a) $x = 3$
- (b) $y = 3$
- (c) $x = -3$
- (d) $y = -3$

133. x and y are positive number. Let g and a be G. M. and AM of these numbers. Also let G be G. M. of $x+1$ and $y+1$. If G and g are roots of equation $x^2 - 5x + 6 = 0$, then

- (a) $x = 2, y = \frac{3}{4}$
- (b) $x = \frac{3}{4}, y = 12$
- (c) $x = \frac{5}{2}, y = \frac{8}{5}$
- (d) $x = y = 2$

134. The co-efficient of x^n in the expansion of

$$\frac{e^{7x} + e^x}{e^{3x}}$$

(a) $\frac{4n-1+(-2)n}{n!}$ (b) $\frac{4n-1+2n}{n!}$

(c) $\frac{4n+(-2)n}{n!}$ (d) $\frac{4^{n-1}+(-2)^{n-1}}{n!}$

135. A pair of tangents are drawn from the origin to the circle $x^2 + y^2 + 20(x + y) + 20 = 0$, then the equation of the pair of tangent are

- (a) $x^2 + y^2 - 5xy = 0$
- (b) $x^2 + y^2 + 2x + y = 0$
- (c) $x^2 + y^2 - xy + 7 = 0$
- (d) $2x^2 + 2y^2 + 5xy = 0$

136. If the sum of a certain number of terms of the A.P. 25, 22, 19, is 116, then the last term is

- (a) 0
- (b) 2
- (c) 4
- (d) 6

137. If 1, a and P are in A. P. and 1, g and P are in G. P., then

(a) $\frac{1}{g} + \frac{2a}{g} + \frac{1}{P} = 0$ (b) $\frac{1}{g} + \frac{2a}{g} + \frac{1}{P} = 0$
(c) $\frac{1}{g} + \frac{2a}{g} + \frac{1}{P} = 0$ (d) $\frac{1}{g} + \frac{2a}{g} + \frac{1}{P} = 0$

138. If $y = \sin x + e^x$ then $\frac{dy}{dx}$ is equal to

- (a) $\frac{\sin x - e^x}{(\cos x + e^x)^2}$
- (b) $\frac{\sin x - e^x}{(\cos x + e^x)^3}$
- (c) $\frac{\sin x + e^x}{(\cos x - e^x)^2}$
- (d) $(-\sin x + e^x) - 1$

139. The foci of the hyperbola

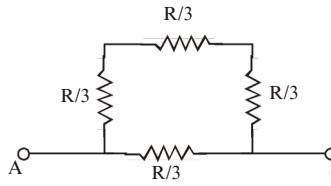
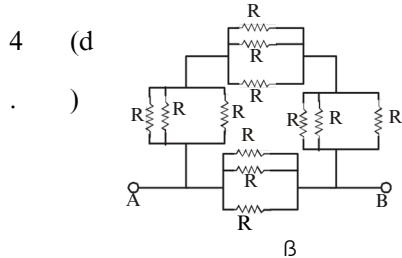
$$4x^2 - 9y^2 - 1 = 0$$

- (a) $(\pm \sqrt{13}, 0)$
- (b) $(\pm \frac{\sqrt{13}}{6}, 0)$
- (c) $(0, \pm \frac{\sqrt{13}}{6})$
- (d) None of these

SOLUTIONS

PART - I : PHYSICS

1. (c) The charged sphere is a conductor. Therefore the field inside is zero and outside it is proportional to $1/r^2$. The direction of propagation of electromagnetic wave is perpendicular to the magnetic field.
2. (c) The variation of electric field E as well as to Young's modulus of wire does not vary with dimension of wire. It is a constant quantity.
3. (b) Redraw the given circuit,



$$R_{\text{net}} \text{ between AB} = \frac{3R}{3 + \frac{3R}{R}} = \frac{R^2}{4R}$$

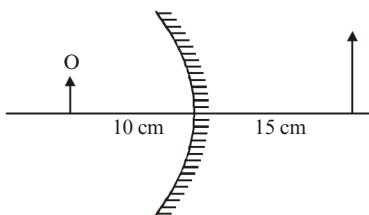
where, $R = 10\Omega$
 $R_{\text{net}} = 4\Omega$

5. (c) According to Kepler's law of planetary motion, $T \propto R^{3/2}$

$$\therefore T_2 = T_1 \frac{R_2^{3/2}}{R_1} \div$$

$$= 5 \frac{4R_u^{3/2}}{R_u} = 40 \text{ hours}$$

6. (d) $= f \frac{\alpha v + \phi}{v - \phi}$
 Here, $f = 600 \text{ Hz}, \alpha = 15 \text{ m/s}$
 $v_s = 20 \text{ m/s}, v = 340 \text{ m/s}$
 $f \phi = 600 \frac{340 + 15}{340 - 20}$
 $\therefore f = 600 \frac{355}{320} \approx 666 \text{ Hz}$
7. (c) Concave mirror is used as a shaving mirror.



