VITEEE 2007 Question Paper

Vellore Institute of Technology Engineering Entrance Examination

SOLVED PAPER

VITE EE 2007

PART - I (PHYSICS)

1. The magnetic moment of the ground state of an atom whose open sub shell is half filled with five electrons is

   (a) $\sqrt{35}\ \sqrt{\mu B}$ (b) $35\ \mu_B$

   (c) $35\sqrt{\mu B}$ (d) $\mu_B \sqrt{35}$

2. Indicate which one of the following statements is NOT CORRECT ?

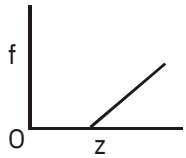
   (a) Intensities of reflections from different crystallographic planes are equal

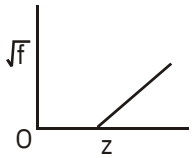
   (b) According to Bragg's law higher order of reflections have high values for a given wavelength of radiation

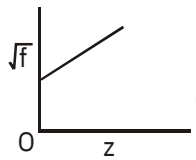
   (c) For a given wavelength of radiation there is a smallest distance between the crystallographic planes which can be determined

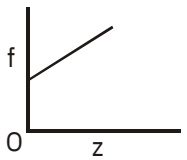
   (d) Bragg's law may predict a reflection from a crystallographic plane to be present but it may be absent due to the crystal symmetry

3. Identify the graph which correctly represents the Moseley's law

   (a) 

   (b) 

   (c) 

   (d) 

4. Assuming f to be the frequency of first line in Balmer series, the frequency of the immediate next (i.e. second) line is

   (a) 0.50 f (b) 1.35 f

   (c) 2.05 f (d) 2.70 f

5. The velocity of a particle at which the kinetic energy is equal to its rest energy is

   (a) $\frac{3c}{2}$ (b) $3\frac{c}{\sqrt{2}}$

   (c) $\frac{3c}{2}^{\frac{1}{2}}$ (d) $\frac{c\sqrt{3}}{2}$

6. One electron and one proton is accelerated by equal potential. Ratio in their deBroglie wavelength is

   (a) 1 (b) $\frac{m}{e}$

   (c) $\frac{mp}{m_e}$ (d) $\frac{m}{p\sqrt{\frac{me}{m_p}}}$

7. Two electrons one moving in opposite direction with speeds 0.8c and 0.4c where c is the speed of light in vacuum. Then the relative speed is about

   (a) 0.4c (b) 0.8c

   (c) 0.9c (d) 1.2 c

8. A photo-sensitive material would emit electrons if excited by photons beyond a threshold. To overcome the threshold, one would increase

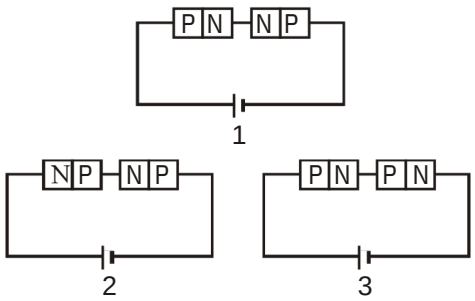
   (a) the voltage applied to the light source

   (b) the intensity of light

   (c) the wavelength of light

   (d) the frequency of light

9. The radius of nucleus is
   (a) proportional to its mass number
   (b) inversely proportional to its mass number
   (c) proportional to the cube root of its mass number
   (d) not related to its mass number
10. Radio carbon dating is done by estimating in specimen
   (a) the amount of ordinary carbon still present
   (b) the amount of radio carbon still present
   (c) the ratio of amount of 14C to 12C still pr esen t
   (d) the ratio of amount of 12C to 14C still pr esen t
11. Ionization power and penetration range of radioactive radiation increases in the order
   (a) and respectively
   (b) and respectively
   (c) and respectively
   (d) and respectively
12. The half life of a radioactive element is 3.8 days. The fraction left after 19 days will be
   (a) 0.124 (b) 0.062
   (c) 0.093 (d) 0.031
13. Two identical P-N junctions are connected in series in three different ways as shown below to a battery. The potential drop across the P-N junctions are equal in

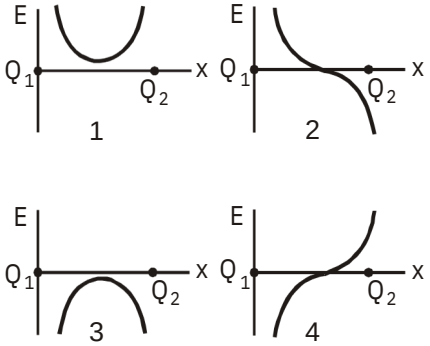


   (a) in circuits 2 and 3 in
   (b) circuits 1 and 2 in
   (c) circuits 1 and 3 in
   (d) none of the circuit
14. The temperature coefficient of a zener mechanism is
   (a) negative (b) positive
   (c) infinity (d) zero
15. Identify the logic gate from the following TRUTH table

Inputs		Output
A	B	Y
0	0	1
0	1	1
1	0	1
1	1	0

In Boolean algebra, AB
   (a) NOR gate (b) NOT gate
   (c) AND gate (d) NAND gate
16. $\overline{\overline{A}\,\overline{B}}$ is equal to
   (a) $\overline{A}\,\overline{B}$ (b) $A\,B$
   (c) A.B (d) A + B
17. Radar waves are sent towards a moving airplane and the reflected waves are received. When the airplane is moving towards the radar, the wavelength of the wave
   (a) decrease
   (b) increase
   (c) remains the same
   (d) sometimes increase or decrease
18. The transmission of high frequencies in a coaxial cable is determined by
   (a) $\frac{1}{(LC)^{1/2}}$ where L and C are inductance and capacitance
   (b) $(LC)^2$
   (c) the impedance L alone
   (d) the dielectric and skin effiect
19. The output stage of a television transmitter is most likely to be a
   (a) plate-modulated class C amplifier
   (b) grid-modulated class C amplifier
   (c) screen-modulated class C amplifier
   (d) grid-modulated class A amplifier
20. The antenna current of an AM transmitter is 8A when only the carrier is sent, but it increases to 8.93A when the carrier is modulated by a single sine wave. Find the percentage modulation.
   (a) 60.1 % (b) 70.1%
   (c) 80.1% (d) 50.1%

21. Two point like charges Q1 and Q2 of whose strength are equal in absolute value are placed at a certain distance from each other. Assuming the field strength to be posiive in the positive direction of x-axis the signs of the charges Q1 and Q2 for the graphs (field strength versus distance) shown in Figures 1,2,3 and 4 are



(a) Q1 positive, Q2 negative; both positive; Q1 negative, Q2 positive; both negative

(b) Q1 negative, Q2 positive;Q1 positive, Q2 negative; both positive; both negative

(c) Q1 positive, Q2 negative; both negative; Q1 negative, Q2 positive; both positive

(d) Both positive; Q1 positive, Q2 negative; Q1 negative, Q2 positive; both negative

22. ABCD is a rectangle. At corners B, C and D of the rectangle are placed charges + 10 × 10–12C, –20×10–12C and 10 × 10–12C respectively. Calculate the potential at the fourth corner. The side AB = 4cm and BC = 3cm

(a) 1.65 V (b) 0.165 V

(c) 16.5 V (d) 2.65 V

23. A parallel plate capacitor of capacitance 100 pF is to be constructed by using paper sheets of 1 mm thickness as dielectric. If the dielectric constant of paper is 4, the number of circular metal foils of diameter 2 cm each required for the purpose is

(a) 40 (b) 20

(c) 30 (d) 10

24. Thee lectric field intensity E, due to an electric

dipole of moment p, at a point on the equatorial line is

(a) parallel to the axis of the dipole and opposite to the direction of the dipole moment p

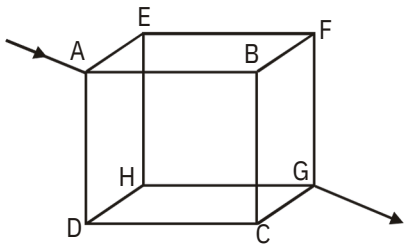
(b) perpendicular to the axis of the dipole and

(c) is directed away from it

(d) parallel to the dipole moment

perpendicular to the axis of the dipole and is directed toward it

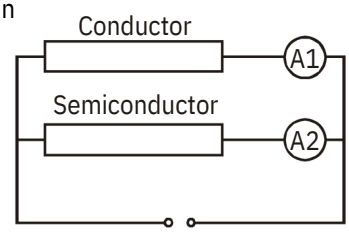
25. Twelve wires of each of resistance 6 ohms are connected to form a cube as shown in the figure. The current enters at a corner A and leaves at the diagonally opposite corner G. The joint resistance across the corners A and G are



(a) 12 ohms (b) 6 ohms

(c) 3 ohms (d) 5 ohms

26. A conductor and a semi-conductor are connected in parallel as shown in the figure. At a certain voltage both ammeters registers the same current. If the voltage of the DC source is increased then



(a) the ammeter connected to the semiconductor will register higher current than the ammeter connected to the conductor

(b) the ammeter connected to the conductor will register higher current than the ammeter connected to the semiconductor

(c) the ammeters connected to both semiconductor and conductor will register the same current

(d) the ammeter connected to both semiconductor and conductor will register no change in the current

27. A uniform copper wire of length 1m and cross-sectional area 5×10–7m2 carries a current of 1A . Assuming that there are 8×1028 free electrons/ m3 in copper, how long will an electron take to drift from one end of the wire to the other

(a) 0.8×103s (b) 1.6×103s

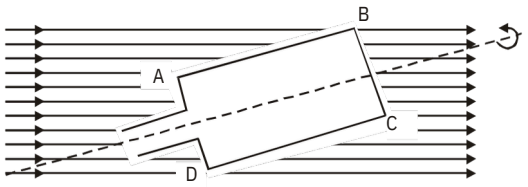
(c) 3.2×103s (d) 6.4×103s

28. The temperature coefficient of resistance of a wire is 0.00125/K. At 300K its resistance is 1 .The resistance of the wire will be 2 at

(a) 1154 K (b) 1100 K

(c) 1400 K (d) 1127 K

29. A rectangular coil ABCD which is rotated at a constant angular velocity about an horizontal as shown in the figure. The axis of rotation of the coil as well as the magnetic field B are horizontal. Maximum current will flow in the circuit when the plane of the coil is



(a) inclined at 30 degrees to the magnetic field

(b) perpendicular to the magnetic field

(c) inclined at 45 degrees to the magnetic field

(d) parallel to the magnetic field

30. If the total emf in a thermocouple is a parabolic function expressed as $E = at + \frac{1}{2}bt2$, which of the following relations does not hold good

(a) neutral temperature $t_n = -a/b$

(b) temperature of inversion $t_i = -2a/b$

(c) thermo-electric power $p = a + bt$

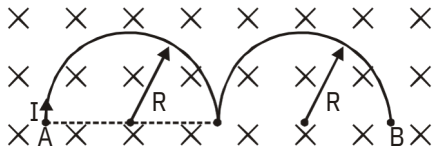
(d) $t_n = a/b$

31. The proton of energy 1 MeV describes a circular path in plane at right angles to a uniform magnetic field of 6.28 ×10–4T. The mass of the proton is 1.7 × 10–27 Kg.The cyclotron frequency of the proton is very nearly equal to

(a) 107 Hz (b) 105 Hz

(c) 106 Hz (d) 104 Hz

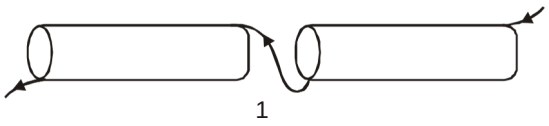
32. A wire AB, in the shape of two semicircular segments of radius R each and carrying a current I, is placed in a uniform magnetic field B directed into the page (see figure). The magnitude of the force due to the field B on the wire AB is



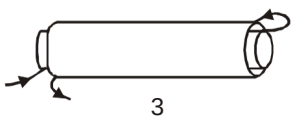
(a) zero (b) 4 RIB

(c) R2 I B (d) 2 R I B

33. There are two solenoids of same length and inductance L but their diameters differ to the extent that one can just fit into the other. They are connected in three different ways in series. 1) They are connected in series but separated by large distance 2) they connected in series with one inside the other and senses of the turns coinciding 3) both are connected in series with one inside the other with senses of the turns opposite as depicted in figures 1,2 and 3 respectively. The total inductance of the solenoids in each of the case 1, 2 and 3 are respectively