From question : $v = 15 \text{ cm}, u = -10 \text{ cm}$
 Radius of curvature, $R = 2f = ?$

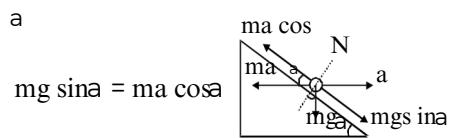
$$\text{Using mirror formula, } \frac{1}{v} + \frac{1}{u} = \frac{1}{f}$$

$$\frac{1}{15} + \frac{1}{(-10)} = \frac{1}{f} \quad \therefore f = -30 \text{ cm}$$

Therefore radius of curvature,
 $R = 2f = -60 \text{ cm}$

8. (c) From free body diagram,
 For block to remain stationary,

a

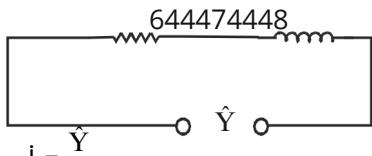


$$\therefore a = g \tan \theta$$

9. (a) On increasing the temperature, angle of contact decreases.
 10. (a) Forward bias opposes the potential barrier and if the applied voltage is more than knee voltage it cancels the potential barrier.

)

33. (d) $=n \hat{Y} = \hat{G} \otimes \text{瞳} \hat{\S} = \hat{C}W$



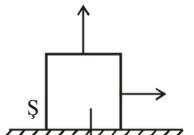
$$\begin{aligned}
 & \text{§} \circ \text{§} \sqrt{n + \omega n} \\
 & \hat{y} = \frac{\hat{G}}{\sqrt{n + (\omega)n}} = \frac{\hat{G}}{\sqrt{\hat{C} + \hat{E}^p \cdot \hat{G}An \cdot \hat{E}}} \\
 & \text{§§§§§§§§} = \frac{\hat{G}}{\sqrt{\hat{C} + (nA\hat{A})n}} \text{§}; \frac{\hat{G}}{nA\hat{A}p} \\
 & \hat{Y} \otimes \diamond \text{ 湿 } \diamond \text{ § } \diamond \text{ 盐 } \hat{i} \text{ 盐 } \hat{i} \\
 & = \frac{\hat{C}}{n} \hat{e}^{\infty} \frac{\hat{G}'\hat{G}}{nA\hat{A} n \hat{A} \hat{A} \hat{A}}
 \end{aligned}$$

34. (b) §C§T?1§C§n§§tCön§§tAöG§C§Cön§

37. (a) $\frac{\partial}{\partial x} \int_{-x}^x f(t) dt = f(x)$

38. (a) \$ 瞽 \S \quad \boxtimes \diamondsuit \quad 隊 \boxtimes \quad \S 爬 \boxtimes \diamondsuit \boxtimes \quad ? +

39. (a) $\boxtimes \S \bowtie \blacklozenge \boxdot \boxtimes$



$$\boxtimes = \frac{\hat{G}}{\star}$$

40. (a) ↗ 曙匱 ☐ 曙匱 ☐ 竜匱 ☐ 曙匱 ☐ 盐匱 ☐ 盐匱 ☐ 盐匱 ☐ 曙匱
 ↗ ZÝSOSÍÉS DÉSOSÍÍCS SPÓCSOSÉÁ

PART - II : CHEMISTRY

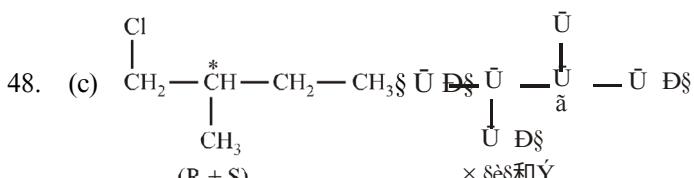
41. (a) \$熏\\$ \otimes \\$\\$ \otimes \text{盐熏} \\$ \boxtimes \text{乙} \text{盐} \otimes \text{A} \text{E} \\$ \text{SRT} \text{ Yi}