1


2

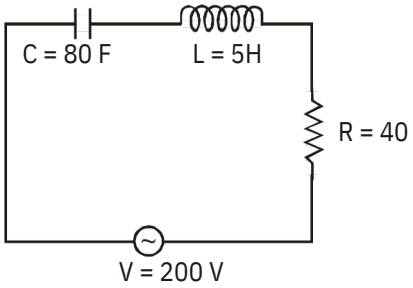

3

(a) 0, $4L_0$ $2L_0$ (b) $4L_0$ $2L_0$ 0,

(c) $2L_0$ 0, $4L_0$ (d) 2 L_0 $4L_0$, 0

34. From figure shown below a series LCR circuit connected to a variable frequency 200V source.

L = 5H, C = 80 F and R = 40. Then the source frequency which drive the circuit at resonance is



(a) 25 Hz (b) $\frac{25}{}$ Hz

(c) 50 Hz (d) $\frac{50}{}$ Hz

35. If the coefficient of mutual induction of the primary and secondary coils of an induction coil is 5H and a current of 10A is cut off in 5×10–4 second, the *emf* induced(in volt) in the secondary coil is

(a) 5×104 (b) 1×105
(c) 25×105 (d) 5×106

36. A voltage of peak value 283 V and varying frequency is applied to a series L, C, R combination in which R = 3 ohm, L=25 mH and C= 400F. The frequency (in Hz) of the source at which maximum power is dissipated in the above is

(a) 51.5 (b) 50.7
(c) 51.1 (d) 50.3

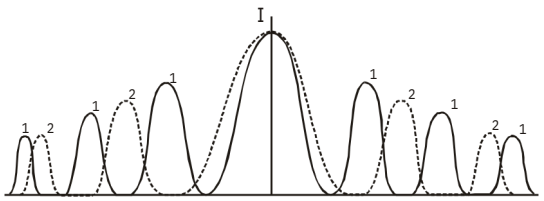
37. Four independent waves are represented by equ at i on s

(1) $X_1 = a\sin_1 t$ (3) $X_2 = a\sin 2t$
(2) $X_3 = a\sin_1 t$ (4) $X_4 = a\sin_1(t+)$

Interference is possible between waves represented by equations

(a) 3 and 4 (b) 1 and 2
(c) 2 and 3 (d) 1 and 4

38. Following diffraction pattern was obtained using a diffraction grating using two different wavelengths and . With the help of the figure identify which is the longer wavelength and their ratios.



(a) is longer than and the ratio of the
(b) longer to the shorter wavelength is 1.5
(c) is longer than and the ratio of the
(d) longer to the shorter wavelength is 1.5
and are equal and their ratio is 1.0
is longer than and the ratio of the longer to the shorter wavelength is 2.5

39. In Young's double slit experiment, the interference pattern is found to have an intensity ratio between bright and dark fringes is 9. This implies the (a) (b) (c) (d)
the intensities at the screen due to two slits are 5 units and 4 units respectively
the intensities at the screen due to the two slits are 4 units and 1 units respectively the amplitude ratio is 7 the amplitude ratio is 6

40. Rising and setting sun appears to be reddish because
(a) Diffraction sends red rays to earth at these
(b) times
(c) Scattering due to dust particles and air
(d) molecules are responsible
Refraction is responsible
Polarization is responsible

PART - II (CHEMISTRY)

41. The catalyst used in Rosenmund reaction is

(a) Zn/Hg (b) Pd/BaSO4
(c) Raney Ni (d) Na in Ethanol

42. $CH_2CO_2 \quad RMX \xrightarrow{H} ?$

(a) $ROOC(CH_2)COOR$
(b) $RCOCH_2CH_2COOH$
(c) RCO OR
(d) RC OO H

43. Identify, which of the below does not possess any element of symmetry?
(a) (+)– Tartaric acid
(b) Carbon tetrachloride
(c) Methane
(d)Mesotartatric acid

44. The weakest acid amongst the following is
(a) $ClCH_2COOH$ (b) HCOOH
(c) FCH_2CH_2COOH (d) $CH_3(I)CO_2OH$

45. $HOOC\ CH_2\ _4\ COOH\ 2C\ H_2OH_5$

$\xrightarrow[\text{Toluene}]{H_2SO_4}\ C_2H_5OOC\ CH_2\ COOC_2H_5$

The purpose of using toluene here is
(a) to make both substances (acid & alcohol) miscible
(b) that the product is insoluble in toluene
(c) that the reactants are insoluble in water
(d) because of the formation of low boiling azotrope

46. Trans esterification is the process of
(a) conversion of an aliphatic acid to ester
(b) conversion of an aromatic acid to ester
(c) conversion of one ester to another ester
(d) conversion of an ester into its components namely acid and alcohol

47. The correct sequence of base strengths in aqueous solution is
(a) $(CH_3)_2NH > CH_3NH_2 > (CH_3)_3N$
(b) $(CH_3)_3N > (CH_3)_2NH > CH_3NH_2$
(c) $(CH_3)_3N > CH_3NH_2 = (CH_3)_2NH$
(d) $(CH_3)_2NH > (CH_3)_3N > CH_3NH_2$

48. When aqueous solution of benzene diazoniumchloride is boiled, the product formed is (a) $C_6H_5CH_2OH$
(b) $C_6H_6N_2$
(c) C_6H_5COOH (d) C_6H_5OH

49. Carbylamine reaction is given by aliphatic
(a) primary amine
(b) secondary amine
(c) tertiary amine
(d) quaternary ammonium salt

50. $C_6H_5CHO\ \xrightarrow[H_2,Ni]{NH_3}\ ?$
(a) $C_6H_5NH_2$
(b) $C_6H_5NHCH_3$
(c) $C_6H_5CH_2NH_2$ (d) $C_6H_5NHC_6H_5$

51. In $TeCl_4$ the central atom tellurium involves
(a) sp3 hybridization
(b) sp3d hybridization
(c) sp3d2 hybridization
(d) dsp2 hybridization

52. The purple colour of $KMnO_4$ is due to the transition
(a) C.T. (L M) (b)C.T. (M L)
(c) d – d (d) p –d

53. A nuclear reaction of $_{92}^{235}U$ with a neutron produces $_{36}^{90}Kr$ and two neutrons. Other element produced in this reaction is
(a) $_{52}^{137}Te$ (b) $_{55}^{144}Cs$
(c) $_{56}^{137}Ba$ (d) $_{56}^{144}Ba$

54. AgCl dissolves in a solution of NH_3 but not in water because
(a) NH_3 is a better solvent than H_2O
(b) Ag+ forms a complex ion with NH_3
(c) NH_3 is a stronger base than H_2O
(d) the dipole moment of water is higher than NH_3

55. Which of the following is hexadenate ligand?
(a) Ethylene diamine
(b) Ethylene diamine tetra acetic acid
(c) 1, 10- phenanthroline
(d) Acetyl acetonato

56. A coordinate bond is a dative covalent bond. Which of the below is true?
(a) Three atoms form bond by sharing their
(b) electrons
(c) Two atoms form bond by sharing their electrons
Two atoms form bond and one of them provides both electrons
(d) Two atoms form bond by sharing electrons obtained from third atom

57. Which of the following complex has zero magnetic moment (spin only)?
(a) [Ni(NH3)6]Cl2 (b) Na3[FeF6]
(c) [Cr(H2O)6]SO4 (d) K4[Fe(CN)6]

58. The IUPAC name of $[Ni(PPh_3)_2Cl_2]$ is
(a) bis dichloro (triphenylphosphine) nickel
(b) (II) dichloro bis (triphenylphosphine)
(c) nickel (II) dichloro triphenylphosphine
(d) nickel (II) triphenyl phosphine nickel (II) dichloride

59. Among the following the compound that is both paramagnetic and coloured is
(a) $K_2Cr_2O_7$ (b) $(NH_4)_2[TiCl_6]$
(c) $VOSO_4$ (d) $K_3Cu(CN)_4$

60. On an X-ray diffraction photograph the intensity of the spots depends on
(a) neutron density of the atoms/ions
(b) electron density of the atoms/ions
(c) proton density of the atoms/ions
(d) photon density of the atoms/ions

61. An ion leaves its regular site occupy a position in the space between the lattice sites is called
(a) Frenkel defect (b) Schottky defect
(c) Impurity defect (d) Vacancy defect

62. The 8:8 type of packing is present in
(a) MgF_2 (b) CsCl
(c) KCl (d) NaCl

63. When a solid melts reversibly
(a) H decreases (b) G
(c) E decreases (d) increases S

64. Enthalpy is equal to (a) increases

(a) $-T^2\left(\frac{G}{T}\right)_V$ (b) $-T2\left(\frac{G/T}{T}\right)_P$

(c) $T^2\left(\frac{G/T}{T}\right)_V$ (d) $2T\left(\frac{G}{T}\right)_P$

65. Condition for spontaneity in an isothermal process is
(a) AW 0 (b) G U 0
(c) A U 0 (d) G–U 0

66. Given:2C s 2O2 g 2CO2 g ;

H –787kJ

$H_2(g) + \frac{1}{2}O2(g) \rightarrow H2O(\ ); \quad H = -286kJ$

$C2H2(g) + 2\frac{1}{2}O2(g) \rightarrow 2CO2(g) + H2O(\ )$

; H –1310kJ

The heat of formation of acetylene is
(a)–1802 kJ(b)+1802 kJ
(c)+237 kJ(d)–800 kJ

67. Given the equilibrium system:
$NH_4Cl(s) \rightleftharpoons NH_4^+(aq) + Cl^-(aq)$
(H = +3.5kcal/mol).
What change will shift the equilibrium to the right?
(a) Decreasing the temperature
(b) Increasing the temperature
(c) Dissolving NaCl crystals in the equilibrium mixture
(d) Dissolving $NHNO_3$ crystals in the equilibrium mixture

68. According to Arrhenius equation, the rate constant (k) is related to temperature (T) as

(a) $\ln\frac{k_2}{k1} \quad \frac{E_a}{R}\ \frac{1}{T_1} - \frac{1}{T_2}$

(b) $\ln\frac{k_2}{k1} \quad -\frac{E_a}{R}\ \frac{1}{T1} - \frac{1}{T2}$

(c) $\ln\frac{k_2}{k1} \quad \frac{Ea}{R}\ \frac{1}{T1}\ \frac{1}{T2}$

(d) $\ln\frac{k_2}{k_1} \quad -\frac{E_a}{R}\ \frac{1}{T_1}\ \frac{1}{T_2}$

69. Equivalent amounts of H_2 and I_2 are heated in a closed vessel till equilibrium is obtained. If 80%
of the hydrogen can be converted to HI, the Kc at this temperature is
(a) 64 (b) 16
(c) 0.25 (d) 4

70. For the reaction $H_2(g)+I_2(g) \rightleftharpoons 2HI(g)$, the equilibrium constant Kp changes with
(a) total pressure
(b) catalyst
(c) the amount H_2 and I_2
(d) temperature

71. How long (in hours) must a current of 5.0 amperes be maintained to electroplate 60g of calcium from molten $CaCl_2$?
(a) 27 hours (b) 8.3 hours
(c) 11 hours (d) 16 hours

72. For strong electrolytes the plot of molar conductance vs $\sqrt{C}$is
(a) parabolic (b) linear
(c) sinusoidal (d) circular

73. If the molar conductance values of Ca2+ and Cl– at infinite dilution are respectively 118.88×10–4 m2 mho mol–1 and 77.33×10–4 m2 mho mol–1 then that of CaCl is (in m2 mho mol–1)
(a) 118.88 × 10–4 (b) 154.66 × 10–4
(c) 273.54 × 10–4 (d) 196.21× 10–4

74. The standard reduction potentials at 298K for the following half reactions are given against each

$Zn^{2+}(aq) + 2e^- \rightarrow Zn(s)$ $E^0 = -0.762$ V
$Cr^{3+}(aq) + 3e^- \rightarrow Cr(s)$ $E^0 = -0.740$ V
$2H^+(aq) + 2e^- \rightarrow H_2(g)$ $E^0 = 0.00$ V
$Fe^{3+}(aq) + 3e^- \rightarrow Fe^{2+}(aq)$ $E^0 = +0.762$ V