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= $\frac{D}{Dn} \cdot \frac{\bar{A} \ddot{o} \bar{A} In C n \tilde{D}}{C} \S \circ \S n \ddot{o} C$



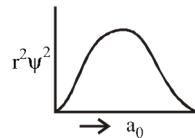
49. (b) $\bar{U}G \frac{\hat{G}}{G} \frac{\bar{U}}{U} \frac{\hat{U}}{U} \bar{U}n \frac{\hat{G}}{G} \frac{\hat{s}}{s} \frac{\bar{U}}{U} \bar{D} \bar{m}$ 4/3

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60. (a)
61. (a)
62. (a)
63. (b)
64. (a)
65. (a)
66.)
- (a)
67. (b)
68. (a)
69. (a)
70. (d)
71. (a)
72. (d)
73. (b)
74. (a)
75. (a)
76. (c)
77. (a)
78. (c)
79. (c)



PART - IV : MATHEMATICS

106. (c) $\frac{x}{s} + \frac{y}{t} = \frac{1}{s+t}$

$$\begin{aligned} & \frac{x}{a} + \frac{y}{b} = \frac{1}{a+b} \\ & \frac{a}{a+b} = \frac{1}{a} \quad \text{or} \quad \frac{b}{a+b} = \frac{1}{b} \\ & \frac{a}{a+b} = \frac{1}{a} \quad \text{or} \quad \frac{b}{a+b} = \frac{1}{b} \\ & \frac{a}{a+b} = \frac{1}{a} \quad \text{or} \quad \frac{b}{a+b} = \frac{1}{b} \end{aligned}$$

$\frac{n}{s} + \frac{m}{t} = \frac{1}{s+t}$

107. (c) $\frac{1}{n} + \frac{1}{m} = \frac{1}{n+m}$

$$\frac{C}{\sqrt{C+n}} = \frac{\sqrt{C}}{\sqrt{C+n}} + \frac{\sqrt{n}}{\sqrt{C+n}}$$

和 $\frac{C}{\sqrt{C+n}}$

$$\frac{C}{\sqrt{n} + \sqrt{D}} = \frac{\sqrt{n} - \sqrt{D}}{\sqrt{n} + \sqrt{D}} = \frac{C}{\sqrt{n} + \sqrt{D}}$$

$$\sqrt{\frac{C}{\sqrt{C+n}}} + \sqrt{\frac{C}{\sqrt{n} + \sqrt{D}}} + \sqrt{\frac{C}{\sqrt{n} + \sqrt{D}}} = \sqrt{\frac{C}{\sqrt{C+n}}} + \sqrt{\frac{C}{\sqrt{n} + \sqrt{D}}} + \sqrt{\frac{C}{\sqrt{n} + \sqrt{D}}}$$

$$= \frac{\sqrt{C} - \sqrt{n}}{\sqrt{C+n}} = \frac{\sqrt{C} - \sqrt{n}}{\sqrt{C+n}}$$

$$= \frac{\sqrt{C} - \sqrt{n}}{\sqrt{C+n}} = \frac{\sqrt{C} - \sqrt{n}}{\sqrt{C+n}}$$

和 $\frac{C}{\sqrt{C+n}}$

$$= -\frac{\frac{C}{\sqrt{C+n}} - \frac{C}{\sqrt{C+n}}}{\sqrt{C+n} + \sqrt{C+n}} = \frac{-C}{\sqrt{C+n} + \sqrt{C+n}}$$

108. (b) $\frac{1}{s+t} = \frac{1}{s} + \frac{1}{t}$

$$\begin{aligned} & \frac{1}{s+t} = \frac{1}{s} + \frac{1}{t} \\ & \frac{1}{s+t} = \frac{s+t}{st} = \frac{1}{s} + \frac{1}{t} \\ & \frac{1}{s+t} = \frac{s+t}{st} = \frac{1}{s} + \frac{1}{t} \\ & \frac{1}{s+t} = \frac{s+t}{st} = \frac{1}{s} + \frac{1}{t} \end{aligned}$$

$\frac{1}{s+t} = \frac{1}{s} + \frac{1}{t}$

$\frac{1}{s+t} = \frac{1}{s} + \frac{1}{t}$

$\frac{1}{s+t} = \frac{1}{s} + \frac{1}{t}$

109. (c) $\frac{1}{s+t} = \frac{1}{s} + \frac{1}{t}$

110. (c) $\frac{1}{s+t} = \frac{1}{s} + \frac{1}{t}$

$s+t = s+t$