The strongest reducing agent is
(a) Zn(s) (b) Cr(s)
(c) H2(g) (d) Fe2+(aq)

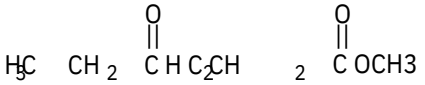
75. The epoxide ring consists of which of the following ?
(a) Three membered ring with two carbon and one oxygen
(b) Four membered ring with three carbon and one oxygen
(c) Five membered ring with four carbon and one oxygen
(d) Six membered ring with five carbon and one oxygen

76. In the Grignard reaction, which metal forms an organometallic bond?
(a) Sodium (b) Titanium
(c) Magnesium (d) Palladium

77. Phenol is less acidic than
(a) p-chlorophenol
(b) p-nitrophenol
(c) p-methoxyphenol
(d) ethanol

78. Aldol condensation is given by
(a) trimethylacetaldehyde
(b) acetaldehyde
(c) benzaldehyde
(d) formaldehyde

79. Give the IUPAC name for



(a) Ethyl-4-oxoheptonate
(b) Methyl-4-oxoheptonate
(c) Ethyl-4-oxohexonate
(d) Methyl-4-oxohexonate

80. In which of the below reaction do we find , -unsaturated carbonyl compounds undergoing a ring closure reaction with conjugated dienes?
(a) Perkin reaction
(b) Diels-Alder reaction
(c) Claisen rearrangement
(d) Hoffman reaction

PART - III (MATHEMATICS)

81. Let the pairsa,b andc, d each determine a plane. Then the planes are parallel if

(a) a c b d 0

(b) a c b d 0

(c) a b c d 0

(d) a b c d 0

82. The area of a parallelogram with $3\hat{i}\hat{j}-2\hat{k}$ and $4\hat{k}\hat{i}-3\hat{j}$ as diagonals is

(a) $\sqrt{72}$ (b) $\sqrt{73}$
(c) $\sqrt{74}$ (d) $\sqrt{75}$

83. If cosx + cos2 x =1then the value of sin12 x+ 3sin10 x + 3sin8x + sin6 x–1 is equal to
(a) (c) The product of(b) all values
of (cos 1 (d) 0

84. equal to +i sin)3/5 is

(a) 1 (b) cos +i sin
(c) cos3 +i sin3 (d) cos5 +i sin5

85. The imaginary part of $\frac{1 \; i^{\;2}}{i\;2i-1}$ is

(a) $\frac{4}{5}$ (b) 0
(c) $\frac{2}{5}$ (d) $-\frac{4}{5}$

86. If $\sin^{-1}x + \sin^{-1}y = \frac{}{2}$, then $\cos^{-1}x + \cos^{-1}y$ is equal to

(a) $\frac{}{2}$ (b) $\frac{}{4}$

(c) (d) $\frac{3}{4}$

87. The equation of a directrix of the ellipse $\frac{x^2}{16} + \frac{y^2}{25} = 1$ is

(a) $3y = 5$ (b) $3y = 25$ (c) $y = 5$ (d) $y = 3$

88. If the normal at (ap2, 2ap) on the parabola y2 = 4ax, meets the parabola again at (aq2, 2aq), then

(a) p2 + pq + 2 = 0 (b) p2 – pq + 2 = 0
(c) q2 + pq + 2 = 0 (d) p2 + pq + 1 = 0

89. The length of the straight line x – 3y = 1 intercepted by the hyperbola x2 – 4y2 = 1is

(a) $\sqrt{10}$ (b) $\frac{6}{5}$

(c) $\frac{1}{\sqrt{10}}$ (d) $\frac{6}{5}\sqrt{10}$

90. The curve described parametrically by x = t2 + 2t–1, y = 3t + 5 represents

(a) an ellipse (b) a hyperbola
(c) a parabola (d) a circle

91. If the normal to the curve y = f(x) at (3,4) makes an angle $\frac{3}{4}$ with the positive x-axis, then f '(3) is equal to

(a) – 1 (b) $\frac{3}{4}$

(c) 1 (d) $-\frac{3}{4}$

92. The function $f(x) = x2\, e^{-2x}$, x > 0. Then the maximum value of f(x) is

(a) $\frac{1}{e}$ (b) $\frac{1}{2e}$

(c) $\frac{1}{e2}$ (d) $\frac{4}{e4}$

93. If $(x + y)\sin u = x2y2$, then $x\frac{u}{x} \quad y\frac{u}{y}$

(a) sin u (b) cosec u
(c) 2 tan u (d) tan u

94. The angle between the tangents at those points on the curve x = t2 + 1 and y = t2 – t – 6 where it meets x-axis is

(a) $\tan^{-1}\frac{4}{29}$ (b) $\tan^{-1}\frac{5}{49}$

(c) $\tan^{-1}\frac{10}{49}$ (d) $\tan^{-1}\frac{8}{29}$

95. The value of $\int_1^4 |x-3|\,dx$ is equal to

(a) 2 (b) $\frac{5}{2}$

(c) $\frac{1}{2}$ (d) $\frac{3}{2}$

96. The area of the region bounded by the straight lines x = 0 and x = 2 and the curves y = 2x and y = 2x –x2 is equal to

(a) $\frac{2}{\log 2} - \frac{4}{3}$ (b) $\frac{3}{\log 2} - \frac{4}{3}$

(c) $\frac{1}{\log 2} - \frac{4}{3}$ (d) $\frac{4}{\log 2} - \frac{3}{2}$

97. The value of $\int_0 \frac{dx}{a^2 \quad x^{2\ 7}}$ is equal to

(a) $\frac{231}{2047}\ \frac{1}{a13}$ (b) $\frac{2351}{2048}\ a^{\frac{}{13}}$

(c) $\frac{232}{2047}\ \frac{1}{a13}$ (d) $\frac{2311}{2048}\ a^{\frac{}{13}}$

98. The value of the integral $e^x \left(\frac{1-x}{1+x^2}\right)^2 dx$ is

(a) $e^x \frac{1-x}{1+x^2} + C$

(b) $e^x \frac{1+x}{1+x^2} + C$

(c) $\frac{e^x}{1+x^2} + C$

(d) $e^x(1-x) + C$

99. If $x\sin\frac{y}{x}\,dy = y\sin\frac{y}{x} - x\,dx$

and $y(1) = \frac{\pi}{2}$, then the value of $\cos\frac{y}{x}$ is equal to

(a) x (b) $\frac{1}{x}$

(c) $\log x$ (d) e^x

100. The differential equation of the system of all circles of radius r in the XY plane is

(a) $\left(1+\left(\frac{dy}{dx}\right)^2\right)^3 = r^2\left(\frac{d^2y}{dx^2}\right)^2$

(b) $\left(1+\left(\frac{dy}{dx}\right)^2\right)^3 = r^2\left(\frac{d^2y}{dx^2}\right)^3$

(c) $\left(1+\left(\frac{dy}{dx}\right)^2\right)^3 = r^2\left(\frac{d^2y}{dx^2}\right)^2$

(d) $\left(1+\left(\frac{dy}{dx}\right)^2\right)^3 = r^2\left(\frac{d^2y}{dx^2}\right)^3$

101. The general solution of the differential equation $\frac{d^2y}{dx^2} - 2\frac{dy}{dx} + y = 2e^{3x}$ is given by

(a) $y = c_1 + c_2x\, e^x + \frac{e^{3x}}{8}$

(b) $y = c_1 + c_2x\, e^{-x} + \frac{e^{-3x}}{8}$

(c) $y = c_1 + c_2x\, e^{-x} + \frac{e^{3x}}{8}$

(d) $y = c_1 + c_2x\, e^x + \frac{e^{-3x}}{8}$

102. The solution of the differential equation $ydx+(x-y^3)dy = 0$ is

(a) $xy = \frac{1}{3}y^3 + C$ (b) $xy = y^4 + C$

(c) $y^4 = 4xy + C$ (d) $4y = y^3 + C$

103. The number of positive integral solutions of the equation $x_1x_2x_3x_4x_5 = 1050$ is

(a) 1870 (b) 1875

(c) 1865 (d) 1880

104. Let A= {1,2,3,....., n}and B ={a,b,c}, then the number of functions from A to B that are onto is

(a) $3^n - 2^n$ (b) $3^n - 2^n - 1$

(c) $3(2^n - 1)$ (d) $3^n - 3(2^n - 1)$

105. Everybody in a room shakes hands with everybody else. The total number of hand shakes is 66. The total number of persons in the room is

(a) 9 (b) 12

(c) 10 (d) 14

106. If(G,*) is a group and the order of an element aG is 10, then the order of the inverse of a* a is

(a) 10 (b) $\frac{1}{10}$

(c) 5 (d) $\frac{1}{5}$

107. A box contains 9 tickets numbered 1 to 9 inclusive. If 3 tickets are drawn from the box one at a time, the probability that they are alternatively either {odd, even, odd} or {even, odd, even} is

(a) $\frac{5}{17}$ (b) $\frac{4}{17}$

(c) $\frac{5}{16}$ (d) $\frac{5}{18}$

108. If P(A) = $\frac{1}{12}$, P(B) = $\frac{5}{12}$ and P(B/A)= $\frac{1}{15}$ then p A B is equal to

(a) $\frac{89}{180}$ (b) $\frac{90}{180}$

(c) $\frac{91}{180}$ (d) $\frac{92}{180}$

109. If the probability density function of a random variable X is f(x) = $\frac{x}{2}$ in 0 x 2, then P(X > 1.5 X > 1) is equal to

(a) $\frac{7}{16}$ (b) $\frac{3}{4}$

(c) $\frac{7}{12}$ (d) $\frac{2}{1}$

110. If X is a poisson variate such that 2P(X = 0)+ P(X = 2) = 2P(X = 1) then E(X) is equal to
(a) 1 (b) 2
(c) 1.5 (d) 1.75

111. If A $\begin{matrix} 1 & \tan \\ -\tan & 1 \end{matrix}$ and AB = I, then (cos2)B is equal to

(a) A( ) (b) A $\frac{}{2}$

(c) A(–) (d) A $\frac{-}{2}$

112. If x = –5 is a root of $\begin{vmatrix} 2x & 1 & 4 & 8 \\ 2 & & 2x & 2 \\ 7 & & 6 & 2x \end{vmatrix}$ = 0, then the other roots are
(a 3, 3.5 (b) 1, 3.5
) 1, 7 (d) 2, 7

113. The simultaneous equations Kx + 2y–z =1, (K –1)y–2z = 2 and (K + 2)z = 3 have only one solution when
(a) K = –2 (b) K = –1
(c) K = 0 (d) K = 1

114. If the rank of the matrix $\begin{matrix} -1 & 2 & 5 \\ 2 & -4 & a-4 \\ 1 & -2 & a \end{matrix}$ is 1, then the value of a is
(a) –1 (b) 2
(c) –6 (d) 4

115. If 4ac for the equation ax4 + bx2 + c = 0, then b2 all the roots of the equation will be real if
(a) b>0, a<0, c>0 (b) b<0, a>0, c>0

(c) b>0, a>0, c>0 (d) b>0, a>0, c<0

116. If x > 0 and $\log x_3$ + log($\sqrt{x}$) + log3($\sqrt[4]{x}$)+ $\log_3 \sqrt[8]{x}$ + $\log_3 \sqrt[16]{x}$ +....= 4, then x equals
(a) 9 (b) 81
(c) 1 (d) 27

117. The number of real roots of the equation $\begin{matrix} x & 1 & 3 \\ -x & x & \frac{1}{x} \\ 0 & & \end{matrix}$ 0 is
(a) 0 (b) 2
(c) 4 (d) 6

118. If H is the harmonic mean between P and Q, then the value of $\frac{H}{P}$ $\frac{H}{Q}$ is

(a) 2 (b) $\frac{PQ}{PQ}$

(c) $\frac{1}{2}$ (d) $\frac{P \quad Q}{PQ}$

119. If aand bare two unit vectors, then the vector (a+ b) × (a×b) is parallel to the vector

(a) a +b (b) 2a +b

(c) a –b (d) 2a –b

120. If is the angle between the lines AB and AC where A, B and C are the three points with coordinates (1,2,–1), (2,0,3), (3,–1,2) respectively, then $\sqrt{462}\cos$ is equal to
(a) 20 (b) 10
(c) 30 (d) 40

2007 SOLUTIONS

PART - I (PHYSICS)

1. (d) The magnetic moment of the ground state of an atom whose open sub-shell is half filled with n electrons is given by

$\sqrt{n(n \quad 2)}.\ _B$

where $_B$ is the gyromagnetic moment of the atom.

Here, n =

5. 5(5 $\sqrt{\quad 2)}.\ _B$ $\sqrt{35}\ _B$

2. (a) Bragg's law gives 2dsin = n, n= order of reflection, d= distance between planes. For same and d, n can be found. If crystal is symmetric reflections from different planes may cancel out.

3. (b)According to Moseley's law, square root of frequency of X-ray is plotted against atomic number it gives straight line, the relation is

$\sqrt{f}$ cZ–1where c = constant

(for $\sqrt{f}$ 0,Z c, for Z = 0, $\sqrt{f}$ –c)

Option (b) is correct. as Z can not be n egative.

4. (b) Balmer series is given for n 1 = 2 and n2 = 3, 4, ...

$$Rc\ \frac{1}{2^2} - \frac{1}{n_2^{\ 2}}$$

For Ist line in spectrum n2 = 3

$$_1\quad Rc\ \frac{1}{2^2} - \frac{1}{n_2^{\ 2}} \qquad Rc\ \frac{1}{4} - \frac{1}{9} \qquad f$$

$$f\quad \frac{5}{36}Rc \qquad Rc\ \frac{36}{5}f$$

For second line n2 = 4

$$_2\quad Rc\ \frac{1}{22} - \frac{1}{42} \qquad Rc\ \frac{1}{4} - \frac{1}{16}$$

$$\frac{4-1}{}$$

$$Rc\ \frac{}{16}$$

$$_2\quad \frac{3}{16}Rc \quad \frac{3}{16}\quad \frac{36}{5}f \quad 1.35f\ .$$

5. (d) Here, Kinetic energy = Rest energy

we know that

Kinetic energy (Relativistic)

$$= (m - m_0)c2,$$

and Rest energy = $mc2$,

where m_0 = rest mass, c = velocity of light

Also, mass (m) of a particle moving with velocity v is given by

$$m\quad \frac{m_0}{\sqrt{1 \quad \frac{v^2}{c2}}}.$$

$(m - m)c2 = mc2$ provides

$$\frac{m0}{\sqrt{1 \quad \frac{v^2}{c2}}}\quad m_0\ c^2 \quad mc2$$

$$\text{or,}\ \frac{1}{\sqrt{1 \quad \frac{v^2}{c^2}}}\quad 1 \quad 1$$

$$\text{or,}\ \frac{1}{\sqrt{1 \quad \frac{v^2}{c^2}}}\quad 2 \quad \text{or,}\quad 1 \quad \frac{v^2}{c^2} \quad \frac{1}{4}$$

$$\text{or,}\ \frac{v^2}{c^2} \quad 1 \quad \frac{1}{4} \quad \frac{3}{4}$$

$$v \quad \frac{\sqrt{3}}{2}c.$$

$$1 - \frac{v^2}{c^2} = \frac{1}{4}$$

$$\frac{v^2}{c^2} = 1 \quad \frac{1}{4} = \frac{3}{4} \qquad v2 = \frac{3}{4}c^2$$

$$v = \sqrt{\frac{3}{2}}\,c$$

6. (d) de-Broglie wavelength () of a particle of mass m and moving with a velocity v is given by, $\frac{h}{mv}$, where h is Planck's constant.

When a particle having charge q is accelerated through a potential V then

$$qV \quad \frac{1}{2}mv^2$$

$$\text{or, } qV \quad \frac{m^2v^2}{2m} \quad mv \quad \sqrt{2mqV}$$

$$\frac{h}{\sqrt{2mqV}}$$

Hence, de-Broglie wavelength of electron,

$$_e \quad \frac{h}{\sqrt{2m_eV}}$$

and de-Broglie wavelength of proton,

$$_p \quad \frac{h}{\sqrt{2m_pV}}$$, where e represents the charge of electron (or proton)

$$\frac{_e}{_p} \quad \sqrt{\frac{m_p}{m_e}}$$

7. (c) Relative speed is given by

$$u \quad \frac{u' \quad v}{1 \quad \frac{u'v}{c^2}}$$

Here, u' = 0.8 c and v = 0.4 c

$$u \quad \frac{0.8c \quad 0.4c}{1 \quad \frac{(0.8c)(0.4c)}{c^2}} \quad \frac{1.2}{1.3}c \quad 0.9\,c.$$

8. (d) By Einstein's equation, photoemission occurs when $h > \quad _0$ or $h > h\,_0$, frequency of incident photon is greater than threshold frequency.

9. (c) Radius of nucleus is given by $R \quad A^{1/3}$

$R \quad R_0A^{1/3}$, where A = mass number

10. (c) Radio carbon dating is done by measuring ratio of 14C present in the sample, since proportion of 14C and 12 C in a body is same; but after death 14C decays. Hence knowing the present ratio of 14C to 12 C, sample can be dated.

11. (b) particles are positively charged He nucleus, it can accept 2e–, rays are negatively charged which are similar to e–, can donate 1e–, are radiations. Hence ionisation power of is maximum. are most energetic and is least energetic

Penetration power of is maximum

12. (d) Given half life T= 3.8 days; t = 19 days

$$\frac{N}{N0} \quad \text{?now } \frac{t}{T} \quad \frac{19}{3.8} \quad 5$$

$$\frac{N}{N0} \quad \frac{1}{2}^{\frac{t}{T}} \quad \frac{1}{2}^{5} \quad \frac{1}{32}$$

$$N \quad \frac{N_0}{32} \quad 0.03N_0$$

13. (a) If the p-n junctions are identical, then their resistances would be same. In circuit 1, the first p-n function is forward biased and second is reverse biased. Hence the voltages across them would be different. In circuit 2 both are reverse biased. So potential drop would be same. In circuit 3 both are forward biased; again voltages would be same. Zener diode is a

14. (a) semiconductor device and for semiconductors, temperature coefficient is negative The truth table corresponds to the logic

15. (a) $\overline{A \quad B}$hence it is NOR gate.

16. (d) $\overline{\overline{A.B}}$ in Boolean algebra, can be written as

$$\overline{\overline{A.B}} \quad \overline{\overline{A}} \quad \overline{\overline{B}} \quad A \quad B$$

17. (a) By Dopplers effect, if source moves towards the observer, frequency received will increase.

As $\frac{1}{}$

wavelength will decrease.

18. (d) A coaxial cable consists of a conducting wire surrounded by a dielectric space, over which there is a sleeve of coppermesh covered with a shield of PVC insulation. The power transmission is regulated by dielectric. At high frequencies energy loss due to mesh is significant (called skin effect). Due to their inherent distortion, plate-

19. (b) modulated class C amplifiers are not used as audio amplifiers. Also, class A amplifiers are not used owing to low efficiency. A grid- modulated amplifier has very high frequency. Therefore, in output stage of a TV transmitter, grid-modulated class C amplifier is used. The rms value of carrier current is I

20. (b) c = 8A.

The rms value of modulated current

= Ic + Im = 8.93 A = It

Percentage modulation = ma × 100

The current relation in AM wave is

$$\frac{I_t^2}{I_c^2} \quad 1 \quad \frac{m_a^2}{2} \qquad m_a \quad \sqrt{\frac{I_{t2}^2}{I^{c2}} - 1 \;\; 2}$$

$$\text{Modulation index, } m_a \quad \sqrt{\frac{8.93^{\,2}}{8^2} - 1 \;\; 2}$$

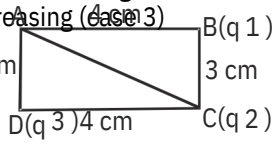
$$\sqrt{\frac{79.7}{64} - 1 \;\; 2} \quad \sqrt{1.24 - 1 \;\; 2}$$

= 0.701

Percentage modulation = ma × 100 =70.1%

21. (d) For a point charge, $E \quad \frac{1}{x2}$. For positive charges, electric field will decrease in positive direction as distance increases, (case 1), for negative charge, as distance increases field will increase (case 4). As we move from positive charge to negative charge, field will keep on decreasing (case 2), As we move from negative to positive charge, field will keep on increasing (case 3)

22. (a)



AC $\sqrt{4^2 \quad 3^2} \quad \sqrt{25} \quad 5$cm.

Potential at A = VB + VC + V$_{D.}$

$$V_A \quad \frac{1}{4 \quad 0} \quad \frac{q^1}{AB} \quad \frac{q2}{AC} \quad \frac{q3}{AD}$$

$$V_A \quad 9 \;\; 109 \;\; \frac{10}{4} \;\; \frac{1^{\;-12}}{0^{\;-2}} \quad \frac{-20\;10^{\;-12}}{5 \;\; 10^{-2}} \quad \frac{10 \;\; 1^{\;-12}}{3 \;\; 0^{\;-2}}$$

$$9 \;\; 109 \;\; \frac{1}{10^{-10}} \;\; \frac{10}{4} \quad \frac{-20}{5} \quad \frac{10}{3} \quad \frac{1}{0}$$

$$0.9 \;\; \frac{150 - 240 \;\; 200}{60}$$

$$\frac{0.9\;110}{60} \quad 1.65\text{V}$$

23. (d) The capacitance (C) of a parallel plate capacitor with dielectric is given by

$$C \quad \frac{_0 \, A}{d - t \;\; 1 - \frac{1}{K}}$$

C = 100 pF = 100 × 10–12 F, $_0$ = 8.85 × 10^{-12},
A = r2 = 3.14×(10–2)2 (r = lcm)
t = 10–3m, d = t, K = 4

$$C1 \quad \frac{8.85\;10^{-12} \quad 3.14\;10^{-4}}{10^{-3} - 10^{\;-3}(1 - \frac{1}{4})}$$

$$\frac{27.79\;10^{\;-16}}{10{-}3 - 0.75\;10^{-3}}$$

$$\frac{27.79\;10^{\;-13}}{0.25}$$

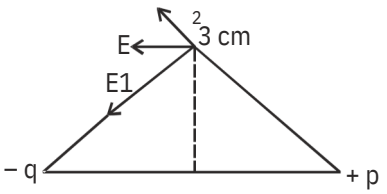
$$\frac{27.79\;10^{\;-13}}{0.25} \quad 111.16\;10^{\;-13}\text{F}$$

(for 1 set)

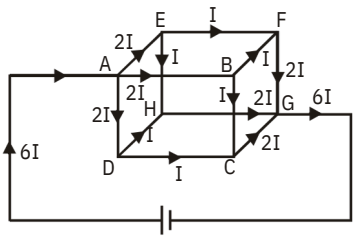
Required C = $100\;10^{\;-12}$F

$$n \quad \frac{C}{C1} \quad \frac{100\;10^{-12}}{111.16\;10^{-13}} \quad 10$$

24. (a) The resultant intensity at a point on equatorial line is E. Eis parallel and opposite to direction of p.



25. (d) Let a total current of 6I enter at A. It divides into three equal parts, each of 2I, along AE, AB and AD. At E, B and D each the current 2I divides into two equal parts, each of I, along EF, EH, BF, BC, DH and DC.
At F, H and C, the two currents each of I, combine together to give a current of 2I at each corner. Thus, at G we get the same current 6I as shown in the figure.



Let r be the value of resistance of each arm of the cube and R be the joint resistance across the conners A and G.
Applying Kirchhoff's law along the loop AEFGA, we get
2Ir + Ir + 2Ir = E
or, 5Ir = E ..... (1)
Also, by Ohm's law,
6I × R = E
6IR = 5Ir, using (1)

or, R $\frac{5}{6}$r.
Here, r = 6

R $\frac{5}{6}$ 6 5

26. (c) Since conductor and semiconductor are connected in parallel hence voltage across them is same. If ammeters show same
reading hence their resistances R $\frac{V}{I}$ are same. If voltage is increased by small value then following the same relation V constant R, both conductor and semiconductor show same current.

27. (d) Time taken by free electrons to cross the conductor

t $\frac{}{vd}$ where drift velocity vd $\frac{I}{neA}$

v_d $\frac{1}{8\ 10^{28}\ 1.6\ 10^{-19}\ 5\ 10^{-7}}$

$\frac{10^{-2}}{64}$ m / s.

t $\frac{1}{vd}$ $\frac{64}{10-2}$ 64 102s 6.4 103sec

28. (b) Temperature coefficient, $\frac{R2 - R_1}{R1\ t2\ \neg 1\ t}$

$t_2 - t1$ $\frac{R2 - R1}{R1}$

t2–t1 $\frac{2-1}{1\ 0.00125}$

t_2 $\frac{1}{0.00125}$ t_1 800 300 1100K

29. (d) Torque on the coil is = nIB A cos. If coil is set with its plane parallel to direction of magnetic field B, then =0°, cos = 1
= nIBA.1 = nIBA = maximum.
Hence, I = maximum (as n,B, A are constant)

30. (d) E at $\frac{1}{2}$bt2is the parabolic equation for thermo emf. The thermoelectric power is

S $\frac{dE}{dt}$

S = a + bt. The graph between S and $\frac{dE}{dt}$ and straight line.

When t = 0, S = a (intercept).

At neutral temperature $\frac{dE}{dt} = 0$ and

$t = t_n \quad 0 = a + bt_n \quad t_n = \frac{-a}{b}$ and at cold

junction $t_i = 2t_n = \frac{-2a}{b}$

31. (d) Energy of proton = K.E. = 1MeV = 10^6 eV.

Frequency $= \frac{Bq}{2m}$

$$= \frac{6.28 \times 10^{-4} \times 1.6 \times 10^{-19}}{2 \times 3.14 \times 1.7 \times 10^{-27}}$$

$= 0.9 \times 10^4$ Hz $\approx 10^4$ Hz

32. (d) Force on a conductor of length due to a magnetic field of strength B is given by

$F = I(\quad B)$

$|F| = I B \sin$

Here, = 90°

$F = IB \quad I \quad R \quad B \quad RIB$

Therefore, a force of magnitude (RIB) will act on each wire. The direction of the forces on each wire will be same. Thus,
total force on the wire AB = RIB + RIB
= 2RIB

33. (d) In first case, the two inductances are in series hence total inductance = $L_0 + L_0$
$= 2L_0$
In second case, the current is same, but no. of turns has doubled since the sense of turning is same.

Thus $L \quad n^2 \quad L \quad 2n^2$

hence inductance $L = 4L_0$
In third case, the sense of turns is opposite. So net inductance cancel each other
$L = 0$

34. (b) At resonance, impedance of circuit is minimum.When impedance of capacitor and inductor is same, $X_L = X_C$ Resonance frequency $= \frac{1}{2\sqrt{LC}}$

$$\frac{1}{2\sqrt{5 \times 80 \times 10^{-6}}} \quad \frac{1}{2\sqrt{400 \times 10^{-6}}}$$

$$\frac{1}{2 \quad 20 \times 10^{-3}}$$

$$\frac{1}{0} \quad \frac{1000}{40} \quad \frac{25}{4}$$

35. (b)M = 5 H, I = 10 A, t = 5×10–4 s.
Now, emf induced in secondary is

$e = -M\frac{dI}{dt}$, (–ve sign shows direction)

$$e = 5 \times \frac{10}{5 \times 10^{-4}} = 1 \times 10^5 V$$

36. (d) Average power of an LCR circuit is P = EvIvcos. Maximum power is dissipated at resonance when $X_L = X_C$. The resonance frequency is given by

$$\frac{1}{2\sqrt{LC}}$$

$$\frac{1}{2 \times 3.14 \times \sqrt{25 \times 10^{-3} \times 400 \times 10^{-6}}}$$

$$\frac{1}{2 \times 3.14 \times 5 \times 20 \times 10^{-5}\sqrt{10}}$$

$$\frac{10^5}{3.14 \times 200\sqrt{10}}$$

$$\frac{10^3}{6.28 \times \sqrt{10}} \quad \frac{159.2}{3.16} \quad 50.31 \text{ Hz}$$

37. (d) Coherent sources should have same frequency and wavelength and constant or zero phase difference. Frequency is same only for wave X 1 and X4 and
phase difference =

38. (c) Since position of central maximum is same, hence asin = is same for both wavelengths. i.e, a = constant, = same,
is same for both waves

and $\frac{1}{2} \quad 1$

39. (b) For bright fringes, $I_{max} = (a + b)^2$

For dark fringes, $I_{min} = (a - b)^2$

Now $\frac{I_{max}}{I_{min}}$ 9 $\frac{a\ b^{\ 2}}{a\ -^{\ 2}\ b}$ $\frac{a\ b}{a\ -\ b}$ 3

$a + b = 3(a - b)$ $a + b = 3a - 3b$ $2a = 4b$

$\frac{a}{b}$ 2 $\frac{a^2}{b^2}$ 4.

Ratio of intensities of the two slits,

$\frac{I_1}{I_2}$ $\frac{a^2}{b^2}$ $\frac{4}{1}$

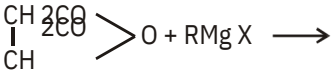
40. (b) Rising and setting sun appears red because light from the sun travels slightly more distance from the horizon, than when it is overhead. Hence blue light is scattered by dust in the atmosphere because scattering

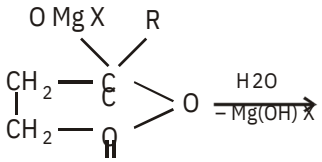
$\frac{1}{4}$ and $^{b <}$ r. Red colour is less scattered and reaches us.

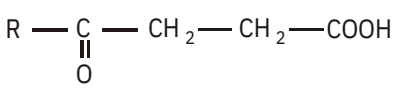
PART - II (CHEMISTRY)

41. (b) Roseumund's reaction –

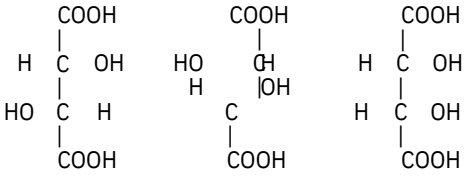
RCOCl H_2 $^{Pd\ /\ BaSO4}$ RCHO HCl

42. (b) 

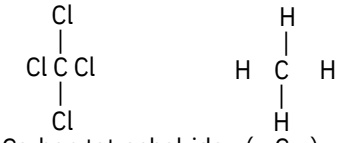




43. (a) (+) – Tartaric acid does not have element of symmetry.



(+) – Tartaric acid (A) Meso tartaric acid (D)
(Plane of symmetry)



Carbon tetracholride (C)
(B) M e t h a n e

44. (b) Proton doners are acids. Electron withdrawing groups (like halogens) increase the acidity of carboxylic acids. Therefore, HCOOH is weakest acid among the given choices.

45. (a) Toluene is used in the reaction to make both substances (acid and

46. (c) alcohol) miscible. Trans esterification is the process of conversion of one ester to another

RCOOR' + R''OH RCOOR'' + R'OH

47. (a) It is expected that the basic nature of amines should be in order tertiary > secondary > primary but the observed order in the case of lower members is found to be as secondary > Primary > tetriary. This anomalous behaviour of tetriary amines is due to steric factors i.e crowding of alkyl groups cover nitrogen atom from all sides thus makes the approach and bonding by a proton relatively difficult which results the maximum steric strain in tetiary amines. The electrons are there but the path is blocked, resulting the reduction in basicity.
Thus the correct order is

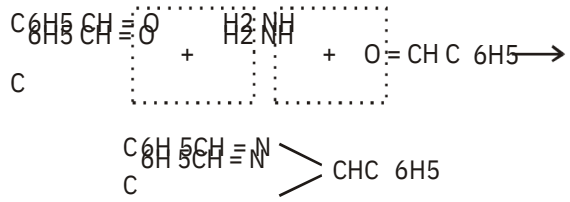
R2 NH > R NH2 > R3N.

48. (d) C_{652}^{HN+C-} l + H2O ^{Boiling}

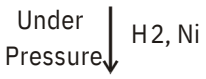
C6H5OH + N2 + HCl
Ph en ol

49. (a) Carbylamine test is given by aliphatic primary amine.
Aliphatic (or aromatic) 1° amine + $CHCl_3$ + KOH → Bad smelling isocyanide.

50. (c) Benzaldehyde reacts with ammonia to form hydrobenzamide.



Hydrobenzamide.



2 $C_6H_5CH_2NH_2$

51. (b) Hybridisation $\frac{1}{2}$ [Number of valence electrons of central atom + no. of monovalent atoms attached to it + negative charge if any – positive charge if any]

$\frac{1}{2}[6 + 4 + 0 - 0] = 5 = sp^3d$

Te52 $Kr4d^{10}5s^25p^4$

52. (a) The permanganate ion has an intense purple colour. Mn (+ VII) has a d0 configuration. So the colour arises from charge transfer and not from d – d spectra. In MnO_4^- an electron is momentarily changing O– – to O– and reducing the oxidation state of the metal from Mn(VII) to Mn (VI). Charge transfer requires that the energy levels on the two different atoms are fairly close.

O = (8) = 2,6
K L

Mn (25) = 2, 8, 15
K L M

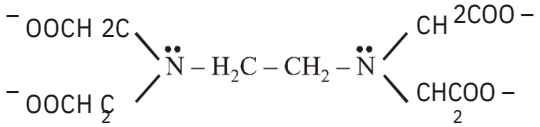
hence the charge transfer occurs from L M.

53. (d) $^{235}_{92}U + ^{1}_{0}n \rightarrow ^{144}_{56}Ba + ^{89}_{36}Kr + 2^{1}_{0}n$

Sum of atomic number of reactants = sum of atomic masses of products.

54. (b) AgCl dissolves in a solution of NH_3 but not in water because Ag^+ forms soluble complex ion with NH_3.

55. (b) EDTA is hexadentate ligand. 4 oxygen and 2 nitrogen atoms act as donars.



56. (c) A co-ordinate bond is a dative covalent bond in which two atoms form a bond and one of them provides both electrons.

57. (d) $[Ni(NH_3)_6]Cl_2$ sp^3d^2 hybridisation
2 unpaired electrons

$Na_3[FeF_6]$ sp^3d^2 hybridisation
3 unpaired electrons

$[Cr(H_2O)_6]SO_4$ d^2sp^3 hybridisation
3 unpaired electrons

$K_4[Fe(CN)_6]$ d^2sp^3 hybridisation
No unpaired electrons
Zero magnetic momoment means all the electrons paired.

58. (c) The IUPAC name is dichloro triphenyl phosphine nickel (II).

59. (c) (d) $(NH_4)_2[TiF_6]$ Cu is in + 1 oxidation state hence has no unpaired electron hence colourless and diamagnetic.

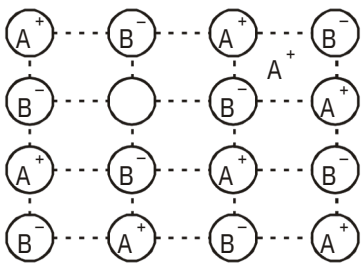
(b) In $(NH_4)_2[TiCl_6]$ Ti is in + 4 oxidation state. hence has no unpaired electron hence colourless and diamagnetic.

(c) In $VOSO_4$ V is in +4 oxidation state hence has one unpaired electron, thus it is coloured and paramagnetic.

(a) In $K_2Cr_2O_7$, Cr is in +6 oxidation. hence has no unpaired electron and thus it is diamagnetic. Though $K_2Cr_2O_7$ has no unpaired electron but it is coloured. This is due to charge transfer spectrum.

60. (b) On an X-ray diffraction photograph the intensity of the spots depends on electron density of atoms/ions.

61. (a) In Frankels defect an ion leaves its regular site and occupy a position in the space between the lattice sites.



62. (b) 8 : 8 type of packing is present in CsCl.
6 : 6 type of packing is present in NaCl and KCl.

63. (d)When solid melts S increases. because when solid changes into liquid randomness increases.

64. (d) Gibbs Helmholtz Equation–

$G \quad H - T S$ .....(1)

differentiate this equation w.r.t. temperature at constant pressure

$$\frac{G}{T}_P \quad \frac{G_y}{T}_P - \frac{G_x}{T}_P \quad \text{....(2)}$$

$= - Sy - (- Sx)$

$= - (S - S) = \quad S$ .....(3)

where S change in entropy

on combining equation (1) & (3) we get

$$G \quad H \quad T \frac{G}{T}_P \quad \text{....(4)}$$

Equation (4) is on alternative form of Gibbs Helmholtz equation

Dividing equ. (4) by T2, we get

$$\frac{G}{T^2} \quad \frac{H}{T^2} \frac{1}{T} \frac{G}{T}_P$$

on rearrangement, we get

$$\frac{G}{T}_P \quad - \frac{H}{T^2}$$

$$H \quad -T2 \frac{G}{T}_P$$

65. (a) Since G A P. V

For a spontaneous process Gshould be negative which is possible only if

AP.VorAW 0.

66. (c)We have to find

$2C_{(s)} \quad H_{2(g)} \quad C_2H_{2(g)} \quad H \quad ?$

This is the equation for formation of acelytene Given

$2C_{(s)} \quad 2O_{2(g)} \quad 2CO_{2(g)}$; H 787 kJ ....(1)

$H_{2(g)} \quad \frac{1}{2}O_{2(g)} \quad H2O_{( )}$; H 286 kJ ...(2)

$C2H_{2(g)} \quad 2\frac{1}{2}O_{2(g)}$

$2CO_{2(g)} \quad H2O_{( )}$;

H 1310kJ ....(3)

Add eq.(1) and (2)

$2C_{(s)} \quad H_{2(g)} \quad 2\frac{1}{2}O_{2(g)} \quad 2CO_{2(g)}$

$H_2O_{( )}$; H 1073 kJ ..... .(4)

Subtract eq. (3) form eq. (4) we get

$2C_{(s)} \quad H_{2(g)} \quad C2H_{2(g)}$;

H 237kJ

67. (b) Endothermic reactions are favoured at high temperature. Therefore, increasing the temperature will shift equilibrium to the right.

68. (a) $\ln \frac{k_2}{k_1} \quad \frac{E_a}{2.303R} \frac{1}{T_1} \quad \frac{1}{T_2}$

69. (a)

	I_2	H_2	2HI
Initial moles	1	1	0
Atequilibrium	(1 x)	(1 x)	2x

$$K_c = \frac{[HI]^2}{[H_2][I_2]} \quad \frac{(2x)^2}{(1 \quad x)^2}$$

Given x = 80% = 0.80

$$K_c = \frac{(2 \quad 0.80)^2}{(1 \quad 0.80)^2} \quad 64$$

70. (d) The equilibrium constant K p will change with temperature for the reaction

$H_2(g) + I_2(g) \rightleftharpoons 2HI(g)$.

Catalyst does not alter the state of equilibrium.
Equilibrium constant depends only upon temperature. The relation of *k* with temp can be shown as

$$\log \frac{k_2}{k_1} \quad \frac{H}{2.303R} \frac{1}{T_1} - \frac{1}{T_2}$$

71. (d) Eq. mass of $Ca^{++} = \frac{Mol.mass}{2} \quad \frac{40}{2} \quad 20$

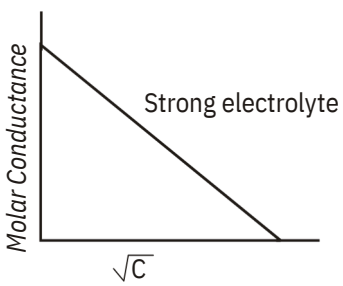
$$Z = \frac{Eq.mass\ of\ water}{96500} \quad \frac{20}{96500}$$

Given w = 60 g, i = 5 amp.
w = zit

or $60 = \frac{20}{96500} \quad 5 \quad t$

$$t = \frac{96500 \quad 60}{20 \quad 5} \quad 57900 \text{ sec} = 16 \text{ h.}$$

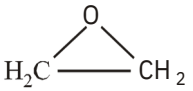
72. (b) According to Debye – Huckel – Onsagar equation $_m \quad ^o_m - A \quad B \quad ^o_m \sqrt{C}$ where A and B are the Debye – Huckel constants. If we plot a graph between molar conductance $_m$ against the square roots of the concentration $\sqrt{C}$ a straight line is obtained



73. (c) Molar conductance of $CaCl_2$
= Molar conductance of Ca^{2++} + 2 × (molar conductance of Cl^-)
= 118.88 × 10–4 + 2 (77.33 × 10–4)
= 273.54 × 10–4 m2 mho mol–1

74. (a) Lower the value of reduction potential, greater will be the reducing power of element. Since Zn has lowest reduction potention hence Zn is the strongest reducing agent.

75. (a) The epoxide ring consists of three membered ring with two carbon atoms and one oxygen.



76. (c) Grignard reagent is a sigma bonded organometallic compound in which Mg is bonded with one alkyl and one halogen group. This can be prepared as

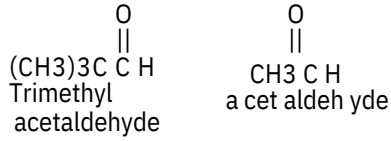
$$R–X + Mg \xrightarrow{ether} R–Mg–X$$

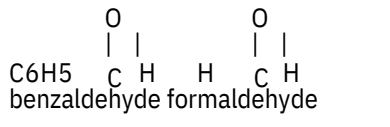
77. (a,c) Presence of electron attracting group like – NO_2, – Cl increases the acidity of phenol as it enables the ring to draw more electrons from the phenoxy oxygen and thus releasing easily the proton. Presence of electron releasing group e.g. – OCH_3 group in benzene sing decreases the acidity of phenol as it strengthens the negative charge on phenoxy oxygen and thus proton release becomes difficult.
Further phenols are much more acidic than alcohols. The acidic nature of phenol is due to the formation of stable phenoxide ion in solution.

$$C_6H_5OH + H_2O \rightleftharpoons C_6H_5O^- + H_3O$$

No resonance is possible in alkoxide ions (RO–) derived from alcohols. The negative charge is localized on oxygen atom. Thus alcohols are not acidic. Aldol condensation is given by aldehydes which have -H atoms. So acetaldehyde gives this reaction.

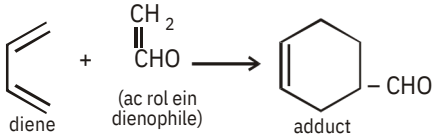
78. (b)





79. (d) H3C CH2 C H2C CH2 C OCH3
6 5 4 3 2 1
Methyl 4 oxohexanoate

80. (b) It is cycloaddition reaction between a conjugated diene and subsituted alkene, unsaturated carboxyl compound.



PART - III (MATHEMATICS)

81. (c) Sinceaandbare coplanar, therefore a×b is a vector perpendicular to the plane containing aandb. Similarlyc×d is a vector perpendicular to the plane containing candd. Thus, the two planes will be parallel if their normals i.e. a×b and c×d are parallel.
(a×b) × (c×d) = 0

82. (d) Given : diagonals d=3i^ 1 j^ 2k^and
d2 i^ 3j^ 4k^

Area of a parallelogram = 1/2 |d1 d2|

Now, d1×d 2 $\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 3 & 1 & 2 \\ 1 & 3 & 4 \end{vmatrix}$

= i^(4 6) j^(12 2) k^(9 1)

d1 × d2 = 2i^ 14j^10k^

|d1 d2|= √(2)2 (14)2 (10)2

= √4 196 100 √300 2√75

Area of parallelogram = 1/2 |d1 d2|

= 1/2 × 2√75 = √75 square units

83. (d) Given : cos x + cos2x = 1
cos x =1– cos2x cos x = sin2 x
cos2x = sin4 x 1– sin2 x = sin 4x
sin4x + sin2x =1
cubic both sides, we have
sin12x + sin6x + 3 sin6x (sin4 x + sin2x) =1
sin12 x + sin6x + 3 sin10x + 3sin8x =1
sin12x + 3sin10x + 3 sin8x + sin6 x–1= 0

84. (c) (cos + i sin)3/5 = (cos 3 + i sin 3)1/5
= [cos (2k + 3) + i sin (2k + 3)]1/5

= cos 2k 3 / 5 isin 2k 3 / 5 ,

where k = 0, 1,2,3,4

Product of all values.

= cos 3/5 isin 3/5 .cos 2 3 / 5

isin 2 3 / 5 . cos 4 3 / 5 isin

4 3 / 5 . cos 6 3 / 5 isin 6 3 / 5 .

cos 8 3 / 5 isin 8 3 / 5

= cos 3/5 2 3 / 5 4 3 / 5 6 3 / 5

8 3 / 5 isin — ——— ———

——— ———

= cos 5/2 2.3/5 5 1. 2/5

isin 5/2 2.3/5 5 1 . . 2/5

= cos 5/2 . 6/5 8/5

isin 5/2 6/5 8/5

= cos (3 + 4) + i sin(3 + 4)
= cos (4 + 3) + i sin (4 + 3)
= cos 3 + i sin 3

85. (d) $\frac{1\ \ i}{i\ \ 2i}^{2}\ \ \frac{1}{1}\ \ \frac{1\ \ i^2\ \ 2i}{2i^2\ \ i}$

$\frac{1\ \ 1\ \ 2i}{2\ \ i}\ \ \frac{2i}{2\ \ i}\ \ \frac{2\ \ i}{2\ \ i}\ \ \frac{4i\ \ 2i^2}{4\ \ i^2}$

[i2 = –1]

$\frac{4i\ \ 2}{4\ \ 1}\ \ \frac{2\ \ 4i}{5}\ \ \frac{2}{5}\ \ \frac{4i}{5}$

The imaginary part = $-\frac{4}{5}$

86. (a) Given $\sin^{-1}x + \sin^{-1}y = \frac{}{2}$ (i)

we know that $\sin^{-1}x + \cos^{-1}x = \frac{}{2}$

$\sin^{-1} x = /2 - \cos^{-1}x$

Equation (1) becomes.

$\frac{}{2} - \cos^{-1}x + \frac{}{2} - \cos^{-1}y = \frac{}{2}$

$\cos^{-1}x + \cos^{-1}y = \frac{}{2}$

87. (c) Equation of ellipse $\frac{x^2}{b^2}\ \ \frac{y^2}{a^2}\ \ 1$, where a > b.

Given, $\frac{x^2}{16}\ \ \frac{y^2}{75}\ \ 1$ b = 4, a = 5

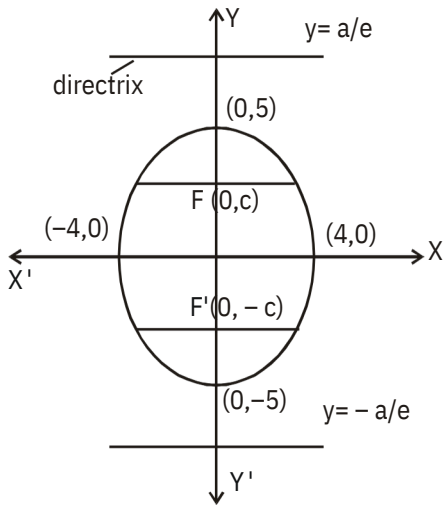
But e = $\sqrt{1\ \ \frac{b^2}{a^2}}\ \ \sqrt{1\ \ \frac{1}{6}}$

$\frac{2}{5}$

e = $\frac{3}{5}$

equation of directrix y = $\frac{a}{e}$

y = $\frac{5}{3/5}$ 3y = ± 25



88. (a) Since the normal at (ap2, 2ap) on y2 = 4ax meets the curve again at (aq2, 2aq), therefore px + y = 2ap + ap3 passes through (aq2 2aq)

paq2 + 2aq = 2ap + ap3

p(q2–p2) = 2(p – q)

p (q + p) = –2

p2 + pq + 2 = 0

89. (d) Given : equation of line, x–3y=1 (1)
and hyperbola x2 – 4y2 = 1 (2)
putting x = 1 + 3y in equation (2), we get
(1 + 3y)2 – 4y2 =1 1+ 9y2 + 6y – 4y2 = 1

5y2 + 6y = 0 y(5y + 6) = 0

y = 0 or y = $\frac{6}{5}$

x = 1 for y = 0 & x =1 $\frac{18}{5} = \frac{13}{5}$

for y = –6/5

the line (1) cuts the hyperbola (2) in at most two point.

co-ordinates of points are P(1,0) &

Q $\frac{13}{5}$, $\frac{6}{5}$

PQ = $\sqrt{1\ \ \frac{13}{5}^{2}\ \ 0\ \ \frac{6}{5}^{2}}$

$$\sqrt{\frac{18}{5}^2 \quad \frac{3}{6}} \qquad \sqrt{\frac{324 \quad 36}{25}} \qquad \sqrt{\frac{360}{25}}$$

$$2$$

$$5$$

length of straight line PQ $= \frac{6\sqrt{10}}{5}$ units.

90. (c) Given $x = t2 + 2t - 1$ & $y = 3t + 5$

$x = t2 + 2t + 1 - 2$ & $y = 3t + 3 + 2$

$x = (t+1)2 - 2$ & $y = 3(t + 1) + 2$ (2)

$(t+1) = \sqrt{x \quad 2}$ ..... (1)

Equation (2) becomes [using equation (1)]

$y = 3\sqrt{x \quad 2} \quad 2$

$y - 2 = 3\sqrt{x \quad 2}$

squaring both sides, we get

$(y-2)2 = 9 (x + 2)$

$Y2 = 9X$ where $Y = y - 2$ & $X = x + 2$

This equation represents a parabola

91. (c) Slope of the normal at (3,4) is the value of

$\frac{1}{f'(x)}$ at $x = 3$ or $\frac{1}{f'(3)} = \tan\frac{3}{4} = -1$

f ` '(3) = 1

92. (c) Given : $f(x) = x2\, e-2x$, $x > 0$

$f'(x) = x2.e-2x(-2) + e-2x.2x$

put $f'(x) = 0$ $2e-2x.\ x(-x + 1) = 0$

$x = 1$ or $x = 0$

$f''(x) = (-4x2 - 6x + 1)e-2x$

$f''(1) = -9e-2x < 0$

$f''(0) = e-2x > 0$

value of f(x) is maximum at x = 1

$f(x) = x2.e-2x \quad f(1) = e-2 = \frac{1}{e2}$

93. (d) Given : $(x + y) \sin U = x2y2$

$\sin U = \frac{x2y2}{x \quad y} = v$ (let)

Here n = 2 – 1 = 1

Euler's theorem $x.\frac{v}{x} \quad y.\frac{v}{y} \quad nv$

$\sin U$

$x\frac{\sin U}{x} \quad y\frac{}{y} \quad \sin U$

$x.\cos U\ \frac{U}{x} \quad y.\cos U.\ \frac{U}{y} \quad \sin U$

$x\frac{U}{x} \quad y\frac{U}{y} \quad \frac{\sin U}{\cos U} \quad \tan U$

$U \qquad U$

$x\frac{}{x} \quad y\frac{}{y} \quad \tan U$

94. (c) Equation of the given curve in parametric for m ,

$x = t2 + 1$ and $y = t2 - t - 6$

Y-coordinate of the point, where the given curve meets X-axis is 0.

When y = 0, then $t2 - t - 6 = 0$

$t^2 \quad 3t \quad 2t \quad 6 \quad 0$

$t(t \quad 3) \quad 2(t \quad 3) \quad 0$

$(t \quad 3)(t \quad 2) \quad 0$

$t \quad 3$ or 2

when t = 3, then x = 10

when t = –2, then x = 5

Hence, the points where the curve meets the X-axis are (10, 0) and (5, 0).

Now, $\frac{dy}{dx} \quad \frac{\frac{dy}{dt}}{\frac{dx}{dt}} \quad \frac{2t \quad 1}{2t}$

Slope of the tangent at point (10, 0)

$m1 \quad \frac{dy}{dx}_{x\ 10} \quad \frac{dy}{dx}_{t\ 3} \quad \frac{5}{6}$

Slope of the tangent at point (5, 0),

$m_2 \quad \frac{dy}{dx}_{x\ 5} \quad \frac{dy}{dx}_{t\ 2} \quad \frac{5}{4} \quad \frac{5}{4}$

If be the angle beween two tangents, then

$$\tan \quad \left|\frac{m2 \quad m1}{1\ m1m2}\right| \quad \left|\frac{\frac{5}{4} \quad \frac{5}{6}}{1 \quad \frac{5}{6}\ \frac{5}{4}}\right|$$

$$\left|\frac{\frac{15 \quad 10}{12}}{\frac{24 \quad 25}{24}}\right| \quad \left|\frac{\frac{5}{12}}{\frac{49}{24}}\right| \quad \left|\frac{10}{49}\right|$$

$$\tan \quad \frac{10}{49}$$

$$\tan^{1} \quad \frac{10}{49} \qquad \tan^{1} \frac{1}{0}$$

$$4$$

$$9$$

95. (b) $\int_1^4 |x-3|\,dx = \int_1^3 (x-3)\,dx + \int_3^4 (x-3)\,dx$

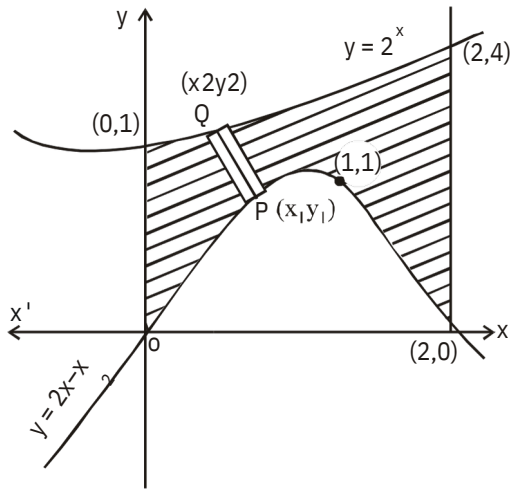
$$\left[\frac{x^2}{2} - 3x\right]_1^3 + \left[\frac{x^2}{2} - 3x\right]_3^4$$

$$\frac{1}{2}[32-12] - 3[3-1] + \frac{1}{2}.[42-32] - 3[4-3]$$

$$\frac{1}{2}(8) - 3(2) + \frac{1}{2}.(7) - 3(1)$$

$$4 - 6 + \frac{7}{2} - 3 + \frac{5}{2}$$

96. (b) Required area $= \int_0^2 (y_2 - y_1)dx$



$$\int_0^2 (2^x - 2x - x^2)dx$$

$$\left[\frac{2^x}{\log 2} - x^2 - \frac{x^3}{3}\right]_0^2$$

$$\frac{4}{\log 2} - 4 - \frac{8}{3} - \frac{1}{\log 2} - \frac{3}{\log 2} - \frac{4}{3}$$

97. (d) Let $I = \int_0 \frac{dx}{(a^2 - x^2)^7}$

Put $x = a\tan$ $dx = a\sec^2 d$

limit at $x = 0$ 0 & x $\frac{}{2}$

$$I = \int_0^{\frac{}{2}} \frac{a\sec^2}{a^{14}(1-\tan^2)^7} d$$

$$\frac{1}{a^{13}} \int_0^{\frac{}{2}} \frac{1}{\sec^{12}} d \quad \frac{1}{a^{13}} . \int_0^{\frac{}{2}} \cos^{12} .d$$

But

$$\int_0^{\frac{}{2}} \sin^{2m-1} .\cos^{2n-1} .d = \frac{1}{2} B(m,n)$$

& $B(m,n) = \frac{\lceil m \lceil n}{\lceil m+n}$ & $\lceil\frac{1}{2} = \lceil$

$$I = \frac{1}{a^{13}} \int_0^{\frac{}{2}} \sin 0 .\cos^{12} d$$

$$m = \frac{1}{2},\ n = \frac{13}{2}$$

$$I = \frac{1}{2a^{13}} \frac{\lceil\frac{1}{2} . \lceil\frac{13}{2}}{\lceil\frac{1}{2}+\frac{13}{2}}$$

$$\frac{1}{2a^{13}} . \frac{\lceil . \frac{11}{2} . \frac{9}{2} . \frac{7}{2} . \frac{5}{2} . \frac{3}{2} . \frac{1}{2} . \lceil}{6.5.4.3.2.1}$$

$$\frac{231}{2048} . \frac{1}{a^{13}}$$

98. (c) Let $I = \int e^x \frac{(1-x)^2}{(1+x^2)^2}dx$

$$I = \int e^x \frac{1+x^2-2x}{(1+x^2)^2}dx$$

$$e^x \left[\frac{1+x^2}{(1+x^2)^2} - \frac{2x}{(1+x^2)^2}\right]dx$$

$$I = \int e^x \frac{1}{(1+x^2)}dx + 2\int e^x.\frac{-x}{(1+x^2)^2}dx$$

$$I = \frac{1}{(1+x^2)}.e^x - \int e^x.(-1)(1+x^2)^{-2}\,2xdx$$

$$-2\int \frac{e^x x}{(1+x^2)^2}dx$$

$$I = \frac{e^x}{1+x^2} + 2\int \frac{e^x x}{(1+x^2)^2}dx$$

$$I = \frac{e^x}{1+x^2} + C$$

99. (c) Given : $x\sin\frac{y}{x}dy = \left(y\sin\frac{y}{x} - x\right)dx$

$$\frac{dy}{dx} = \frac{y\sin\frac{y}{x} - x}{x\sin\frac{y}{x}}$$

Put $\frac{y}{x} = z \Rightarrow \frac{dy}{dx} = z.1 + x.\frac{dz}{dx}$

$$x.\frac{dz}{dx} + z = \frac{zx\sin z - x}{x\sin z} = z - \operatorname{cosec} z$$

$$x\frac{dz}{dx} = -\operatorname{cosec} z \Rightarrow -\sin z\,dz = \frac{dx}{x}$$

$$\cos z = \log x + c$$

$$\cos\frac{y}{x} = \log x + c$$

But $y(1) = \frac{\pi}{2} \Rightarrow x = 1, y = \frac{\pi}{2}$

$$\cos\frac{\pi}{2} = \log(1) + c \Rightarrow c = 0$$

$$\cos\frac{y}{x} = \log x$$

100. (c) The equation of the family of circles of radius r is

$(x-a)^2 + (y-b)^2 = r^2$ ...(1)

Where a & b are arbitrary constants.

Since equation (1) contains two arbitrary constants, we differentiate it two times w.r.t x & the differential equation will be of second order.

Differentiating (1) w.r.t. x, we get

$$2(x-a) + 2(y-b)\frac{dy}{dx} = 0$$

$$(x-a) + (y-b)\frac{dy}{dx} = 0 \quad \text{...(2)}$$

Differentiating (2) w.r.t. x, we get

$$1 + (y-b)\frac{d^2y}{dx^2} + \left(\frac{dy}{dx}\right)^2 = 0 \quad \text{...(3)}$$

$$(y-b) = -\frac{1+\left(\frac{dy}{dx}\right)^2}{\frac{d^2y}{dx^2}} \quad \text{...(4)}$$

On putting the value of (y – b) in equation (2), we get

$$x - a = \frac{\left(1+\left(\frac{dy}{dx}\right)^2\right)\frac{dy}{dx}}{\frac{d^2y}{dx^2}} \quad \text{...(5)}$$

Substituting the values of (x – a) & (x – b) in (1), we get

$$\frac{\left(1+\left(\frac{dy}{dx}\right)^2\right)^2\left(\frac{dy}{dx}\right)^2}{\left(\frac{d^2y}{dx^2}\right)^2} + \frac{\left(1+\left(\frac{dy}{dx}\right)^2\right)^2}{\left(\frac{d^2y}{dx^2}\right)^2} = r^2$$

$$\left(1+\left(\frac{dy}{dx}\right)^2\right)^3 = r^2\left(\frac{d^2y}{dx^2}\right)^2$$

101. (c) Given : $\frac{d^2y}{dx^2} + 2\frac{dy}{dx} + y = 2e^{3x}$

The auxiliary equation is

$D^2 + 2D + 1 = 0$ or $m^2 + 2m + 1 = 0$

$(m+1)(m+1) = 0 \quad m = -1, -1$

i.e., repeated roots

Complementary function = $(c_1 + c_2x)e^{-x}$

Now Particular Integral (P.I.)

$$\frac{1}{D^2 + 2D + 1}.2e^{3x} \qquad [D = 3]$$

$$\text{P.I.} \quad \frac{1}{3^2 + 2.3 + 1}.2.e^{3x} = \frac{2e^{3x}}{16} = \frac{e^{3x}}{8}$$

Solution y = C. F. + P. I.

$$y = c_1 + c_2x\, e^{-x} + \frac{e^{3x}}{8}$$

102. (c) Given $ydx + (x - y^3)\, dy = 0$

$$y\frac{dx}{dy} + x = y^3 \quad 0 \quad \frac{dx}{dy} + \frac{1}{y}x = y^2$$

compare this equation to general equation

$$\text{i.e. } \frac{dx}{dy} + Px = Q \qquad P = \frac{1}{y},\ Q = y^2$$

$$\text{I.f.} \quad e^{\int P dy} = e^{\int \frac{1}{y} dy} = e^{\log y} = y$$

$$x \times \text{I.f.} = \int Q\, \text{I.f.} dy + C_1$$

$$x.y = \int y^2.y\, dy + C_1 \quad xy = \frac{y^4}{4} + C_1$$

$$4xy = y^4 + 4C_1$$

$$y = 4xy + 4C_1 = y^4 + 4C_1$$

where $C = -4C_1$

103. (b) Given, $x\,1x2x3x4x5 = 1050$

$$x1x2x3x4x5 = 2 \times 3 \times 5^2 \times 7$$

Each of 2, 3 or 7 can take 5 places and 52 can be disposed in 15 ways.

Hence, number of positive integral solution = 53 × 15 = 1875

104. (d) Number of onto functions: If A & B are two sets having m & n elements respectively such that $1 \le n \le m$ then number of onto functions from A to B is

$$\sum_{r=1}^{n} (-1)^{n-r}.{}^nC_r r^n$$

Given A = {1, 2, 3, ---- n} & B = {a, b, c}

Number of onto functions

$$\sum_{r=1}^{3} (-1)^{3-r}.{}^3C_r\, r^n$$

$$1^n\, {}^3C_1 + (-1)^2\, {}^3C_2\, 2^n + (-1)^3\, {}^3C_3\, 3^n$$

$${}^3C_1 - {}^3C_2 2^n + {}^3C_3 3^n$$

$$\frac{3!}{2!1!} - \frac{3!}{2!1!}2^n + \frac{3!}{3!0!}3^n$$

$$3 - 3.2^n + 3^n$$

$$3^n - 3.2^n + 1$$

105. (b) Let there are n persons in the room. The total number of hand shakes is same as the number of ways of selecting 2 out of n.

$${}^nC_2 = 66 \quad \frac{n(n-1)}{2!} = 66$$

$$n^2 - n - 132 = 0$$

$$n = 12,\ n = -11 \quad 0 \quad n = 12$$

106. (a) We know that, let (G, 0) be a group & e be the identity then

$(a * a)^{-1} = a^{-1} \circ a^{-1}$

$= (a^{-1})^{-1} = a.$

107. (d) No. of tickets = 9

No. of odd numbered tickets = 5

No. of even numbered tickets = 4

Required probability = P {odd, even, odd} + P (even, odd, even)

$$\frac{{}^5C_1}{{}^9C_1} \cdot \frac{{}^4C_1}{{}^8C_1} \cdot \frac{{}^3C_1}{{}^7C_1} + \frac{{}^4C_1}{{}^9C_1} \cdot \frac{{}^5C_1}{{}^8C_1} \cdot \frac{{}^3C_1}{{}^7C_1}$$

$$\frac{5}{9} \cdot \frac{4}{8} \cdot \frac{3}{7} + \frac{4}{9} \cdot \frac{5}{8} \cdot \frac{3}{7} = \frac{5}{18}$$

108. (a) Given

$P(A) = \frac{1}{12}, P(B) = \frac{5}{12}, P(B/A) = \frac{1}{15}$

We know that $P(B/A) = \frac{P(A \cap B)}{P(A)}$

$\frac{1}{15} = \frac{P(A \cap B)}{1/12}$

$P(A \cap B) = \frac{1}{15} \cdot \frac{1}{12} = \frac{1}{180}$

But,

$P(A \cup B) = P(A) + P(B) - P(A \cap B)$

$P(A \cup B) = \frac{1}{12} + \frac{5}{12} - \frac{1}{180} = \frac{89}{180}$

109. (a) $\int_{1.5}^{2} f(x)\,dx$, where $f(x) = \frac{x}{2}$

$\int_{1.5}^{2} \frac{x}{2} dx = \frac{1}{2} \cdot \int_{1.5}^{2} x\,dx = \frac{1}{2} \left[\frac{x^2}{2}\right]_{1.5}^{2}$

$\frac{1}{4}(4 - 2.25) = \frac{1}{4}(1.75) = \frac{17}{5} = \frac{7}{16}$

110. (b) If $2P(X = 0) + P(X = 2) = 2P(X = 1)$

Let probability distribution of X be given by

$P(X = r) = \frac{e^{-m} m^r}{r!}$ where r = 0, 1, 2, ...

$2 \cdot \frac{e^{-m} m^0}{0!} + \frac{e^{-m} m^2}{2!} = 2 \cdot \frac{e^{-m} m}{1!}$

$2 + \frac{m^2}{2} = 2m$

$m^2 - 4m + 4 = 0$

$(m - 2)^2 = 0 \Rightarrow m = 2$

111. (c) $A = \begin{bmatrix} 1 & \tan\theta \\ -\tan\theta & 1 \end{bmatrix}$ & $AB = I$

$B = IA^{-1}$, $A_{11} = 1$, $A_{12} = -\tan\theta$,

$A_{21} = \tan\theta$, $A_{22} = 1$

$|A| = \begin{vmatrix} 1 & \tan\theta \\ -\tan\theta & 1 \end{vmatrix} = 1 + \tan^2\theta = \sec^2\theta$

$A^{-1} = \frac{1}{\sec^2\theta} \cdot \begin{bmatrix} 1 & -\tan\theta \\ \tan\theta & 1 \end{bmatrix}$

$B = \frac{1}{\sec^2\theta} \cdot \begin{bmatrix} 1 & 0 \\ 0 & 1 \end{bmatrix} \begin{bmatrix} 1 & -\tan\theta \\ \tan\theta & 1 \end{bmatrix}$

$B = \frac{1}{\sec^2\theta} \cdot \begin{bmatrix} 1 & -\tan\theta \\ \tan\theta & 1 \end{bmatrix}$

$B = \frac{1}{\sec^2\theta} \cdot A$

$\sec^2\theta \cdot B = A$

$\cos^2\theta^{-1} \cdot B = A$

112. (b) Given : $\begin{vmatrix} 2x & 1 & 4 & 8 \\ 2 & 2x & 2 & 16x \\ 7 & 6 & 2x & 56 \end{vmatrix} = 0$

$2x(4x^2 - 12) - (4x - 14) - 8(12 - 14x) = 0$

$8x^3 - 24x - 4x^2 + 12 - 96 + 112x = 0$

$8x^3 - 4x^2 - 152x - 140 = 0$

$(x + 5)$ is a factor of above equation

$8x^3 + 40x^2 - 36x^2 - 180x - 28x - 140 = 0$

$8x^2(x + 5) - 36x(x + 5) - 28(x + 5) = 0$

$(x + 5)(8x^2 - 36x - 28) = 0$

$(x + 5)(4 - 2x)(2 - 9x - 7) = 0$

$(4 - x)(5 - 2x)(2 - 7x - 2x - 7) = 0$

$(4 - x)(5 - x)(2x - 7)(x + 1) = 0$

$x = -5, 3.5, 1$

113. (b) For only one solution $|A| \neq 0$

$$\begin{vmatrix} k & 2 & 1 \\ 0 & k & 1 \\ 0 & 0 & k \end{vmatrix} \begin{matrix} 2 \\ 2 \end{matrix} \neq 0$$

$$k\,k\,1\,k\,2 \neq 0$$

$$k \neq 0, k \neq 1, k \neq 2. \quad k \neq 1$$

114. (c) Let

$$A \begin{vmatrix} 1 & 2 & 5 \\ 2 & 4 & a \\ 1 & 2 & a \end{vmatrix} \begin{matrix} 4 \\ 1 \end{matrix} \begin{vmatrix} 1 & 2 & 5 \\ 0 & 0 & a \\ 0 & 0 & a \end{vmatrix} \begin{matrix} 6 \\ 6 \end{matrix}$$

$R_2 \to R_2 - 2R_1, R_3 \to R_3 - R_1$

clearly rank of A is 1 if $a = -6$

115. (c) Given : $ax^4 + bx^2 + c = 0$

Equation will be real if $D \geq 0$

$$b^2 - 4ac \geq 0 \quad b^2 \geq 4ac$$

116. (a) Given : $\log_3 x + \log_3 \sqrt{x} + \log_3 \sqrt[4]{x} + \log_3 \sqrt[8]{x} + \log_3 \sqrt[16]{x} = 4$

$$\log_3 x^{1 + \frac{1}{2} + \frac{1}{4} + \frac{1}{8} + \frac{1}{16}} = 4$$

$$\log_3 x^{\frac{1}{1 - \frac{1}{2}}} = 4 \qquad S = \frac{a}{1 - r}$$

$$\log_3 x^2 = 4 \quad x^2 = 3^4 \quad x = \pm 9$$

117. (a) $\left(x + \frac{1}{x}\right)^3 + \left(x - \frac{1}{x}\right) = 0$

$$\left(x + \frac{1}{x}\right)\left[\left(x + \frac{1}{x}\right)^2 + 1\right] = 0$$

$$x + \frac{1}{x} = 0 \quad x^2 + 1 = 0 \quad x = \pm i$$

Thus, the given equation has no real roots.

118. (a) Given : H is the harmonic mean between P & Q

$$H = \frac{2PQ}{P + Q} \quad \frac{1}{H} = \frac{P + Q}{2PQ}$$

$$\frac{2}{H} = \frac{1}{Q} + \frac{1}{P} \quad \frac{H}{P} + \frac{H}{Q} = 2$$

119. (c) We have $|\vec{a} + \vec{b}| = |\vec{a} - \vec{b}|$

$$|\vec{a} + \vec{b}|^2 = |\vec{a} - \vec{b}|^2$$

$$(\vec{a} + \vec{b}).(\vec{a} + \vec{b}) = (\vec{a} - \vec{b}).(\vec{a} - \vec{b})$$

$$\vec{a}.\vec{a} + \vec{a}.\vec{b} + \vec{b}.\vec{a} + \vec{b}.\vec{b} = \vec{a}.\vec{a} - \vec{a}.\vec{b} - \vec{b}.\vec{a} + \vec{b}.\vec{b}$$

$$\vec{a}.\vec{b} + \vec{a}.\vec{b} = 0$$

$$|\vec{a}|\,|\vec{b}|\cos\theta + |\vec{a}|\,|\vec{b}|\cos\theta = 0, \quad |\vec{a}| \neq 0, |\vec{b}| \neq 0$$

$$\cos\theta = 0 \quad \theta = \frac{\pi}{2}$$

The given vector is parallel to $\vec{a} \times \vec{b}$.

120. (a) Given : A, B & C are three points with co-ordinates (1, 2, –1), (2, 0, 3) & (3, –1, 2) respectively.

Now, direction ratio's of AB = 2 – 1, 0 – 2, 3 + 1 = 1, – 2, 4 & direction ratio's of AC = 3 –1, –1 –2, 2 + 1 = 2, – 3, 3

we know that

$$\cos\theta = \frac{a_1a_2 + b_1b_2 + c_1c_2}{\sqrt{a_1^2 + b_1^2 + c_1^2}.\sqrt{a_2^2 + b_2^2 + c_2^2}}$$

$$\cos\theta = \frac{1 \cdot 2 + (-2)(-3) + 4 \cdot 3}{\sqrt{1 + 4 + 16}\sqrt{4 + 9 + 9}}$$

$$\frac{2 + 6 + 12}{\sqrt{21}\sqrt{22}} = \frac{20}{\sqrt{462}}$$

$$\sqrt{462}\cos\theta = 20$$